

THE JOB OF THE  
PROFESSIONAL  
SWIMMING  
POOL  
OPERATOR



# HOUSEKEEPING

- Introduction of Instructors
- Please mute your microphone
- Questions will be taken at the end of each chapter & during the presentation through the messaging area
- Breaks will happen after each chapter and will last about 10-15 minutes



# INTRODUCTION

- OVER THE PAST 30 YEARS AQUATICS HAS BECOME ONE OF AMERICA'S MOST POPULAR RECREATIONAL ACTIVITIES.
- SWIMMING POOLS CAN BE A GREAT SOURCE OF FUN AND EXCITEMENT AND THEY'RE RAPIDLY INCREASING IN NUMBERS. WITH INCREASED USAGE POOLS, IF NOT PROPERLY OPERATED, CAN BECOME A DANGEROUS RISK TO THE HEALTH AND SAFETY OF THE PUBLIC. THE JOB OF THE SWIMMING POOL OPERATOR, POOL MANAGERS, AND LIFEGUARDS IS TO PROVIDE THE MAXIMUM AMOUNT OF BENEFIT WITH THE LEAST AMOUNT OF RISK.
- ALTHOUGH IT IS IMPOSSIBLE TO SUPPLY EVERY BIT INFORMATION REGARDING POOL OPERATIONS, AQUATIC TRAINING AND CONSULTING SERVICES HOPES THAT ITS MANUAL AND COURSE WILL PROVIDE YOU THE BASIC FUNDAMENTALS NECESSARY TO RUN A SAFE AND HEALTHY FACILITY.



# AQUATIC TRAINING SERVICE'S MANUAL

- THE PROFESSIONAL SWIMMING POOL OPERATOR
- WATER BALANCE
- PREVENTION OF DISEASE TRANSMISSION
- POOL AND SPA WATER DISINFECTION
- POOL AND SPA WATER TREATMENT PROBLEMS
- WATER TESTING
- CHEMICAL AUTOMATION
- WATER CIRCULATION
- FILTRATION SYSTEMS
- SPA MAINTENANCE
- CHEMICAL SAFETY
- MAINTENANCE & OPERATION
- LEGALITY AND LIABILITY



**AQUATIC TRAINING &  
CONSULTING SERVICES**



# 12 THINGS THAT CLOSE YOUR POOL

1. NO POOL OPERATOR ON DUTY
2. NO LIFEGUARD WITH CURRENT CERTIFICATIONS IN THE STAND
3. CHEMICALS ARE NOT IN SPECIFIED RANGES (ESPECIALLY CL AND PH)
4. CHEMICAL FEEDERS AT NOT WORKING FOR 24 HOURS
5. FILTRATION SYSTEM NOT WORKING FOR 1 HOUR
6. WATER IS BELOW THE SKIMMERS
7. MAIN DRAIN IS NOT EASILY VISIBLE
8. BATHROOM AND DRAINS ARE NOT OPERATIONAL
9. FACILITY PRESENTS DANGER TO PATRONS
10. FLOW RATE IS INSUFFICIENT FOR ADEQUATE TURNOVER
11. PHONE IS NOT OPERATIONAL
12. HEALTH INSPECTOR IS DENIED ACCESS TO THE FACILITY



# THE PROFESSIONAL SWIMMING POOL OPERATOR'S QUALIFICATIONS:

- AT LEAST 16 YEARS OF AGE
- ATTEND ALL SESSIONS
- 75% OR BETTER ON EXAM
- CERTIFICATION IS VALID FOR 3 YEARS IN MOST JURISDICTIONS



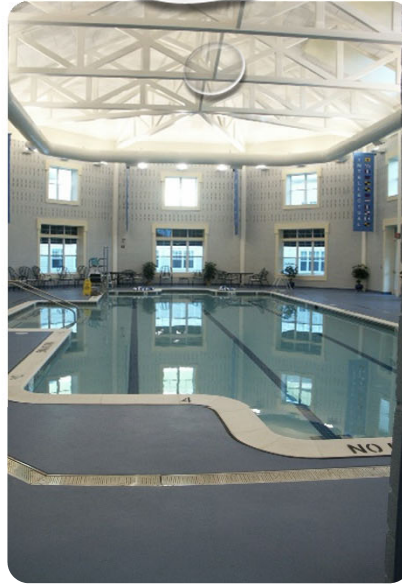
# THE PROFESSIONAL SWIMMING POOL OPERATOR (PSPO)

A PSPO is someone with swimming pool and spa operations knowledge, including but not limited to water disinfection and chemistry, along with circulation, filtration, patron safety & health concerns, and other miscellaneous aquatic topics.

In short, a PSPO is

- RESPONSIBLE FOR THEIR FACILITY'S
  - SAFETY
  - HEALTH
  - MAINTENANCE
- RESPONSIBLE FOR THE MANAGEMENT OF THEIR
  - PATRONS
  - STAFF
  - SELF





**REGARDLESS OF THE SIZE OR SHAPE OF YOUR POOL  
THERE ARE SOME BASICS WE ALL FOLLOW!**





# Basic Pool Operation Need To Knows

- A. CERTIFIED POOL OPERATOR MUST BE PRESENT (OR IN IMMEDIATE CONTROL) FOR A POOL TO BE OPEN
  
- B. GUARD TO SWIMMING PATRON RATIO IS 50 TO 1 FOR MOST STATE REGULATIONS, ATCS RECOMMENDS 25 TO 1
  
- C. FACTORS THAT AFFECT NUMBER OF GUARDS ON DUTY: SKILL OF SWIMMERS, NUMBER OF SWIMMERS IN THE POOL, SHAPE OF FACILITY, TEMPERATURE OR ACTIVITY.
  
- D. POOL READINGS CHLORINE AND PH TAKEN EVERY TWO HOURS, TOTAL ALKALINITY, CALCIUM HARD AND CYANURIC ACID EVERY WEEK.
  
- E. PHONE MUST BE WORKING. IF A CELL PHONE IS DESIGNATED AS THE PHONE IT MUST BE FOR EMERGENCIES ONLY, OTHERWISE THE POOL MUST BE CLOSED
  
- F. RESULTS IF NO LIFEGUARD IS ON THE DECK IS THAT POOL WILL BE CLOSED BY A HEALTH INSPECTOR
  
- G. POOL CERTIFICATION MUST BE POSTED. BOTH POOL OPS AND LGT CERTS SHOULD BE POSTED IN THE GUARD ROOM AT EACH POOL.

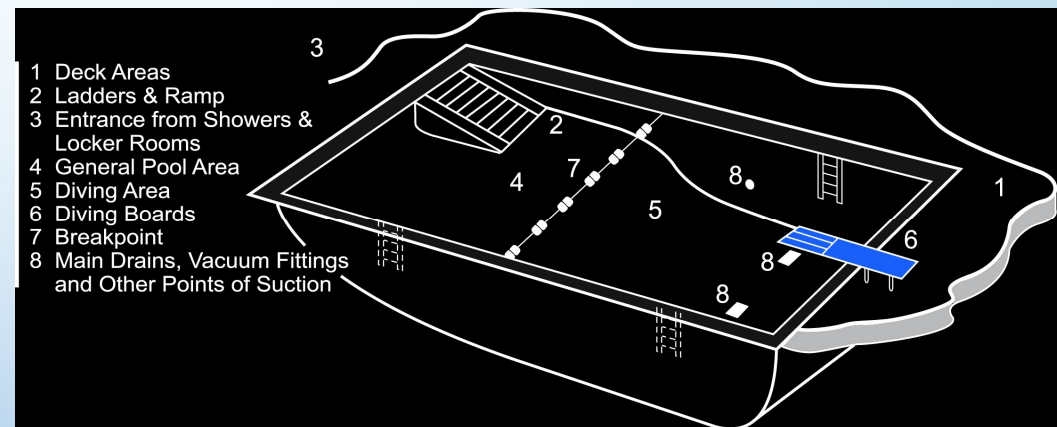
**QUESTIONS ON ALMOST EVERY POOL OPERATOR TEST!**



# LIFEGUARDS HOW MANY?

- MINIMUM NUMBER BY CODE
- SIZE OF POOL
- SHAPE OF POOL
- OBSTRUCTIONS TO VISION
- SKILL OF SWIMMERS
- LOCATION OF SWIMMERS

Consider the danger areas!!



# JOB DESCRIPTION OF A POOL OPERATOR

- CONDUCT A WEEKLY INSPECTION OF THE POOL OR SPA AND RELATED FACILITIES (BATHROOMS, EQUIPMENT ROOM, ETC.).
- ARRIVE AT THE FACILITY PRIOR TO OPENING TO MEASURE, RECORD, AND ADJUST, IF NECESSARY, ALL CHEMICAL LEVELS.
- BE ON-SITE OR PRESENT ACCORDING TO THE LOCAL JURISDICTIONAL CODE.
- MEASURE AND RECORD ALL NECESSARY INFORMATION IN THE TIME FRAME SPECIFIED: MOST PUBLIC FACILITIES REQUIRE THE CHLORINE AND PH TO BE TESTED EVERY TWO HOURS IN A MAIN POOL, EVERY HOUR IN A SPA, AND FOR THESE RECORDS TO REMAIN ON-SITE FOR TWO YEARS AND POSSESSED FOR AT LEAST THREE YEARS
- CLEAN THE FILTRATION SYSTEM WHEN APPROPRIATE
- MAINTAIN DISINFECTANT RESIDUALS ACCORDING TO THE LOCAL CODES.
- MAINTAIN WATER CHEMISTRY ACCORDING TO THE LOCAL CODES.
- ENSURE THAT CHEMICALS ARE USED ACCORDING TO MANUFACTURERS' INSTRUCTIONS AND ARE SAFELY DEPOSITED IN THE WATER.
- ENSURE THAT THE WATER TEMPERATURE FOR A HEATED PUBLIC POOL OR SPA DOES NOT EXCEED THE LOCAL CODES.
- SHOCK THE POOL/SPA ACCORDING TO THE LOCAL CODES.
- CLOSE A POOL OR SPA WHEN APPROPRIATE
- SIGN AND DATE ALL DOCUMENTS.



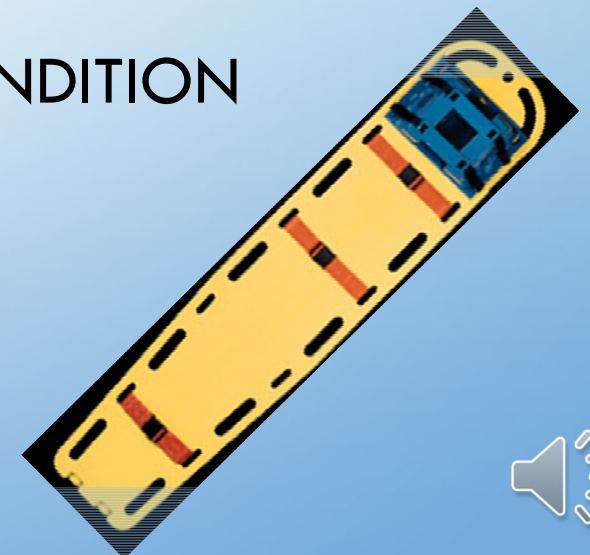


## SAFETY EQUIPMENT



A POOL OPERATOR MUST ENSURE

- THE FACILITY HAS ALL REQUIRED SAFETY EQUIPMENT
- THE EQUIPMENT IS IN OPERATIONAL CONDITION
- THE STAFF TRAINED IN PROPER USE



# SOME WORDS OF WISDOM

## RUNNING A POOL IS A LOT ABOUT PUBLIC RELATIONS

- IT'S NOT ONLY WHAT YOU SAY, BUT HOW YOU SAY IT
- EDUCATE THE PATRONS AND STAFF
- ACT PROFESSIONALLY AND PEOPLE WILL TREAT YOU LIKE A PROFESSIONAL
- ENFORCE ALL THE RULES, ALL THE TIME
- ITS OKAY NOT TO KNOW EVERYTHING, JUST DON'T LIE



THE ACTIONS AT MY FACILITY SHOULD:

BE SAFE

Look Safe

Be Defendable

Be Liked By Patrons



# LIFEGUARDS

- MAY OR MAY NOT BE REQUIRED BY CODE
- IF REQUIRED, LIFEGUARDS MUST 15 YEARS OR OLDER
- RED CROSS LIFEGUARD CERTIFICATIONS ARE VALID FOR TWO YEARS
- RED CROSS CPR CERTIFICATIONS ARE VALID FOR TWO YEARS
- IF LIFEGUARDS ARE NOT REQUIRED: CHECK LOCAL CODE FOR EXACT WORDING



# ROLES OF A LIFEGUARD

## 5 ROLES OF A LIFEGUARD

- ① RECOGNITION
- ② PREVENTION
- ③ RESCUE
- ④ INITIATION
- ⑤ SUPPORT





## SOME STATISTICS YOU SHOULD KNOW:

- 25% OF NON-BREATHERS ARE UNDER 3 YEARS OF AGE AND IN LESS THAN 3 FEET OF WATER.
- MOST RESCUES INVOLVE A VICTIM YOUNGER THAN 7, UNDER 4 FEET TALL AND IN 4 FEET OF WATER OR LESS.
- A BOY IS FOUR TIMES MORE LIKELY TO DROWN THAN A GIRL.
- DROWNING IS THE SECOND LEADING CAUSE OF DEATH IN YOUNG PEOPLE.
- 1 OUT OF EVERY 124,000 PATRONS WILL DIE DUE TO DROWNING AT SWIMMING POOLS IN THE U.S. THIS SUMMER.
- MOST DROWNINGS WHEN A LIFEGUARD IS PRESENT HAPPEN WITHIN SIX FEET OF THE LIFEGUARD



# BE PREPARED!

Prevention is the key to a safe facility but when accident happen your staff should be ready.

- Do regular in-service trainings
- Review the Emergency Action Plan
- Always have a guard in the chair
- Post signs clearly for the public
- Do drills during regular hours
- Do not allow drugs and alcohol

EMERGENCY ACTION PLAN	
SINGLE & MULTI-GUARD FACILITIES	
PRIMARY RESCUER	SUPPORT STAFF OR BYSTANDERS
Victim Recognition	Clear Pool & Maintain Surveillance
Enter the Water & Approach Victim	Assist With Rescue if Needed
Perform Appropriate Rescue	Call EMS & Meet Ambulance at Entrance
Bring Victim to Safety	Provide Necessary Emergency Equipment or Assistance with Care
Assess Condition & Designate Someone to Contact EMS	Help with Crowd Control
Provide Appropriate Care Until EMS Arrives	Follow-Up & Report Incident
Follow-Up & Report Incident	



# RECORD KEEPING

- TYPICAL RECORDS

- DAILY/WEEKLY OPERATING RECORD
- POOL AND SPA ENTRAPMENT HAZARD CHECKLIST
- ACCIDENT/ILLNESS REPORTS
- BATHER REPRIMANDS
- INSPECTION REPORTS
- PREVENTIVE MAINTENANCE RECORDS
- STAFF RECORDS

**Aquatic Training & Consulting Services**  
448 W South Street  
Frederick, MD 21701  
301-760-7114  
[info@aquatictrainingservice.com](mailto:info@aquatictrainingservice.com)

Address \_\_\_\_\_

---

**Incident/Accident**

\_\_\_ Accident \_\_\_ Incident      Staff on duty \_\_\_\_\_

Type of violation (if any): 1. \_\_\_\_\_ 2. \_\_\_\_\_  
3. \_\_\_\_\_ 4. \_\_\_\_\_

\_\_\_ Pool and health club regulation      Authorized patron    YES NO  
\_\_\_ Health department regulation  
\_\_\_ General safety  
\_\_\_ Other \_\_\_\_\_

Was 911 called    YES NO      Was this persons pool privilege suspended    YES NO  
Length of suspension \_\_\_\_\_

---

**Person(s) Involved / Injured**

Name \_\_\_\_\_ Age \_\_\_\_\_ Name of parent if persons if a minor \_\_\_\_\_  
Address \_\_\_\_\_ Telephone \_\_\_\_\_  
Name \_\_\_\_\_ Age \_\_\_\_\_ Name of parent if persons if a minor \_\_\_\_\_  
Address \_\_\_\_\_ Telephone \_\_\_\_\_

---

**Incident/Accident Description**

Date and time \_\_\_\_\_  
Location (Health Club, Indoor Pool, Jacuzzi, Etc.) \_\_\_\_\_  
Description of Incident/Accident \_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_  
Was this person taken to the hospital? \_\_\_\_\_ Which one \_\_\_\_\_

---

**Person(s) notified/witnesses**

Witness \_\_\_\_\_ Address \_\_\_\_\_ Telephone \_\_\_\_\_  
Witness \_\_\_\_\_ Address \_\_\_\_\_ Telephone \_\_\_\_\_  
Witness \_\_\_\_\_ Address \_\_\_\_\_ Telephone \_\_\_\_\_

---

**Office Use**

Insurance company notified \_\_\_\_\_ Date \_\_\_\_\_ By \_\_\_\_\_  
Follow up / disposition \_\_\_\_\_  
\_\_\_\_\_

SWIMMING POOL OPERATING RECORD      Sheet #: \_\_\_\_\_ POOL: \_\_\_\_\_ Month: \_\_\_\_\_ Year: \_\_\_\_\_

DATE	TIME of Test	Pool Operator	TEST EVERY TWO (2) HOURS				RECORD THREE (3) TIMES PER DAY				RECORD AS REQUIRED (Minimum once per day)								OTHER CHEMICAL	NOISE				
			FAC ppm	TAC ppm	CAC ppm	pH	Temp Water	CLARITY	PRESSURE		VACUUM GAUGE		WATER		SHOWER	TRUCK	WHEELS	AMOUNT			ISIC			
			CI=Clear	Effluent	Effluent	Effluent	INFLAT	INFLAT	INFLAT	INFLAT	WATER	WATER	WATER	WATER	WATER	WATER	WATER	WATER	WATER	WATER	WATER	WATER	WATER	WATER
			CL=Cloudy	Filter 1	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2	Filter 2

Time	PH	TA	CH	TEMP	SATUR. INDEX
32	0.0	5	0.3	5	0.7
37	0.1	25	1	25	1.4
46	0.2	50	1.3	50	1.7
53	0.3	75	1.5	75	1.9
60	0.4	100	1.6	100	2.0
66	0.5	150	1.8	150	2.2
76	0.6	200	1.9	200	2.3
94	0.7	300	2.1	300	2.5
94	0.8	400	2.2	400	2.6
106	0.9	800	2.5	800	2.9
128	1.0	1000	2.6	1000	3.0

**SATURATION INDEX**  
RECORD AT LEAST ONCE PER WEEK  
SI = pH + TA + CH + TEMP  
SI = pH + AF + CF + TF - 12.1

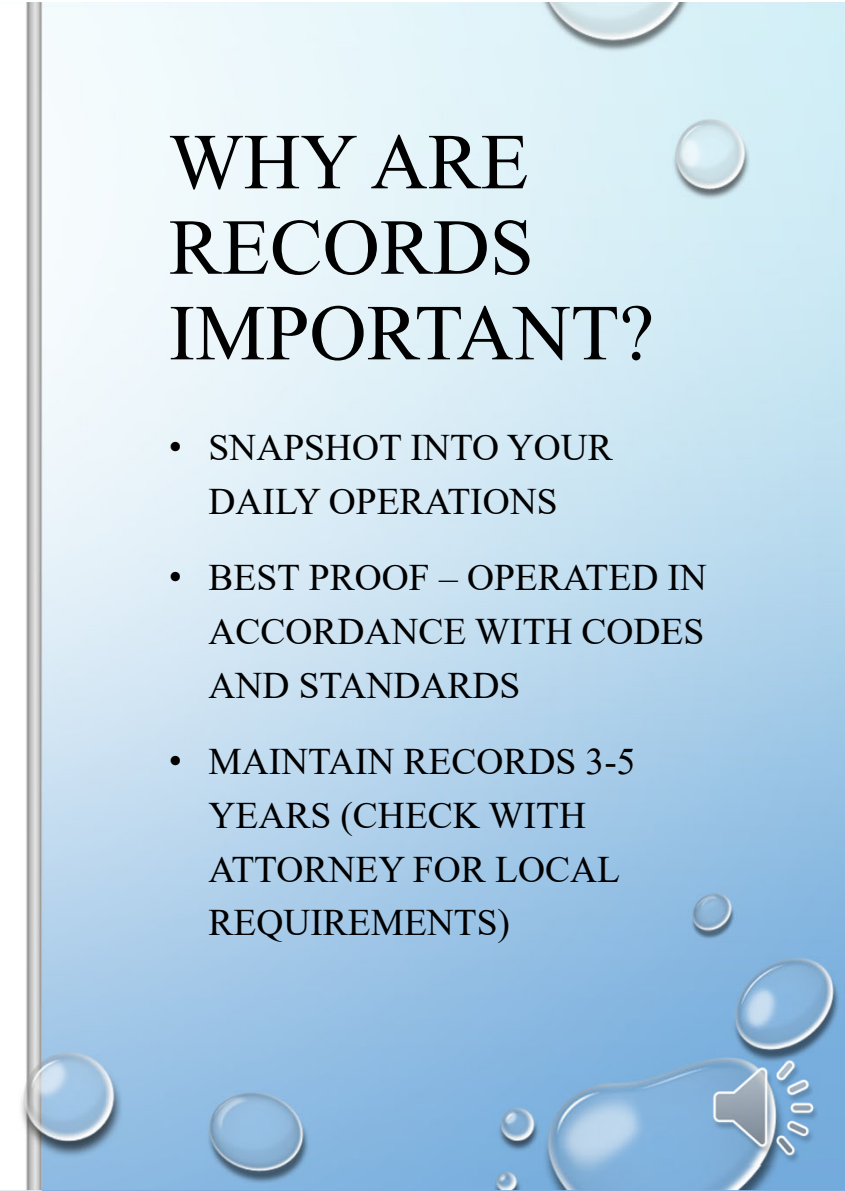
DATE	PH	TIME	PH	AF	CF	TF	-	12.1	=	SATURATION INDEX
							-	12.1	=	
							-	12.1	=	
							-	12.1	=	
							-	12.1	=	
							-	12.1	=	
							-	12.1	=	
							-	12.1	=	
							-	12.1	=	

POOL OPERATORS	
Initial	Name



# WHY ARE RECORDS IMPORTANT?

- SNAPSHOT INTO YOUR DAILY OPERATIONS
- BEST PROOF – OPERATED IN ACCORDANCE WITH CODES AND STANDARDS
- MAINTAIN RECORDS 3-5 YEARS (CHECK WITH ATTORNEY FOR LOCAL REQUIREMENTS)



## ROLE OF THE HEALTH DEPARTMENT

- FORMULATE THE GUIDELINES
- REGULATE THE FACILITIES
- ENFORCE THE REGULATIONS
- INSPECT DUE TO RIGHT OF ACCESS
- ADMINISTER COMPLIANCE  
VIOLATIONS



# RULES EVERY FACILITY SHOULD HAVE:



Require everyone to shower before entering the pool



Swimming is only permitted when a lifeguard is on duty



**NO ONE** is permitted in the pool with open cuts, sores, rashes, or bandages (consider having all band-aids removed before entering the pool)



No glass containers



No running, pushing, or horseplay



No diving in less than 5 feet



Only 1 bounce and 1 person on the diving board at once





**CHAPTER  
TWO**

**WATER  
BALANCE**

HOW TO CHEMICALLY  
BALANCE YOUR SWIMMING  
POOL



# WATER BALANCE “POOL HAPPINESS”

FIVE FACTORS CRITICAL TO ACHIEVING  
WATER BALANCE:

- PH
- TOTAL ALKALINITY (TA)
- CALCIUM HARDNESS (CH)
- TOTAL DISSOLVED SOLIDS (TDS)
- TEMPERATURE (T)

Too Low of Readings Causes Corrosion

Too High of Readings Causes Scaling





# PH

## PH IS THE MOST IMPORTANT PART OF WATER BALANCE

pH is how acidic or basic a solution is.

pH is measured on a 0-14 scale. That means a neutral pH is 7.0 (distilled water).

The pH of the human eye is 7.5. Therefore the pH should be maintained between 7.2-7.8.

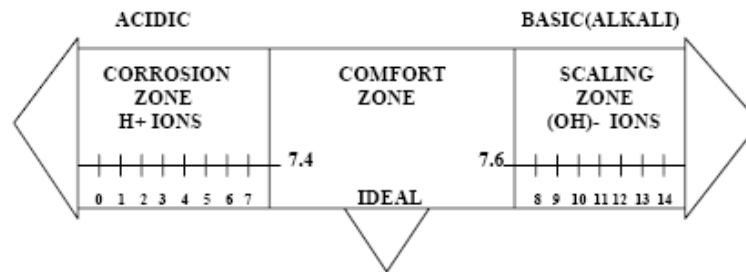
pH should be tested every two hours in larger pools and every hour in a spa with phenol red.

High pH causes scaling & the loss of Chlorine's effectiveness

Low pH is corrosive & will damage metals in your pool and mechanical system.



# WHAT IS pH?



## PROBLEMS

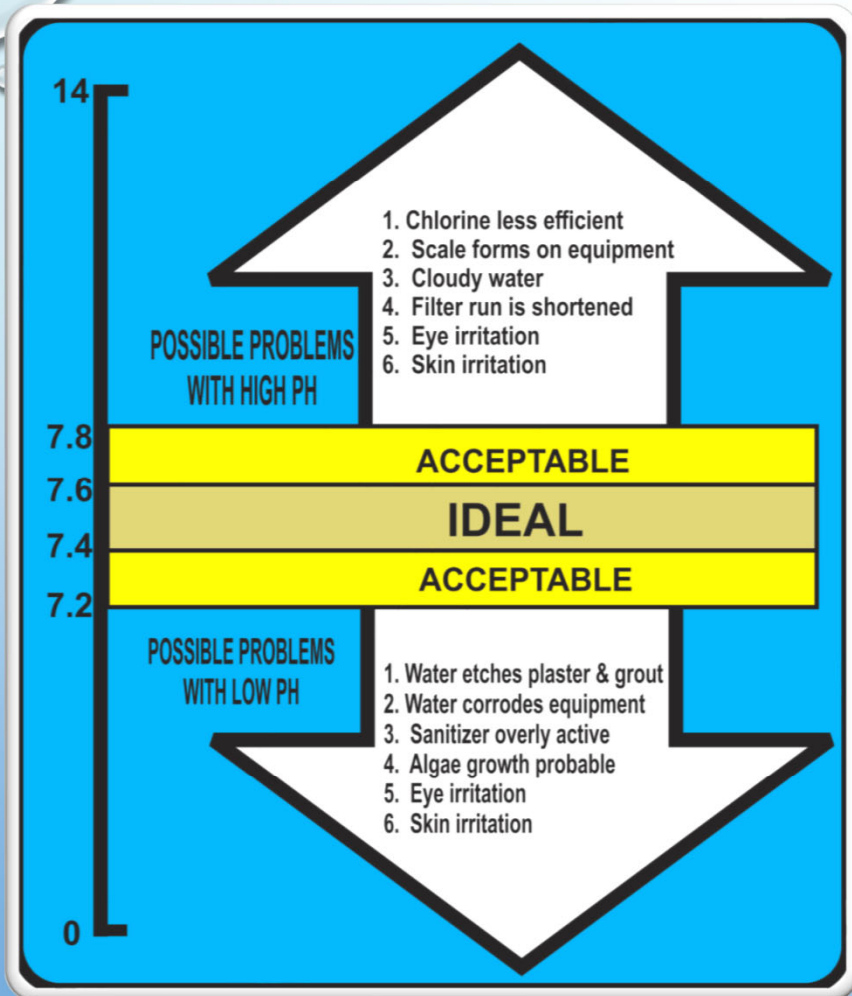
### LOW pH

- FASTER CHLORINE LOSS
- FORMS CHLORAMINES
- EYE IRRITATION
- ETCHING OF PLASTER
- WRINKLING OF LINER
- CORRODES EQUIPMENT AND METAL ACCESSORIES

### HIGH pH

- FORMS SCALES
- CLOUDS WATER
- SHORT FILTER RUNS
- POOR CHLORINE
- HYPOCHLOROUS ACID
- FEWER H+ IONS PRESENT





## FACTORS TRYING TO AFFECT THE PH

SWIMMERS (PERSPIRATION)

SANITIZER

MAKE-UP WATER

### Sanitizers effect on pH CHANGE

Chemicals	Gas Chlorine	Gas Cl <sub>2</sub>	Lowers pH
	Sodium Hypochlorite	NaOCl	Raises pH
	Calcium Hypochlorite	Ca(Ocl) <sup>2</sup>	Raises pH
	Lithium Hypochlorite	LiOCl	Raises pH
	DiChlorocyanuric	DiChlor	Nearly Neutral
	Trichlorcyanuric	TriChlor	Lowers pH
	Bromine		Lowers pH

# PHENOL RED

PHENOL RED, LIQUID OR TABLET, IS THE MOST COMMON PH-TESTING REAGENT. IT IS AN ACID/BASE INDICATOR THAT CHANGES COLOR BETWEEN PH 6.4 AND 8.0.

THE MORE YELLOW THE COLOR, THE LOWER THE PH.

THE MORE RED THE COLOR, THE HIGHER THE PH.

THERE ARE PH METERS, WHICH ARE VERY ACCURATE AND ALSO VERY EXPENSIVE. IF THE DISINFECTANT LEVEL IS HIGH (ABOVE 10 PPM), ADD 1 DROP OF LIQUID SODIUM THIOSULFATE (DISINFECTANT NEUTRALIZER) TO NEUTRALIZE THE SAMPLE, BEFORE ADDING THE PHENOL RED.

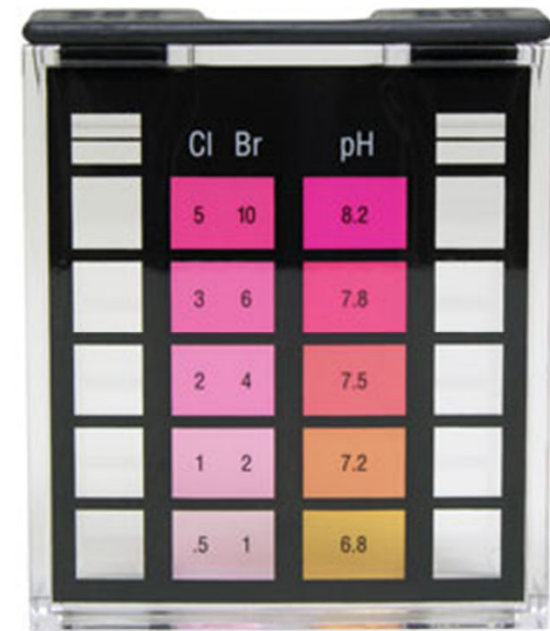
THE PH USUALLY NEEDS TO BE TESTED EVERY 1-2 HOURS, CHECK WITH THE LOCAL HEALTH DEPARTMENT.

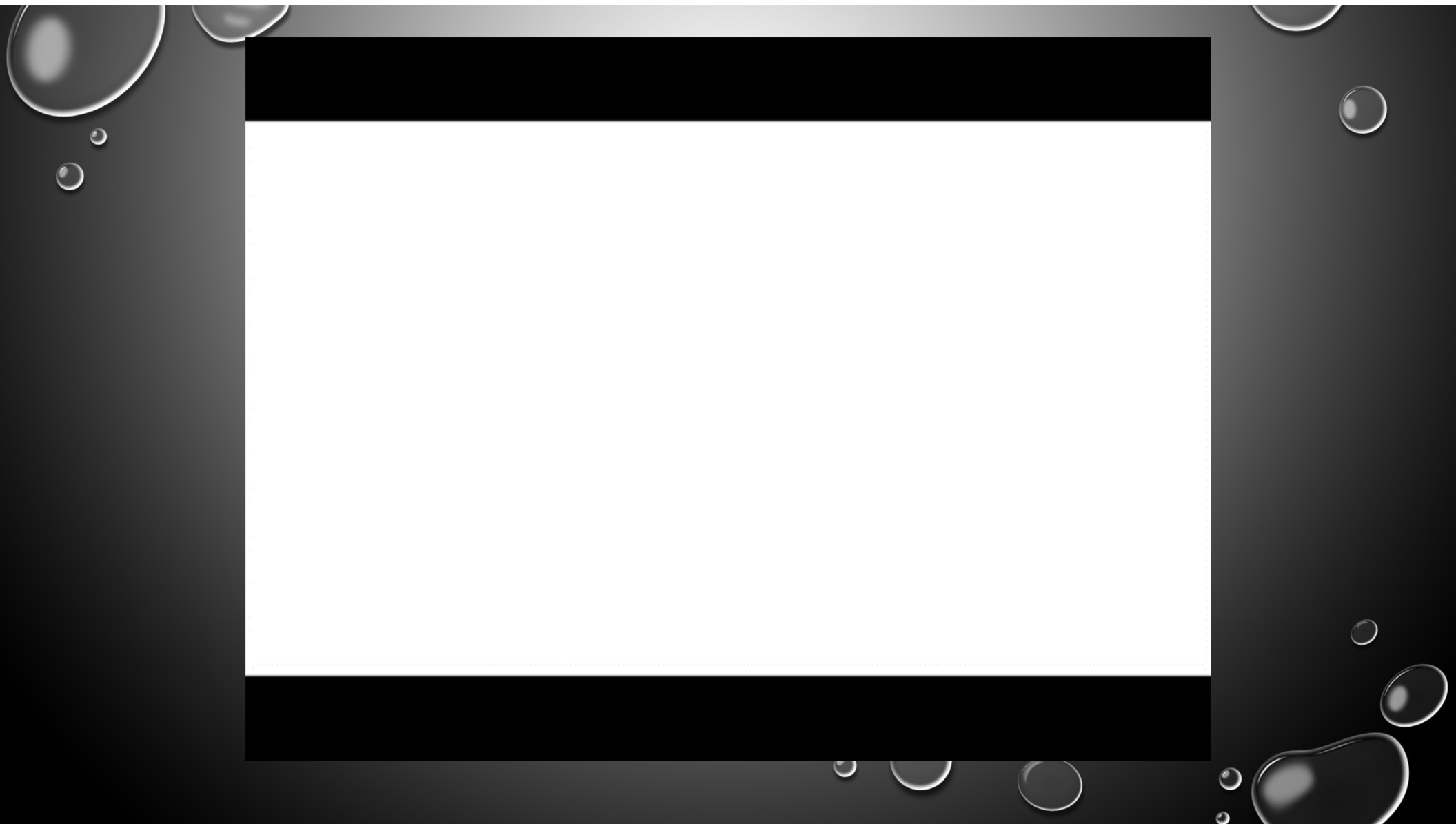


# TESTING PH

## PH TEST

- YOU NEED TO WASH AND FILL THE LARGE COMPARATOR TUBE WITH 44 ML WATER SAMPLE.
- INSTILL 5 DROPS OF PH IND SOL. CLOSE THE CAP AND SWIRL TO MIX.
- MATCH THE RESULTED COLOR WITH THE COLORS IN THE CHART AND RECORD IT AS PH UNITS. IF THE RESULT SUGGESTS A PH
- ADJUSTMENT, SAVE THE SAMPLE FOR FUTURE REFERENCE.
- FOR A SAMPLE COLOR THAT FALLS BETWEEN TWO VALUES, THE AVERAGE OF THESE TWO VALUES IS YOUR PH UNIT.
- TO DECREASE PH: YOU NEED TO SEE ACID DEMAND TEST.
- TO INCREASE PH: YOU NEED TO SEE BASE DEMAND TEST.





If the pH of water is allowed to be either in the acidic or basic range for too long a period of time, problems can start to develop. These are long-range problems that can have serious effects on the pool surface and equipment as well as water quality. To adjust pH the pool operator should use one of the following methods:

# PH ADJUSTMENT

TO INCREASE PH

- SODA ASH (SODIUM CARBONATE)



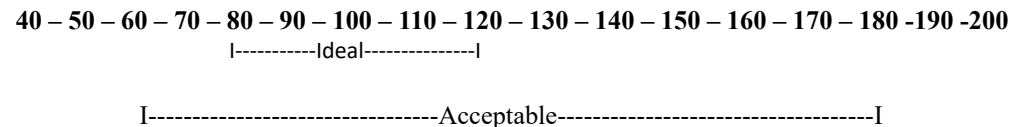
TO DECREASE PH

- MURIATIC ACID
- SODIUM BISULFATE
- CARBON DIOXIDE



# TOTAL ALKALINITY

- MEASUREMENT OF CARBONATES, BICARBONATES, HYDROXIDES, AND CYANURATES IN THE WATER
- MEASURE OF WATER'S ABILITY TO RESIST A CHANGE TO ITS PH
- BUFFERING CAPACITY (THE PH BUFFER) IF THE TA IS TOO LOW THE PH WILL BOUNCE



Recommended Range:

60-180 ppm

Ideal Range:

80-120 ppm

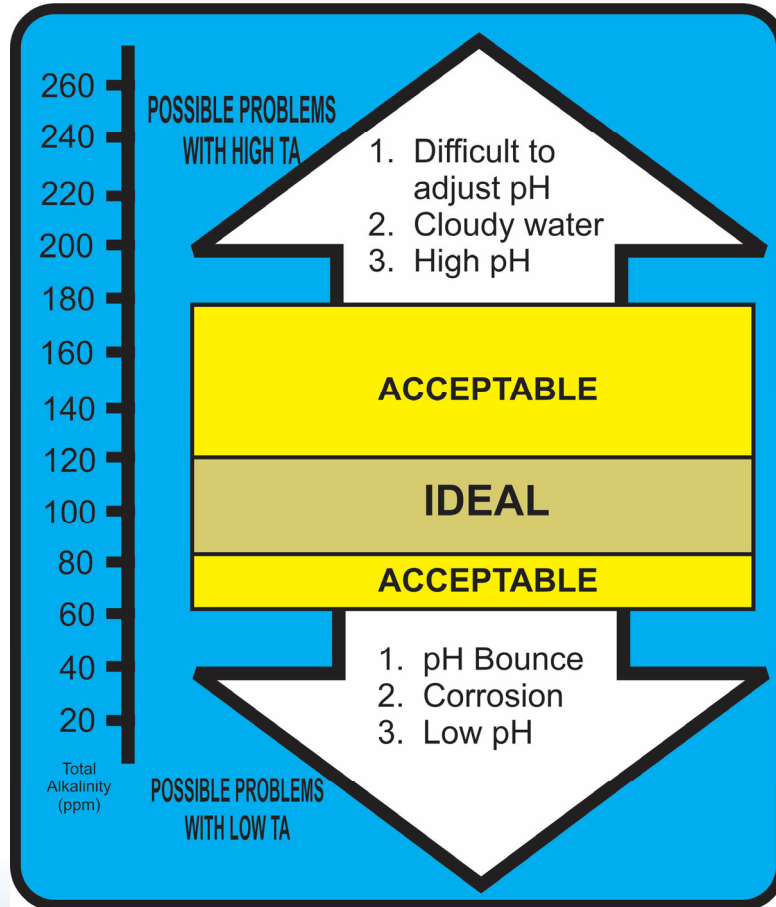
80 – 100 with basic sanitizers

100 – 120 with acidic sanitizers





# TOTAL ALKALINITY



Too High

pH is locked in place

Water is Cloudy

Scaling

Too Low

pH "bounces" up and down

Corrosion and Etching



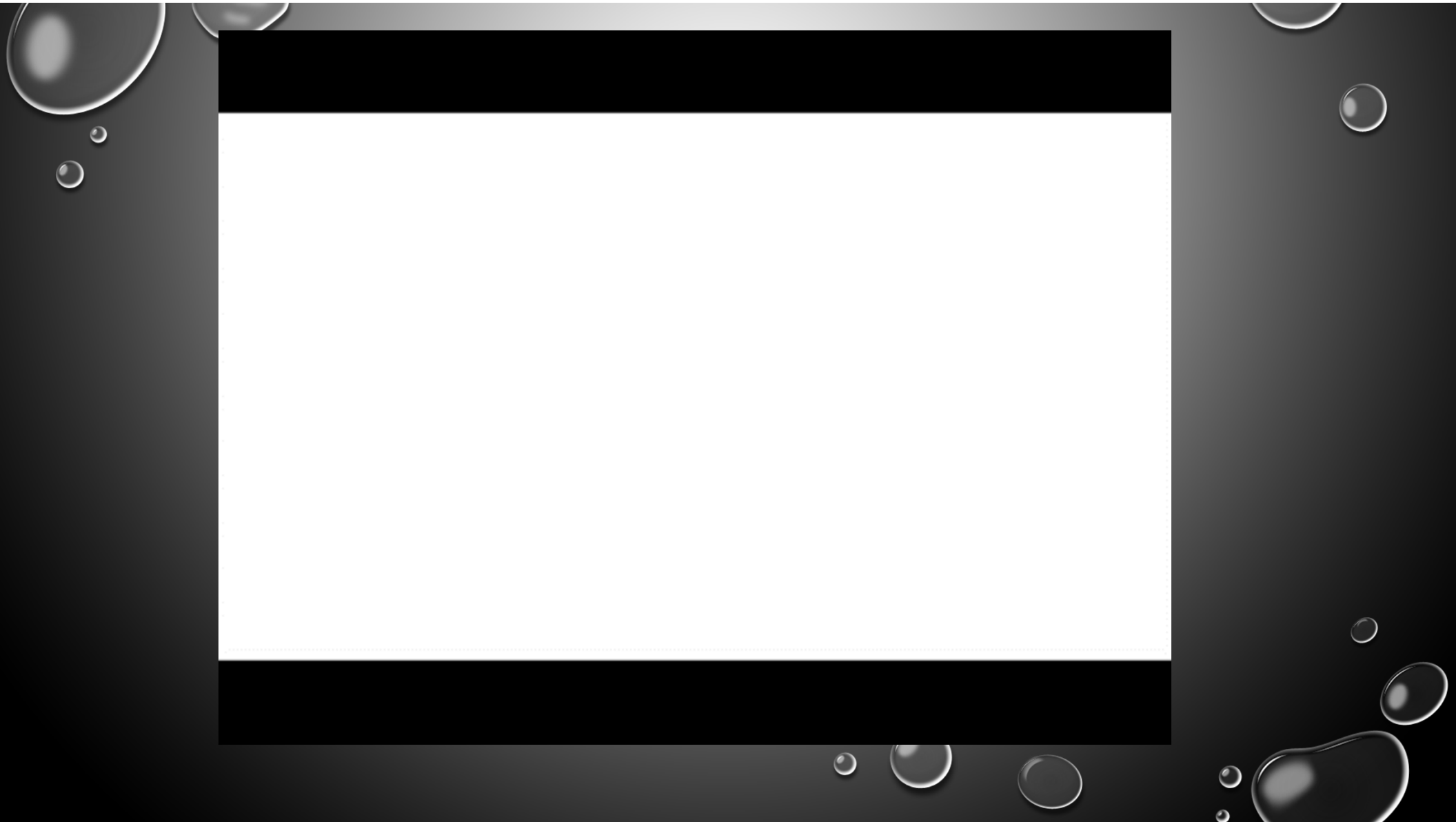
# TOTAL ALKALINITY

## TOTAL ALKALINITY TEST

- You need to wash and fill the large comparator tube with 25 ml water sample.
- Instill 2 drops of R-007 THIOSULFATE N/10. Close the cap and mix.
- Instill 5 drops of R-008 TOTAL ALKALINITY INDICATOR. Close the cap and mix. You should notice the sample turning green.
- Instill R-009 SULFURIC ACID. You need to count every drop, and then mix after each drop until the water sample's color changes from green to red.
- Multiply the drops of SULFURIC ACID by ten (10). Record the result as parts per million



TEST ONCE PER WEEK IN A MAIN POOL AND ONCE PER DAY IN A SPA



Not controlling the Total Alkalinity is one of the biggest mistakes made by pool operators. Low TA will cause the pH to bounce making maintaining your facility nearly impossible. Facilities that use Carbon Dioxide at their facilities to maintain pH will experience high TA over time. To adjust TA the pool operator should use one of the following methods:

# TOTAL ALKALINITY ADJUSTMENT

TO INCREASE TA

- BAKING SODA  
(SODIUM  
BICARBONATE)



TO DECREASE TA

- MURIATIC ACID
- SODIUM BISULFATE



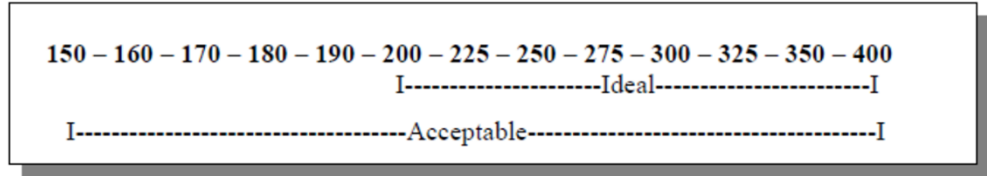
# CALCIUM HARDNESS

## POOL/SPA WATER NEEDS CALCIUM HARDNESS

- TOO HIGH CAUSES SCALING & CAN RESTRICT WATER FLOW IN YOUR PIPES AND DAMAGE YOUR HEATER.
- TOOL LOW CAUSES CORROSION AND WILL RUIN THE SIDES AND BOTTOM OF YOUR POOL STRUCTURE.

AMOUNT OF CALCIUM THAT CAN BE DISSOLVED IS CONTROLLED BY:

- WATER TEMPERATURE
- PH
- CARBONATE ALKALINITY



Recommended Range:

150-400 ppm

Ideal Range:

200-400 ppm



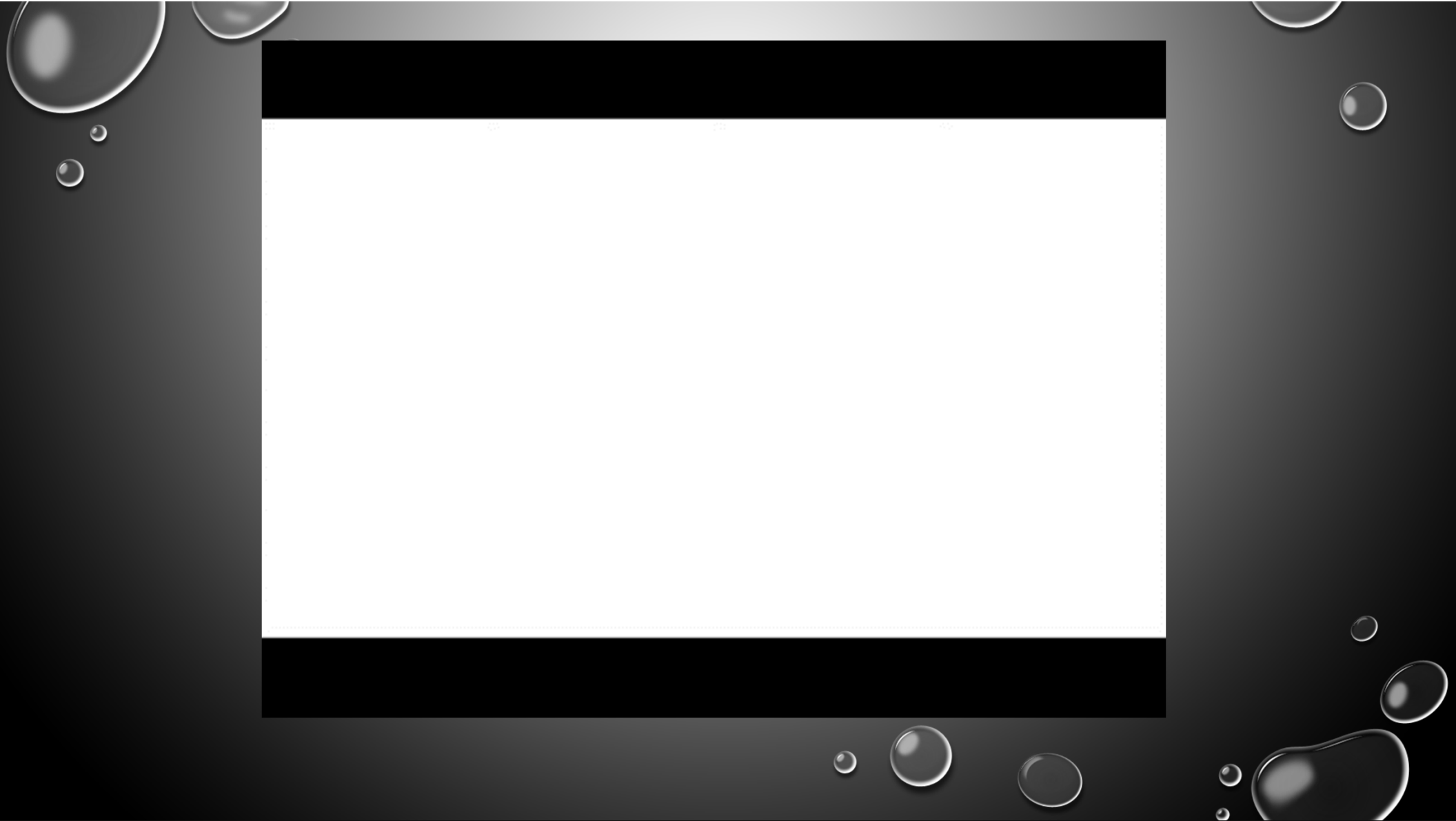
# CALCIUM HARDNESS

## Calcium Hardness Test

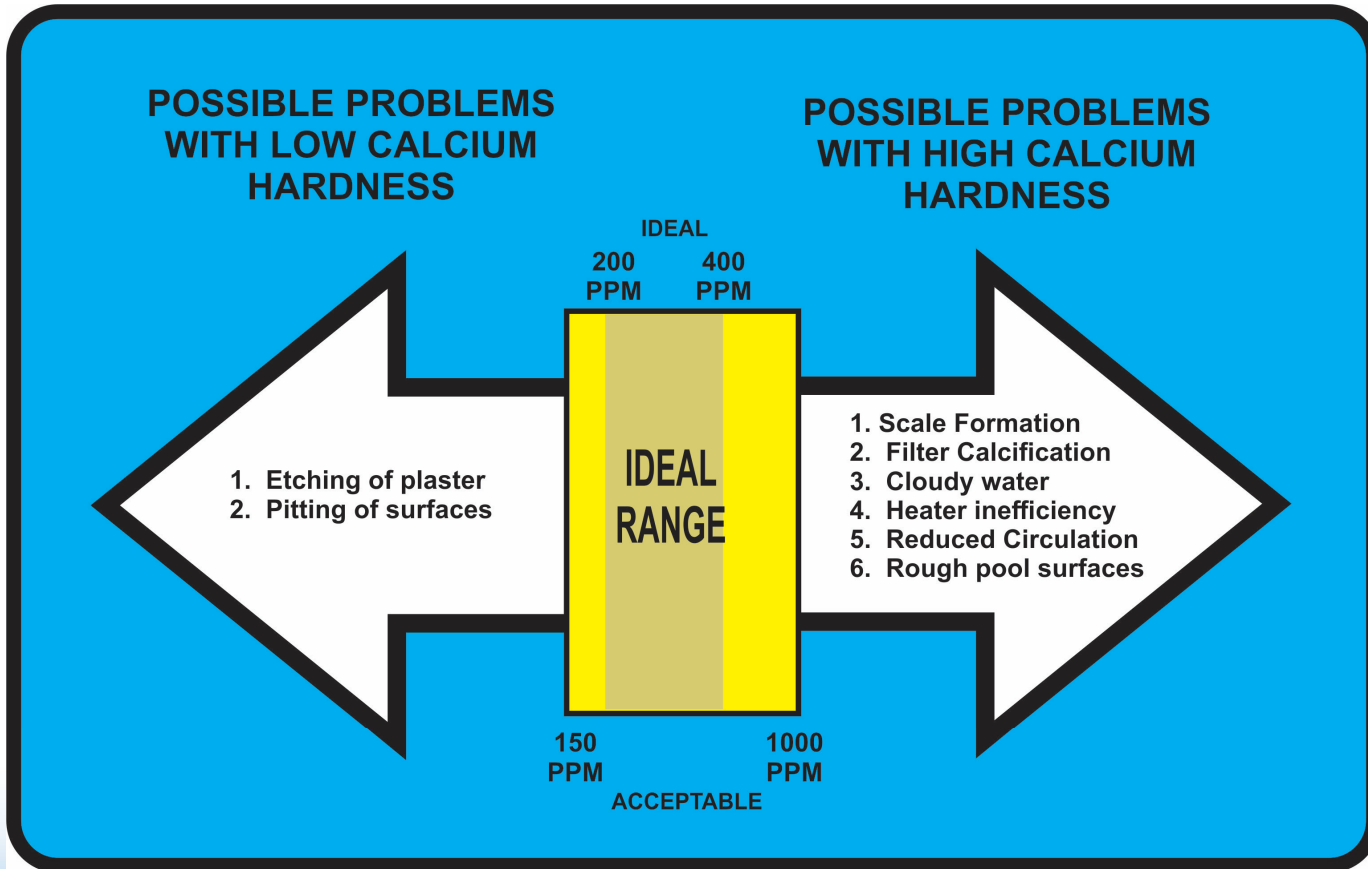
- You need to wash and fill the large comparator tube with 25 ml water sample.
    - Add 20 drops R-0010. Swirl to mix.
  - Add 5 drops R-0011L. Swirl to mix. If Calcium Hardness is present, sample will turn red.
  - Add R-0012 dropwise. After each drop, count and swirl to mix until color changes from red to blue.
    - Multiply drops in Step 4 by 10. Record as parts per million (ppm) calcium hardness as calcium carbonate.
- \* When high CH is anticipated, this procedure may be used: Use 10 mL sample, 10 drops R-0010, 3 drops R-0011, and multiply drops in Step 4 by 25



TEST ONCE PER WEEK IN A MAIN POOL AND ONCE PER DAY IN A SPA



# CALCIUM HARDNESS





Remember to make all chemical adjustments when possible at the end of the night. Calcium Chloride is also used as Ice Melt and that is not something, we want floating in the water for your swimmers to swallow!

To adjust CH the pool operator should use one of the following methods:

# CALCIUM HARDNESS ADJUSTMENT

## TO INCREASE CH

- CALCIUM CHLORIDE



## TO DECREASE CH

- REPLACE THE WATER WITH WATER THAT HAS LESS CALCIUM (DRAIN & FILL)

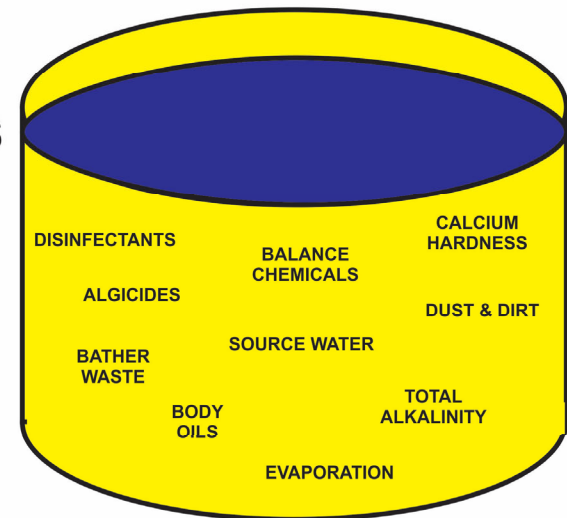


# TOTAL DISSOLVED SOLIDS

All minerals, metals and matter that can not be eliminated with a chemical. Total Dissolved Solids should be kept below 1500 ppm (unless you operating a saltwater pool) and should be checked everyday in an indoor swimming pool.

## SOURCES OF TOTAL DISSOLVED SOLIDS IN POOL WATER INCLUDE:

POTABLE WATER	200-600 PPM
BRACKISH WATER	3,000-5,000 PPM
SEA WATER	35,000 PPM
GREAT SALT LAKE	260,000 PPM



# HOT TUBS ARE A TDS PROBLEM!



Cold water climates are not generally required to test TDS as their pools are used only for a portion of the year and are drained and refilled each season. Indoor pools and hot tubs are required to be checked once per!

To adjust TDS the pool operator should use one of the following methods:

# TOTAL DISSOLVED SOLIDS ADJUSTMENT

TO DECREASE TDS

- REPLACE THE WATER WITH WATER THAT HAS LESS TDS (DRAIN & FILL)



# TEMPERATURE

- AFFECTS SOLUBILITY OF CHEMICALS
- MOST CHEMICALS MORE SOLUBLE AS TEMPERATURE INCREASES (CALCIUM CARBONATE DECREASES)
- TEMPERATURE SHOULD NOT BE CONSIDERED IN CHANGING WATER BALANCE



Ideal range for pools:

- Most pools operate between 83°F and 85°F
  - Most spas operate between 101°F and 102°F
- 78° – 82° F

Maximum recommended for

General use pools:	88° F
Therapy pools:	96° F
Spas:	104° F



## Langelier Saturation Index

$$SI = pH + F_T + F_{TA} + F_{CH} - 12.1$$

PH = PH TEST KIT READING

$F_T$  = TEMPERATURE FACTOR

$F_{TA}$  = ALKALINITY FACTOR

$F_{CH}$  = CALCIUM HARDNESS FACTOR

12.1 = STANDARD NUMBER FOR TDS

LANGELIER SATURATION INDEX

# LANGELIER SATURATION INDEX

Langelier Saturation Index

$$\text{SI} = \text{pH} + F_T + F_{TA} + F_{CH} - 12.1$$

The closer to 0 the SI is, the more balanced the water is.

Acceptable range: -0.3 to +0.5

Ideal range: -.3 to +.3

Below -0.5 – corrosion is produced

Above +0.5 – scale is produced

$$\text{Formula} = \text{pH} + \text{CF} + \text{AF} + \text{TF} - 12.1$$

Calcium Hardness		Total Alkalinity		Temperature	
PPM	CF	PPM	AF	F	TF
5-24	0.3	5-24	0.7	32-26	0.0
25-49	1.0	25-49	1.4	37-45	0.1
50-74	1.3	50-74	1.7	46-52	0.2
75-99	1.5	75-99	1.9	53-59	0.3
100-140	1.6	100-149	2.0	60-65	0.4
150-199	1.8	150-199	2.2	66-75	0.5
200-299	1.9	200-299	2.3	76-83	0.6
300-399	2.1	300-399	2.5	84-93	0.7
400-799	2.2	400-799	2.6	94-104	0.8
800-999	2.5	800-999	2.9	105-127	0.9
1000 +	2.6	1000 +	3.0	128 +	1.0



**LANGELIER  
SATURATION  
INDEX  
EXAMPLE 1**

**Langelier Saturation Index**

$$SI = pH + F_T + F_{TA} + F_{CH} - 12.1$$

PH	7.2	<u>7.2</u>
CH	200 PPM	<u>1.9</u>
TA	70 PPM	<u>1.7</u>
T	82° F	<u>0.6</u>
TDS	BELOW 1000	<u>-12.1</u>

Index is  $11.4 - 12.1 = -0.7$

**Recommendations:**

Add Soda Ash and raise pH to 7.5

Add Baking Soda and raise TA to 100



LANGELIER  
SATURATION  
INDEX  
EXAMPLE 2

Langelier Saturation Index

$$SI = pH + F_T + F_{TA} + F_{CH} - 12.1$$

PH	7.6	<u>7.6</u>
CH	320 PPM	<u>2.1</u>
TA	160 PPM	<u>2.2</u>
T	85° F	<u>0.7</u>
TDS	BELOW 1000	<u>-12.1</u>

Index is  $12.6 - 12.1 = +0.5$

Recommendations:

Add Muriatic Acid and lower pH to 7.4

This should also lower TA closer to 100.





# TAYLOR TEST KIT: DETERMINING TA WITHOUT THE MATH





## DETERMINING WHAT TO CHANGE

- THERE IS NO RIGHT ANSWER
- GENERALLY IF THE OTHER READING ARE ACCEPTABLE, THE POOL OPERATOR WILL ADJUST PH
- FOCUS ON THE IDEALS
- TYPICALLY THE POOL OPERATOR WON'T ADJUST TEMPERATURE OR TDS TO ACHIEVE WATER BALANCE



# CHEMICAL ADDITION

- IF ADDING LARGE AMOUNTS OF CHEMICALS NEED TO BE ADDED IT IS ADVISABLE TO ADD SMALLER AMOUNTS OVER SEVERAL DAYS
- ADD CHEMICALS AT NIGHT WHEN POSSIBLE WHEN NO ONE WILL BE SWIMMING. IF ADDING PRIOR TO THE SWIM DAY, THE POOL OPERATOR SHOULD DILUTE THE CHEMICAL AND KEEP THE POOL CLOSED FOR AT LEAST 30 MINUTES
- SEQUENCE FOR CHEMICAL ADJUSTMENT
  - IF TOTAL ALKALINITY NEEDS TO BE ADJUSTED, CHANGE FIRST;
  - IF CALCIUM HARDNESS NEEDS TO BE ADJUSTED, DO THIS NEXT;
  - IF PH NEEDS TO BE ADJUSTED, DO THIS NEXT (RETEST FIRST, IF ALKALINITY ALSO CHANGED)
  - IF SANITIZERS, CYA, OTHER CHEMICAL PARAMETERS NEEDS TO BE ADJUSTED THEN DO THIS LAST



# HOW MUCH CHEMICALS DO I USE?

## NEED TO KNOW:

- GALLONAGE OF POOL/SPA
- CALCULATED CHANGE
- PROPER CHEMICAL TO MAKE THE CHANGE

## NEED TO REMEMBER:

- WEAR YOUR PERSONAL PROTECTIVE EQUIPMENT
- REFER TO YOUR SAFETY DATA SHEETS (A.K.A. MATERIAL SAFETY DATA SHEETS)
- CHEMICALS GO INTO WATER (NEVER WATER INTO CHEMICALS)
- READ THE DIRECTIONS



# DOSAGES REQUIRED TO CHEMICALLY TREAT 10,000 GALLONS OF WATER

This information can be found in your book, in the Taylor Test Kit, and on the packaging

## **PARAMETER – CHEMICAL**

### **DECREASE PH**

AMOUNT OF MURIATIC ACID TO CHANGE FROM 7.8 TO 7.6

## **REQUIRED DOSAGE**

12 FLUID OZS.

### **INCREASE PH**

AMOUNT OF SODA ASH TO CHANGE PH FOR 7.0 TO 7.4

12 OZS.

### **DECREASE TOTAL ALKALINITY**

MURIATIC ACID TO REDUCE 10 PPM

1.6 PTS.

MURIATIC ACID TO REDUCE 50 PPM

1.0 GALLON

MURIATIC ACID TO REDUCE 100 PPM

2.0 GALLONS

### **INCREASE TOTAL ALKALINITY**

SODIUM BICARBONATE TO INCREASE 10 PPM

1.5 LBS.

SODIUM BICARBONATE TO INCREASE 50 PPM

7.5 LBS.

SODIUM BICARBONATE TO INCREASE 100 PPM

15 LBS.

### **INCREASE CALCIUM HARDNESS**

CALCIUM CHLORIDE (100%) TO INCREASE 10 PPM

1 LB.

CALCIUM CHLORIDE (100%) TO INCREASE 50 PPM

5 LBS.

CALCIUM CHLORIDE (100%) TO INCREASE 100 PPM

10 LBS.



LANGELIER  
SATURATION  
INDEX  
ADJUSTING  
EXAMPLE 1  
AND  
EXAMPLE 2

In a pool that is 100,000 gallons

IN EXAMPLE 1:

ADD 90 OZ (75% OF 120 OZ) TO RAISE PH  
FROM 7.2 TO 7.5. THIS WILL ALSO BUMP  
THE TA ALSO.

IN EXAMPLE 2:

ADD 120 FL OZ TO LOWER PH & TA



# LET'S REVIEW:

## PH

7.2 TO 7.8

CHECK EVERY TWO HOURS



Muriatic  
Acid



Soda  
Ash

## TOTAL ALKALINITY

60 TO 180 PPM

CHECK ONCE PER WEEK



Muriatic  
Acid



Baking  
Soda

## CALCIUM HARDNESS

150 TO 400

CHECK ONCE PER WEEK



Drain &  
Fill



Calcium  
Chloride

## TOTAL DISSOLVED SOLIDS

UNDER 1500 PPM

CHECK ONCE PER WEEK



Drain &  
Fill



Do not  
Raise

## TEMPERATURE

UNDER 88 FOR MAIN POOL

UNDER 104 FOR SPA





**CHAPTER  
THREE**

**PREVENTING  
DISEASE  
TRANSMISSION**



# RECREATIONAL WATER ILLNESSES

*Disease transmission seldom occurs in swimming pools in the United States. Aquatic facilities that are maintained with an appropriate disinfectant like chlorine will eliminate the source of disease before infection can occur. In the rare circumstance that a pool does become the pathway for disease transmission usually there has been a breakdown in the disinfectant levels at the facility. A pool should be considered this pathway if multiple patrons are affected with the same disease or show similar symptoms.*

- THE POOL OPERATOR IS RESPONSIBLE FOR ENSURING THE FACILITY IS SAFE AND HEALTHY
- THE POOL OPERATOR'S MAIN SOURCE OF CONTAMINANTS IS THE SWIMMERS



## HOW DO MICROORGANISMS END UP IN THE POOL?

- UP TO 38,000 MICRO-ORGANISMS ARE DISCHARGED WITH CLEARING OF THE NOSE.
- 100 MILLION TO 1 BILLION MICRO-ORGANISMS ARE DISCHARGED FROM 1 GRAM (1/3 OF AN OUNCE) OF SALIVA.
- AN ACTIVE SWIMMER CAN CONTRIBUTE 1-2 PINTS OF BODILY FLUIDS PER HOUR. (TO INCLUDE PERSPIRATION, URINE, SALIVA, NOSE DISCHARGE ETC.)
- HANDS MAY SHED 1/2 TO 5 MILLION ORGANISMS IN A SINGLE WASHING.
- A PERSON SHEDS A LAYER OF DEAD SKIN EACH TIME THEY TAKE A SHOWER, GET INTO A POOL, ETC.

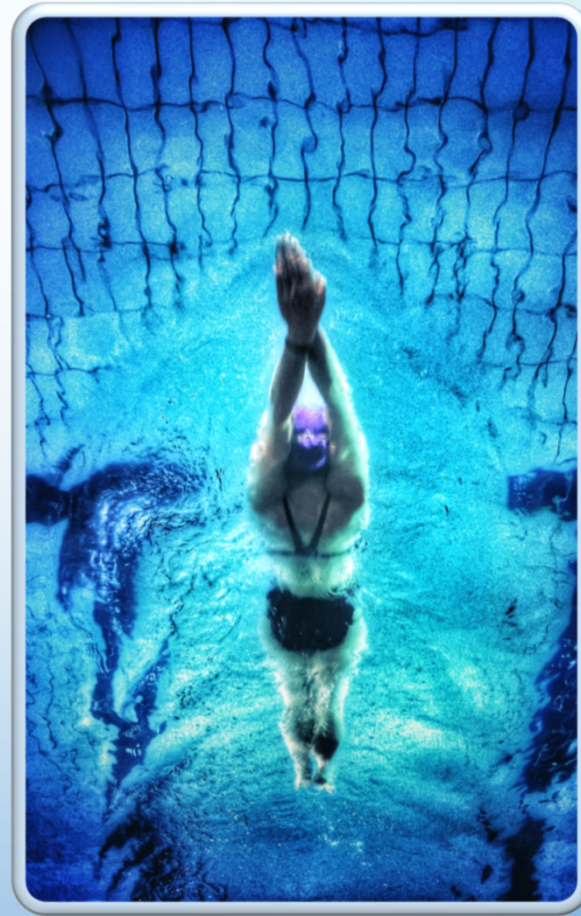


Photo by sabrina  
shulz



# RECREATIONAL WATER ILLNESSES

## DISEASES OF CONCERN



Photo by Sergio Souza

### BACTERIA

- PSEUDOMONAS AERUGINOSA
- E. COLI
- STAPHYLOCOCCI
- STREPTOCOCCI

### VIRUSES

- HEPATITIS A
- LEGIONELLA
- NORWALK-LIKE
- HIV

### PARASITES

- GIARDIA
- CRYPTOSPORIDIUM PARVUM



# RECREATIONAL WATER ILLNESSES PREVENTION



Photo by  
VisionPic.Net

- PROHIBIT USE BY PERSONS WITH OPEN WOUNDS OR KNOWN INFECTIONS
- TREAT AND TEST THE WATER ACCORDING TO CODE
- REQUIRE SOAP SHOWERS PRIOR TO ENTRY
- CONTROL INTRODUCTION OF DIRT AND INFECTIOUS MATERIAL BY MAKING SURE SWIM CLOTHES ARE CLEAN
- REMOVE DEBRIS FROM BOTTOM
- MAINTAIN CIRCULATION AND FILTRATION
- DISINFECT LOCKER ROOMS, TOILET AREAS DAILY; CHECK HOURLY
- IF PROVIDING TOWELS, BE SURE THEY ARE PROPERLY LAUNDERED
- WORK WITH YOUR LOCAL HEALTH DEPARTMENT



# HOW IS DISEASE TRANSMITTED:

IN ORDER FOR A DISEASE TO BE TRANSMITTED, THE FOLLOWING FOUR ELEMENTS MUST BE PRESENT:

1. A **PATHOGEN** MUST BE PRESENT
2. THERE IS **ENOUGH OF THAT PATHOGEN** PRESENT TO CAUSE A DISEASE
3. A PERSON MUST BE **SUSCEPTIBLE** TO THE PATHOGEN
4. THE PATHOGEN PASSES THROUGH AN **ENTRY SITE** AND INTO THE BODY

*(WITHOUT EACH OF THESE FOUR CRITERIA, INFECTION CANNOT OCCUR)*

IN ADDITION, THERE ARE FOUR WAYS FOR A PATHOGEN TO ENTER THE BODY:

1. **DIRECT CONTACT** - TOUCHING ANOTHER PERSON'S BODY FLUIDS
2. **INDIRECT CONTACT** - TOUCHING A CONTAMINATED OBJECT
3. **AIRBORNE** - INHALING INFECTED DROPLETS SUCH AS A SNEEZE OR COUGH
4. **VECTOR-BORNE** - SKIN PIERCED BY AN INFECTIOUS MATERIAL (I.E. INSECT STINGS, ANIMAL BITES)



According to the Center for Disease Controls website:

*"The germs that cause recreational water illnesses (RWIs) can be spread when swallowing water that has been contaminated with fecal matter (poop). How? If someone has diarrhea, that person can easily contaminate an entire aquatic venue. The water is shared by all swimmers. It's communal bathing water. It's not sterile. It's not drinking water."*

*"The good news is that most germs causing RWIs are killed by chlorine within a few minutes. However, chlorine doesn't kill everything right away. It takes time to kill germs and some germs, such as Cryptosporidium (or Crypto for short), can live in properly treated water for days. This means even the best maintained aquatic venues can spread germs."*

Germ	Time
E. coli O157:H7 Bacterium	Less than 1 minute
Hepatitis A Virus	About 16 minutes
Giardia Parasite	About 45 minutes
Crypto Parasite	About 15,300 minutes or 10.6 days <sup>1</sup>



# WHAT TO DO ABOUT BLOOD & VOMIT ACCORDING TO THE CDC WEBSITE

## VOMIT

“VOMITING IN THE POOL WHILE SWIMMING IS A COMMON EVENT. OFTEN, VOMITING RESULTS FROM SWALLOWING TOO MUCH WATER, MEANING THAT THE VOMIT IS PROBABLY NOT INFECTIOUS. HOWEVER, IF THE CONTENTS OF THE STOMACH ARE VOMITED, IT IS IMPORTANT TO ACT IMMEDIATELY.

### **RESPONDING TO A VOMIT INCIDENT (WHEN VOMIT CONTAINS MORE THAN REGURGITATED WATER)**

THE GERMS MOST LIKELY TO BE SPREAD BY VOMIT ARE NOROVIRUSES (ALSO KNOWN AS NORWALK-LIKE VIRUSES).

RESPOND TO THE VOMIT INCIDENT AS YOU WOULD RESPOND TO A FORMED FECAL INCIDENT, USING [CDC'S FECAL INCIDENT RESPONSE RECOMMENDATIONS FOR POOL STAFF CDC-PDF\[PDF – 4 PAGES\]](#). THE TIME AND CHLORINE LEVEL COMBINATIONS NEEDED TO KILL NOROVIRUSES AND *GIARDIA* ARE SIMILAR. SINCE KILLING *GIARDIA* IS THE BASIS FOR CDC'S FORMED FECAL INCIDENT RESPONSE RECOMMENDATIONS, THIS PROTOCOL SHOULD BE ADEQUATE FOR DISINFECTING A POTENTIALLY INFECTIOUS VOMIT INCIDENT.”

## BLOOD

“GERMS FOUND IN BLOOD (FOR EXAMPLE, HEPATITIS B VIRUS OR HIV) ARE SPREAD WHEN INFECTED BLOOD OR CERTAIN BODY FLUIDS GET INTO THE BODY AND BLOODSTREAM (FOR EXAMPLE, BY SHARING NEEDLES OR BY SEXUAL CONTACT). CHLORINE KILLS GERMS FOUND IN BLOOD AND CDC IS NOT AWARE OF ANY INSTANCES IN WHICH A PERSON HAS BECOME INFECTED WITH THESE GERMS AFTER BEING EXPOSED TO A BLOOD SPILL IN A POOL.

### **DOES CHLORINE KILL THE GERMS IN BLOOD?**

YES. THESE GERMS DO NOT SURVIVE LONG WHEN DILUTED INTO PROPERLY CHLORINATED POOL WATER.

### **SWIMMERS WANT SOMETHING TO BE DONE AFTER A BLOOD SPILL. SHOULD THE POOL BE CLOSED FOR A SHORT PERIOD OF TIME?**

THERE IS NO PUBLIC HEALTH REASON TO RECOMMEND CLOSING THE POOL AFTER A BLOOD SPILL. HOWEVER, SOME POOL STAFF CHOOSE TO DO SO TEMPORARILY TO SATISFY PATRONS.”

PLEASE REFER TO [HEALTHYSWIMMING.ORG](https://healthyswimming.org) FOR MORE INFORMATION FROM THE CDC





# RECREATIONAL WATER ILLNESSES FECAL ACCIDENT PREVENTION

- **PLEASE** DON'T SWIM WHEN YOU HAVE DIARRHEA...THIS IS ESPECIALLY IMPORTANT FOR KIDS IN DIAPERS. YOU CAN SPREAD GERMS INTO THE WATER AND MAKE OTHER PEOPLE SICK.
- **PLEASE** WEAR SWIM DIAPERS
- **PLEASE** DON'T SWALLOW THE POOL WATER. IN FACT, TRY YOUR BEST TO AVOID EVEN HAVING WATER GET IN YOUR MOUTH.
- **PLEASE** WASH YOUR HANDS WITH SOAP AND WATER AFTER USING THE TOILET OR AFTER CHANGING DIAPERS. YOU CAN PROTECT OTHERS BY REALIZING THAT GERMS ON YOUR BODY END UP IN THE WATER.

Photo by edneil  
Jocusol



*Regardless of the type of diaper, rubber pants with elastic around the legs and waist must be worn!*



# RECREATIONAL WATER ILLNESSES

## FECAL ACCIDENT RESPONSE

(CHECK WITH YOUR LOCAL HEALTH DEPARTMENT FOR YOUR EXACT PROCEDURES!)

### DIARRHEA OR LOOSE-STOOL DISCHARGE:

- DIRECT EVERYONE TO LEAVE POOL
- REMOVE FECAL MATERIAL
- RAISE FREE CHLORINE TO AT LEAST 20.0 PPM; MAINTAIN PH 7.2 – 7.5
- MAINTAIN FOR AT LEAST 8 HOURS
- DOCUMENT CLOSURE ON LOG SHEETS
- REOPEN ONCE FREE CHLORINE IN ACCEPTABLE RANGE

### SOLID FECAL DISCHARGE:

- DIRECT EVERYONE TO LEAVE POOL
- REMOVE FECAL MATERIAL
- RAISE FAC TO 10 PPM IN THE IMMEDIATE AREA OF THE CONTAMINATION (10 FT BY 10 FT)
- RAISE FREE CHLORINE TO AT LEAST 2.0 PPM; MAINTAIN PH 7.2 – 7.5
- MAINTAIN FOR AT LEAST 30 MIN.
- DOCUMENT CLOSURE ON LOG SHEETS
- REOPEN WHEN DISINFECTION LEVEL IS ACCEPTABLE

# Document The Event


## WATER CONTAMINATION RESPONSE LOG

Person Conducting Contamination Response	
Supervisor on Duty	
Date (mm/dd/yyyy) of Incident Response	
Time of Incident Response	
Water Feature or Area Contaminated	
Number of People in Water	
Type/Form of Contamination in Water: Fecal Accident (Formed Stool or Diarrhea), Vomit, Blood	
Time that Water Feature was Closed	
Stabilizer Used in Water Feature (Yes/No)	
	Water Quality Measurement:
	Level at Closure    1    2    3    4    Level Prior to Reopening
Free Residual Chlorine (1-4 are measurements: spread evenly thru the closure time)	
pH (1-4 are measurements: spread evenly thru the closure time)	
Date (mm/dd/yyyy) that Water Feature was Reopened	
Time that Water Feature was Reopened	
Total Contact Time (Time from when disinfectant reached desired level to when disinfectant levels were reduced prior to opening)	
Remediation Procedure(s) Used and Comments/Notes:	

HEALTHY SWIMMING

### Fecal Incident Response Recommendations for Aquatic Staff

What do you do when you find poop in the water?




Check for existing guidelines from your local or state regulatory agency before use. CDC recommendations do not replace existing state or local regulations or guidelines.

These recommendations are for responding to fecal incidents in chlorinated aquatic venues (for example, pools and water playgrounds).

Improper handling of chlorine-based disinfectants can cause injury. Follow proper occupational safety and health requirements when following these recommendations. For more pool chemical safety information, visit [www.cdc.gov/healthyswimming/aquatic-professionals/preventing-pool-chemical-events.html](http://www.cdc.gov/healthyswimming/aquatic-professionals/preventing-pool-chemical-events.html)

**CLOSURES:** Fecal incidents are a concern and an inconvenience to both aquatic staff and patrons. Aquatic staff should carefully explain to patrons why the aquatic venue needs to be closed in response to a fecal incident. Explaining the reasons for closing the venue (for proper disinfection and protection of swimmer health) is likely to promote patron understanding and increase their frustration. Closures allow chlorine to do its job—kill germs and help prevent recreational water illnesses (RWIs).

Hot tubs/pools and some water playgrounds can have much smaller amounts of water in response to formed or diarrheal fecal incidents in small-volume venues; it might be more efficient to completely drain as much water as possible from the incident and associated plumbing scrub and clean all accessible surfaces in contact with contaminated water, replace or clean filter media when appropriate, and refill with uncontaminated water from an approved source (for example, municipal water system).

 U.S. Department of Health and Human Services  
Centers for Disease Control and Prevention

Released June 22, 2018


### What do I do about... formed fecal matter (poop) in the water?

Formed fecal incidents pose a risk of spreading germs, including moderately chlorine-tolerant *Clostridia*. To disinfect the water following a formed fecal incident, aquatic staff should follow the steps below, which are based on killing or inactivating *Clostridia*.

1. Close the aquatic venue to swimmers. If you have multiple venues that use the same filtration system—all of the venues will have to be closed to swimmers. Do not allow anyone to enter the venue(s) until the disinfection process is completed.
2. Remove as much of the fecal matter as possible (for example, using a net or bucket) and dispose of the fecal matter in a sanitary manner. Clean and disinfect the item used to remove the fecal matter (for example, after cleaning, leave the net or bucket immersed in the water during disinfection). **VACUUMING FECAL MATTER FROM THE WATER IS NOT RECOMMENDED.**
3. Using unabsorbed chlorine (for example, sodium hypochlorite), raise the water's free chlorine concentration to 2 parts per million (ppm) if less than 2 ppm. Maintain free chlorine concentration at 2 ppm and water at pH 7.5 or less for 20–30 minutes. Other concentrations or closure times can be used (see Table below). State or local regulations may require higher free chlorine concentration in the presence of chlorine stabilizer<sup>1</sup>, which are known to slow the rate at which free chlorine inactivates or kills germs.
4. Confirm that the filtration system is operating while the water reaches and is maintained at the proper free chlorine concentration and pH for disinfection.
5. Allow swimmers back into the water only after the disinfection process has been completed and the free chlorine concentration and pH are within the operating range allowed by the state or local regulatory authority.

**Establish a fecal incident log.** Document each fecal incident by recording date and time of the event, whether it involved formed fecal matter or diarrhea, and the free chlorine concentration and pH at the time of observation of the event. Before reopening the aquatic venue, record the procedures followed in response to the fecal incident (including the process used to adjust free chlorine concentration and pH). If necessary, the free chlorine concentration and pH and the disinfection time. You can download a Water Contamination Response Log at <http://www.cdc.gov/healthyswimming/aquatic-professionals/fecal-incident-response.html>

Guides Kill or Inactivation Time for a Formed Fecal Incident		
Free Chlorine Concentration (ppm)	Disinfection Time <sup>2</sup>	
1.0	45 minutes	
2.0	25–30 minutes	
3.0	15 minutes	



1. Chlorine stabilizers include compounds such as cyanuric acid, dichloro and trichloro.
2. Based on the water temperature (based on 70°F [20°C]) or higher during the disinfection process.
3. Alternative options could include circulating the water through a secondary disinfection system (for example, ultraviolet light or ozone) to biologically reduce the number of *Clostridia* species in the aquatic venue(s) before use (see [www.cdc.gov/healthyswimming/aquatic-professionals/secondary-disinfection.html](http://www.cdc.gov/healthyswimming/aquatic-professionals/secondary-disinfection.html)).
4. CDC does not recommend using the net for *Cryptosporidium* after hyperchlorination is completed through hyperchlorination during *Cryptosporidium* incident. Do not reuse the net for the duration of the outbreak of the event.
5. Based on a study comparing the use of *Cryptosporidium* parvum under chlorinated recreational water conditions (Water Health, 2008;10:173–76).

### What do I do about... diarrhea in the water when chlorine stabilizer<sup>1</sup> is NOT in the water?

A diarrheal incident is a high-risk event for contamination caused by *Cryptosporidium* (or "Crypto"), an extremely chlorine-tolerant parasite. Therefore, it is important that aquatic staff educate patrons not to swim when ill with diarrhea. To disinfect the water following a diarrheal incident, aquatic staff should hyperchlorinate or raise the free chlorine concentration to a high concentration for a long period of time. If necessary, before attempting to hyperchlorinate, consult an aquatic professional to determine the feasibility, the most optimal and practical methods, and needed safety considerations.

1. Close the aquatic venue to swimmers. If you have multiple venues that use the same filtration system—all of the venues will have to be closed to swimmers. Do not allow anyone to enter the venue(s) until the hyperchlorination process is completed.
2. Remove as much of the fecal matter as possible (for example, using a net or bucket) and dispose of the fecal matter in a sanitary manner. Clean and disinfect the item used to remove the fecal matter (for example, after cleaning, leave the net or bucket immersed in the water during hyperchlorination).
3. Using unabsorbed chlorine (for example, sodium hypochlorite), raise the water's free chlorine concentration (see Table below) and maintain water at pH 7.5 or less.<sup>2</sup>

**Establish a fecal incident log.** Document each fecal incident by recording date and time of the event, whether it involved formed fecal matter or diarrhea, and the free chlorine concentration and pH at the time of observation of the event. Before reopening the aquatic venue, record the procedures followed in response to the fecal incident (including the process used to adjust free chlorine concentration and pH). If necessary, the free chlorine concentration and pH and the disinfection time. You can download a Water Contamination Response Log at <http://www.cdc.gov/healthyswimming/aquatic-professionals/fecal-incident-response.html>

Use the formula below to calculate the time required to inactivate or kill <i>Cryptosporidium</i>		
Concentration of available free chlorine (ppm)	Free chlorine concentration (parts per million)	Time (in minutes)
15.000	+ 20 <sup>3</sup>	= 780 (or 12.75 hours)
15.300	+ 10	= 1,530 (or 25.5 hours)

1. Chlorine stabilizers include compounds such as cyanuric acid, dichloro and trichloro.
2. Based on the water temperature (based on 70°F [20°C]) or higher during the hyperchlorination process.
3. Alternative options could include circulating the water through a secondary disinfection system (for example, ultraviolet light or ozone) to biologically reduce the number of *Cryptosporidium* species in the aquatic venue(s) before use (see [www.cdc.gov/healthyswimming/aquatic-professionals/secondary-disinfection.html](http://www.cdc.gov/healthyswimming/aquatic-professionals/secondary-disinfection.html)).
4. CDC does not recommend using the net for *Cryptosporidium* after hyperchlorination is completed through hyperchlorination during *Cryptosporidium* incident. Do not reuse the net for the duration of the outbreak of the event.
5. Based on a study comparing the use of *Cryptosporidium* parvum under chlorinated recreational water conditions (Water Health, 2008;10:173–76).

### What do I do about... diarrhea in the water when chlorine stabilizer<sup>1</sup> is in the water?

A diarrheal incident is a high-risk event for contamination caused by *Cryptosporidium* (or "Crypto"), an extremely chlorine-tolerant parasite. Therefore, it is important that aquatic staff educate patrons not to swim when ill with diarrhea. To disinfect the water following a diarrheal incident, aquatic staff should hyperchlorinate or raise the free chlorine concentration to a high concentration for a long period of time. If necessary, before attempting to hyperchlorinate, consult an aquatic professional to determine the feasibility, the most optimal and practical methods, and needed safety considerations.

1. Close the aquatic venue to swimmers. If you have multiple venues that use the same filtration system—all of the venues will have to be closed to swimmers. Do not allow anyone to enter the venue(s) until the hyperchlorination process is completed.
2. Remove as much of the fecal matter as possible (for example, using a net or bucket) and dispose of the fecal matter in a sanitary manner. Clean and disinfect the item used to remove the fecal matter (for example, after cleaning, leave the net or bucket immersed in the water during hyperchlorination).
3. Using unabsorbed chlorine (for example, sodium hypochlorite), raise the water's free chlorine concentration (see Table below) and maintain water at pH 7.5 or less.<sup>2</sup>
4. Hyperchlorinate. Chlorine stabilizer slows the rate at which free chlorine concentration and pH are within the operating range allowed by the state or local regulatory authority.

**Establish a fecal incident log.** Document each fecal incident by recording date and time of the event, whether it involved formed fecal matter or diarrhea, and the free chlorine concentration and pH at the time of observation of the event. Before reopening the aquatic venue, record the procedures followed in response to the fecal incident (including the process used to adjust free chlorine concentration and pH). If necessary, the free chlorine concentration and pH and the disinfection time. You can download a Water Contamination Response Log at <http://www.cdc.gov/healthyswimming/aquatic-professionals/fecal-incident-response.html>

Use the formula below to calculate the time required to inactivate or kill <i>Cryptosporidium</i>		
Concentration of available free chlorine (ppm)	Free chlorine concentration (parts per million)	Time (in minutes)
15.000	+ 20 <sup>3</sup>	= 780 (or 12.75 hours)
15.300	+ 10	= 1,530 (or 25.5 hours)

1. Chlorine stabilizers include compounds such as cyanuric acid, dichloro and trichloro.
2. Based on the water temperature (based on 70°F [20°C]) or higher during the hyperchlorination process.
3. Alternative options could include circulating the water through a secondary disinfection system (for example, ultraviolet light or ozone) to biologically reduce the number of *Cryptosporidium* species in the aquatic venue(s) before use (see [www.cdc.gov/healthyswimming/aquatic-professionals/secondary-disinfection.html](http://www.cdc.gov/healthyswimming/aquatic-professionals/secondary-disinfection.html)).
4. CDC does not recommend using the net for *Cryptosporidium* after hyperchlorination is completed through hyperchlorination during *Cryptosporidium* incident. Do not reuse the net for the duration of the outbreak of the event.
5. Based on a study comparing the use of *Cryptosporidium* parvum under chlorinated recreational water conditions (Water Health, 2008;10:173–76).

# HEALTHYSWIMMING.ORG IS HERE TO HELP!

## THE CDC'S WEBSITE HAS GREAT INFORMATION



# OCCUPATIONAL SAFETY AND HEALTH ADMINISTRATION (OSHA)

PREVENTING DISEASE TRANSMISSION AT YOUR  
FACILITY INCLUDES KEEPING YOUR EMPLOYEES  
SAFE IN THEIR WORKPLACE!

The U.S. Department of Labor, OSHA, governs many aspects of worker safety. One of those areas is protection of workers against transmission of disease.

The following information is reprinted from "Occupational Exposure to Blood borne Pathogens: Precautions for Emergency Responders", a publication from the U.S. Department of Labor, Occupational Safety and Health Administration. OSHA 3130, 1992, (reprinted).

- OSHA recognizes the need for a regulation that prescribes safeguards to protect workers against the health hazards from exposure to blood and certain body fluids containing blood borne pathogens and to reduce their risk of this exposure.
- OSHA's blood borne pathogens standard requires the employer to prepare a written exposure control program.
- The standard applies to every employer with one or more employees who can reasonably be expected to come into contact with blood and other specified body fluids in performing their duties.
- Occupational exposure means a "reasonably anticipated skin, eye, mucous membrane, or parenteral contact with blood or other potentially infectious materials that may result from the performance of the employees' duties."
- A written exposure control plan is necessary for the safety and health of workers. Employees must develop a plan that identifies and documents the tasks, procedures, and job classifications covering instances where there is exposure to blood or other potentially infectious materials.
- As required under the standard, a written exposure plan is required that provides documentation of the following key elements.
  - Identification of job classification
  - A schedule of how and when the provisions of the standard will be implemented
  - Procedures for evaluating the circumstances of an exposure incident.
- All persons with a potential for exposure must be provided with adequate training and information including general explanations of the modes of transmission, symptoms, epidemiology, warning signals relating to possible exposure, and procedures to follow if exposure occurs.
- Employers must make available, free of charge and at a reasonable time and place, the Hepatitis B vaccine and vaccination series to all employees who are at risk of occupational exposure.
- Employees should immediately report exposure incidents. Employers must treat reports in the strictest confidence. At the time of the exposure incident, the exposed employee must be directed to a health care professional.
- There are 2 types of employee-related records, required by the blood borne pathogens standard: medical and training. The blood borne pathogens standard also requires employers to maintain and to keep accurate training records for 3 years and to include the following:
  - Training dates
  - Content or a summary of the training
  - Names and qualifications of trainers(s), and
  - Names and job titles of trainees.

For a complete copy of this or any other free OSHA publication, contact them at:

U.S. Dept. of Labor, OSHA, 200 Constitution Avenue, Washington, DC 20210

1-202-693-1888, 1-800-321-OSHA

[www.osha.gov](http://www.osha.gov), for over 300 articles on the bloodborne pathogen standard



## **EMPLOYEES HAVE THE RIGHT TO:**

- THEY MAY COME IN CONTACT WITH BLOOD IN THEIR WORKPLACE
- THEY MAY COME IN CONTACT WITH DANGEROUS CHEMICALS IN THEIR WORKPLACE
- PERSONAL PROTECTIVE EQUIPMENT IS PROVIDED BY THEIR EMPLOYER
- MATERIAL SAFETY DATA SHEETS ARE AVAILABLE FOR ALL THE CHEMICALS THEY USE

## **EMPLOYEES HAVE THE RESPONSIBILITY TO:**

- KNOW WHERE TO GET INFORMATION ON HAZARDOUS CHEMICALS USED ON THE JOB.
- LEARN TO READ AND UNDERSTAND LABELS AND MSDS.
- IDENTIFY CHEMICAL HAZARDS BEFORE STARTING A JOB AND ASK QUESTIONS.
- USE PERSONAL PROTECTIVE EQUIPMENT AND FOLLOW SAFE WORK PRACTICES.
- FOLLOW EMPLOYER'S PROCEDURES FOR DISPOSAL AND CLEAN UP.
- LEARN EMERGENCY PROCEDURES.
- EAT, DRINK, APPLY COSMETICS OR SMOKE AWAY FROM HAZARDOUS CHEMICALS





## CHAPTER FOUR

# DISINFECTION

- **DISINFECTION** - (SANITIZE): THE PROCESS OF DESTROYING LIVING MICROORGANISMS AND BACTERIA IN SUFFICIENT NUMBERS (BY DEFINITION – 99.9%) TO PREVENT THE TRANSMISSION OF DISEASE. (REFERENCE 1)
- **DISINFECTANT** – (SANITIZER): THE CHEMICAL OR DEVICE THAT KILLS OR INACTIVATES THE MICROORGANISMS PRESENT IN POOL/SPA WATER (REFERENCE 4). THE MOST COMMON IN THE INDUSTRY ARE CHLORINE AND BROMINE



**People like a pool that looks good. The bluish color comes from the light refraction on the water. That is why the deep end looks bluer than the shallow. Poorly taken care of water can become cloudy or even turn green.**

**A pool that has been balanced according to the Saturation Index will be blue and look clear and inviting. Now it is time to add disinfection. Once people begin using the pool, disinfection becomes critical. A swimming pool is not the source of the disease, but rather the pathway of transmission if the disinfectant levels are NOT kept in the proper ranges as recommended by the local Health Department. The prevention of and subsequent elimination of the possible spread of disease is one of the PSPO's main jobs.**

**Slide 2**

---

**RF1**

Rob Fox, 3/21/2020



Photo by Ezequiel de silva

## WATER BALANCE IS IMPORTANT!

- DISINFECTANTS WILL LAST LONGER IN BALANCED WATER
- BATHER DISCOMFORT IS USUALLY CAUSED BY INCORRECT PH

# SANITATION CHEMISTRY OF CHLORINATION

## HALOGEN FAMILY



**Periodic Table of the Elements**

1 1A 11A	2 IIA 2A											13 IIIA 3A	14 IVA 4A	15 VA 5A	16 VIA 6A	17 VIIA 7A	18 VIIIA 8A	
1 H Hydrogen 1.008																	2 He Helium 4.003	
3 Li Lithium 6.941	4 Be Beryllium 9.012											5 B Boron 10.811	6 C Carbon 12.011	7 N Nitrogen 14.007	8 O Oxygen 15.999	9 F Fluorine 18.998	10 Ne Neon 20.180	
11 Na Sodium 22.990	12 Mg Magnesium 24.305	3 IIIB 3B	4 IVB 4B	5 VB 5B	6 VIB 6B	7 VIIB 7B	8 VIII 8		9 VIII 8	10	11 IB 1B	12 IIB 2B	13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.066	17 Cl Chlorine 35.453	18 Ar Argon 39.948
19 K Potassium 39.098	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.88	23 V Vanadium 50.942	24 Cr Chromium 51.996	25 Mn Manganese 54.938	26 Fe Iron 55.833	27 Co Cobalt 58.933	28 Ni Nickel 58.693	29 Cu Copper 63.546	30 Zn Zinc 65.39	31 Ga Gallium 69.723	32 Ge Germanium 72.61	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80	
37 Rb Rubidium 84.468	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94	43 Tc Technetium 98.907	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.906	46 Pd Palladium 106.42	47 Ag Silver 107.868	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.71	51 Sb Antimony 121.760	52 Te Tellurium 127.6	53 I Iodine 126.904	54 Xe Xenon 131.29	
55 Cs Cesium 132.905	56 Ba Barium 137.327	57-71 Lanthanide Series		72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.85	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.22	78 Pt Platinum 195.08	79 Au Gold 196.967	80 Hg Mercury 200.59	81 Tl Thallium 204.383	82 Pb Lead 207.2	83 Bi Bismuth 208.980	84 Po Polonium [209]	85 At Astatine [210]	86 Rn Radon 222.018
87 Fr Francium 223.020	88 Ra Radium 226.025	89-103 Actinide Series		104 Rf Rutherfordium [261]	105 Db Dubnium [262]	106 Sg Seaborgium [266]	107 Bh Bohrium [264]	108 Hs Hassium [265]	109 Mt Meitnerium [268]	110 Ds Darmstadtium [271]	111 Rg Roentgenium [272]	112 Cn Copernicium [277]	113 Uut Ununtrium [289]	114 Fl Flerovium [289]	115 Uup Ununpentium [289]	116 Lv Livermorium [293]	117 Uus Ununseptium [294]	118 Uuo Ununoctium [294]
		57 La Lanthanum 138.905	58 Ce Cerium 140.115	59 Pr Praseodymium 140.908	60 Nd Neodymium 144.24	61 Pm Promethium 144.913	62 Sm Samarium 150.36	63 Eu Europium 151.965	64 Gd Gadolinium 157.25	65 Tb Terbium 158.925	66 Dy Dysprosium 162.50	67 Ho Holmium 164.930	68 Er Erbium 167.26	69 Tm Thulium 168.934	70 Yb Ytterbium 173.04	71 Lu Lutetium 174.967		
		89 Ac Actinium 227.028	90 Th Thorium 232.038	91 Pa Protactinium 231.036	92 U Uranium 238.029	93 Np Neptunium 237.048	94 Pu Plutonium 244.064	95 Am Americium 243.061	96 Cm Curium 247.070	97 Bk Berkelium 247.070	98 Cf Californium 251.080	99 Es Einsteinium [254]	100 Fm Fermium 257.095	101 Md Mendelevium 258.1	102 No Nobelium 259.101	103 Lr Lawrencium [262]		
		Alkali Metal	Alkaline Earth	Transition Metal	Semimetal	Nonmetal	Basic Metal	Halogen	Noble Gas	Lanthanide	Actinide							

**Halogen:** Any element found in the Group VII of the periodic table. Due to their tremendous chemical reactivity, the halogens never occur free in nature, they must be prepared from their stable salt (i.e. sodium chloride [NaCl]). Chlorine, (considered to be the best oxidizer and algacide) bromine, and iodine are examples of halogens.





Photo by PIXABAY

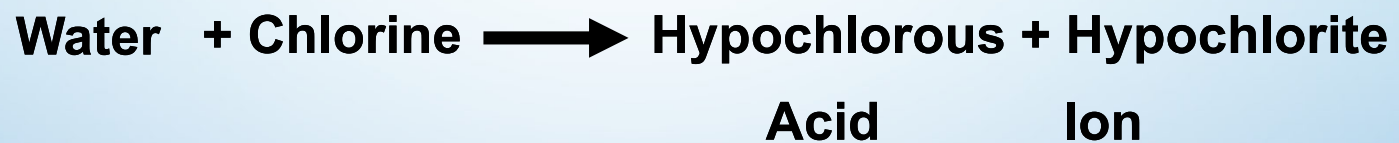
**CHLORINE IS THE  
BEST BECAUSE IT IS  
THE BEST OXIDIZER!**

# OXIDATION

- OXIDATION IS THE REMOVAL OF ORGANIC AND INORGANIC MATTER NOT DESIRED IN THE WATER
- OXIDATION CAN BE DONE WITH:
  - CHLORINE
  - POTASSIUM MONOPERSULFATE
  - OZONE
  - FORMULATED PRODUCTS
  - HYDROGEN PEROXIDE



## CHEMISTRY OF CHLORINATION



### FREE AVAILABLE CHLORINE (FAC)

- FORMED WHEN CHLORINE IS ADDED TO WATER
- THE ACTIVE SANITIZER
- COMBINATION OF HYPOCHLOROUS ACID AND HYPOCHLORITE IONS AND CHLORINE ASSOCIATED WITH CYANURIC ACID



# STABILIZER

THERE ARE 6 TYPES OF COMMONLY USED CHLORINE, DIVIDED INTO 2 CATEGORIES:

- **UNSTABILIZED (INORGANIC):** THESE COMPOUNDS DO NOT CONTAIN CARBON AND ARE VERY SENSITIVE TO UV LIGHT. THIS MEANS THAT WHEN THE SUN COMES OUT, THEY WILL DISSIPATE (GO AWAY.)
- **STABILIZED (ORGANIC):** THESE COMPOUNDS DO CONTAIN CARBON USUALLY IN COMBINATION WITH ELEMENTS SUCH AS: HYDROGEN, OXYGEN, NITROGEN, AND SULFUR, AND ARE NOT SENSITIVE TO UV LIGHT.

UNSTABILIZED OR INORGANIC CHLORINES ARE SENSITIVE TO THE SUN'S UV RAYS. USE OF THESE IS VERY IMPRACTICAL FOR AN OUTDOOR FACILITY. THESE FACILITIES NEED TO USE AN ADDITIONAL CHEMICAL KNOWN AS A **STABILIZER**, TO PREVENT THE UV RAYS FROM DECOMPOSING (EATING UP) THE CHLORINE RESIDUAL (OR CHLORINE LEVEL).







## CYANURIC ACID (CYA):

- **CYANURIC ACID (CYA):** the chemical used to stabilize an outdoor pool. CYA is added directly through a skimmer. CYA is added directly through a skimmer after being thoroughly diluted in a bucket of water or kneaded through old hosiery (note: this is one of very few chemicals that should ever be added through a skimmer. Always add chemicals directly to the pool unless otherwise directed.)
- For your initial addition, add cya slowly over the course of several days. Depending on the size of the pool, it may take quite a bit of CYA to get the pool “stabilized” in the beginning of the season.
- **{Dosage for cyanuric acid is approximately 3 lb. Per 10,000 gallons to increase 30 ppm}.**
- Put the cyanuric acid into the skimmers and add only a cup or so per skimmer per day until the desired level is reached. Be sure the pool has been recently backwashed (see ch. 9 for info. On backwashing) BEFORE adding the CYA. Most jurisdictions do not require daily testing because the level does not change that quickly. CYA will need to be added periodically throughout the summer. Wait 24-48 hours after each addition before adding more. This will assure adequate circulation and re-testing to avoid overdosing.



# CYANURIC ACID (CYA):

**Health Department regulation for Cyanuric Acid (CYA) is 30-50 in outdoor pools and it must be tested every week for a pool.**

**CYA is not used indoors.**



When the CYA value is tested and found to be too low, begin adding the Cyanuric Acid as described previously. Most jurisdictions are strict on the CYA levels. In some jurisdictions, the Chlorine level is based on the CYA level and other factors will at times play into what the Health Department considers an acceptable level.



If the CYA level gets too high especially if it goes above 100 ppm, the pool must be partially or totally drained in order to lower the level. There is no chemical to lower the CYA level.

**DRAIN & FILL TO LOWER!**



The pool operator should shoot for a reading of 50 ppm to maximize chlorine efficiency. As CYA levels go past 50 reduced stabilization occurs, If the CYA level is above 100 ppm, no additional stabilization is obtained, and reports have shown reduced disinfection and oxidation at elevated levels sometimes referred to as chlorine lock.



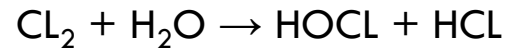
# SIX TYPES OF CHLORINE

6

- GAS CHLORINE
- SODIUM HYPOCHLORITE
- CALCIUM HYPOCHLORITE
- LITHIUM HYPOCHLORITE
- TRICHLOROCYANURIC
- DICHLOROCYANURIC



# GAS CHLORINE

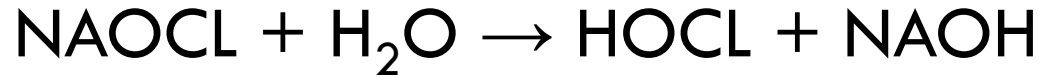


- Gas is an excellent sanitizer, however, due to its toxicity and for this reason most jurisdictions do not allow its use at public pools. The gas is sold in steel tanks that are kept in a room separate from the rest of the circulation system. **This room must have ventilation at floor level** and be able to exchange the air within 1-4 minutes due to the fact that the gas is heavier than air and will sink in the event of a leak.
- **A bottle of household ammonia is also required** within the confines of the gas tank room. When ammonia comes in contact with the gas in the event of a leak, a white mist is formed identifying the leak. The tank that is presently being used, will sit on a scale, and is weighed daily. This is how the CPO will know when the tank is almost empty. The other tanks, not currently in use, **must be chained to the wall to prevent tipping over.**
- **A self-contained breathing apparatus is also required.** A person, preferably the CPO, specially trained in the proper use of the self-contained breathing apparatus must be on site at all times. Keep the self-contained breathing apparatus in operational order and OUT of the chlorine room.
- It is important not to throw water on a chlorine gas leak. If water comes in contact with a chlorine gas leak, hypochlorous acid is formed which will increase corrosion and leakage. Careful handling of the equipment will help ensure safety and prevent leaks.
- **Gas chlorine has an extremely low ph.** Straight hydrochloric acid is formed when it mixes with water. Therefore, sodium carbonate (soda ash) is needed on a daily basis to keep the ph in the proper range.
- *{It takes approximately 1.3 oz. Of gas per 10,000 gallons of water to raise the free chlorine level 1 ppm.}*



# SODIUM HYPOCHLORITE

NaOCl

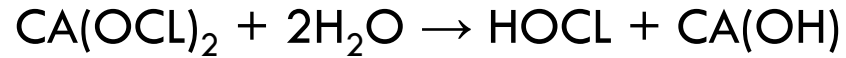


- **Sodium hypochlorite is one of the most widely used chlorines in the industry.** This is most likely due to the ease of use and the low cost. Sodium is a thick, yellowish liquid. It is usually stored in large vats and is delivered by a tanker truck. Always keep dry chlorine on hand, in case the liquid runs dry.
- Sodium hypochlorite, “bleach,” comes in at an extremely high ph, above 13, therefore the addition of muriatic acid is necessary on a daily basis to maintain ph in the proper range. **Never mix muriatic acid and chlorine directly together as it makes mustard gas** (a fatal gas used as a chemical weapon used in world war 1).
- Sodium hypochlorite is very unstable and loses its composition quickly. The hotter it is, the quicker it decomposes. It needs to be stored in a cool dark place. To ensure the quality of the bleach. Liquid chlorine has a short shelf life, so only order a month supply at a time.
- In most systems, liquid chlorine is fed into the system via a chemical feeder.

***{It takes approximately 10 oz. Of sodium hypochlorite per 10,000 gallons of water to raise the free chlorine level 1 ppm.}***



# CALCIUM HYPOCHLORITE



- Calcium hypochlorite is either a dry granular or a tablet form. It can be dissolved in water to produce a liquid form, although this is not recommended for daily disinfection. Calcium hypochlorite is very hard to dissolve. This property is definitely a disadvantage, especially for vinyl liners and pools with fresh plaster or paint. Dissolving it in lukewarm water will help.
- The ph of calcium hypochlorite is around 11-12; therefore the addition of muriatic acid will be needed on a daily basis to keep the ph in the proper range.
- For many years, granular calcium hypochlorite was not very practical to use as the daily disinfectant due to its insolubility. However, with the invention of erosion feeders and the introduction of calcium hypochlorite in a tablet or puck form, it is practical. The tablets are fed through an erosion feeder, which is plumbed into the circulation system. The tablets dissolve easily and with its high chlorine content, it helps to keep the pool sparkling clear. Be sure to use only compatible tablets or pucks in your erosion feeders.



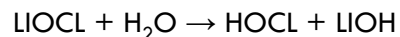
**Erosion feeders** are enclosed apparatuses used to dispense disinfectants. Water flows through the feeder eroding the solids inside and sending the disinfected water into the system.

- A possible disadvantage of calcium hypochlorite is that over time, it will raise the calcium hardness level in the pool. In areas with “soft” water this is usually not a problem. If the fill water is “hard,” this can be a disadvantage. Regular testing will allow the CPO to monitor the calcium hardness level.
- Calcium hypochlorite is an excellent choice to use for cleaning up bodily fluid spills. See the local health department regulations for specific procedures in handling these spills. It is an excellent oxidizer. **Oxidizers**, like chlorine and bromine, rid the contaminants and micro-organisms out of the pool water. Their chemical action literally burns the bad stuff off sending the gases into the atmosphere.
- *{It takes approximately 2 oz. Of calcium hypochlorite per 10,000 gallons of water to raise the free chlorine level 1 ppm.}*



# LITHIUM HYPOCHLORITE

LiOCl



- Lithium hypochlorite is also a granular. The main advantage of using lithium over calcium is that it is very soluble. The ACC is much less than calcium, but it is better for vinyl liners and freshly painted or plastered pools because of its solubility. Pound for pound, it is more expensive than calcium. This disinfectant is mainly used by backyard pool owners who are willing to pay for the speed that lithium hypochlorite dissolves.
- The ph of lithium hypochlorite is about 11; therefore the addition of muriatic acid will be needed in order to keep the ph in the proper range.
- Lithium hypochlorite is also an excellent choice for cleaning up bodily fluid spills. See the local health department regulations for specific procedures in handling these spills. It is a good choice to use for “shocking” needs.

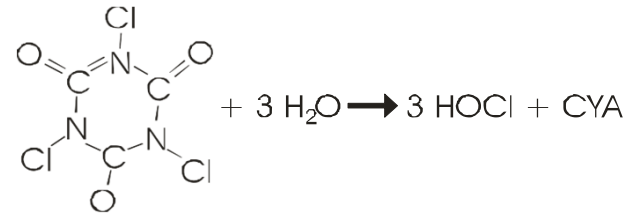
***{It takes approximately 3.8 oz. Of lithium hypochlorite per 10,000 gallons of water to raise the free chlorine level 1 ppm.}***



# TRICHLOR

Pre-Stabilized

- Tri-Chloro-S-Triazinetrione: (Trichloroisocyanuric Acid)



- Tri-chlor is a solid and is available in sizes ranging from 1" tabs to large sticks. It has a very low solubility rate and therefore is excellent for erosion feeders, floaters, and skimmer use. (Skimmer use is NOT allowed at a commercial facility in most jurisdictions). The size of the erosion feeder (based on the size of the pool), determines the size of the tablets. Note: be careful to reorder the appropriate size and always use the feeder exclusively according to manufacturers recommendations.
- Chemically, tri-chlor breaks down into the hypochlorous acid ion, and cyanuric acid. Some jurisdictions do not allow tri-chlor to be used on an indoor pool.
- Tri-chlor has a low ph (2.8-3.0), therefore the addition of sodium carbonate (soda ash) is needed on a daily basis to keep the ph in the proper range.
- ***{It takes approximately 1.5 oz. Of trichloro-s-triazetrione per 10,000 gallons of water to raise the free chlorine level 1 ppm.}***

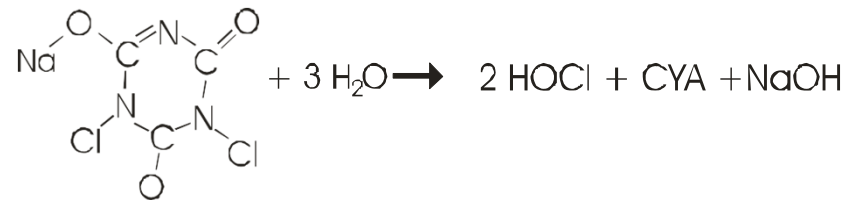




# DICHLOR

Pre-Stabilized

•Sodium Dichloro-S-Triazinetrione: (Sodium Dichloroisocyanuric Acid)



- Di-chlor is granular chlorine and it contains cyanuric acid. It is very soluble, and is usually used by dissolving it in water, adding a little soda ash, and feeding it into the system with chemical feeder like those used to feed sodium hypochlorite.
- The general mixing formula is as follows; 12 pounds of di-chlor, 1 ½ pounds soda ash, and 55 gallons of water
- Fill a 55- gallon vat ¾ full of water, mixing in the soda ash as it fills. All at once, pour in the di-chlor, put the lid on, and allow to sit for about 10 minutes. Stir until completely dissolved (using a wooden handle, not metal) and fill the vat the rest of the way.
- The ph of di-chlor in the dry form is about 4.0. By adding the soda ash and water, the ph rises to 7.5.



# SANITIZER EFFECT ON PH

- THE PH OF SANITIZER WILL EFFECT PH OF WATER
  - GAS CHLORINE <1.0
  - TRICHLOR 3.0
  - BROMINE TABLETS 4.5
  - BIGUANIDE 5.5
  - DICHLOR 6.5
  - CALCIUM HYPOCHLORITE 10.6
  - LITHIUM HYPOCHLORITE 11.7
  - SODIUM HYPOCHLORITE 12.8

*BASED ON A 1% SOLUTION*



# ACCEPTABLE FREE AVAILABLE CHLORINE LEVELS



Photo by David  
ortega

## THE MODEL AQUATIC HEALTH CODE RECOMMENDS:

- FOR SWIMMING POOLS NOT USING CYANURIC ACID TO STABILIZE :  
1.0 - 10.0 PPM
- FOR SWIMMING POOLS USING CYANURIC ACID TO STABILIZE :  
2.0 - 10.0 PPM
- FOR SPAS :  
4.0 - 10.0 PPM
- IDEAL RANGE FOR POOLS:  
2.0 – 4.0 PPM
- IDEAL RANGE FOR SPAS:  
3.0 – 5.0 PPM

Public swimming pools should check the main pool every 2 hours and spas & baby pools every hour





# SANITATION PH AFFECT ON CHLORINE

## CHLORINE'S KILLING POWER DEPENDENT ON PH

PH	%HOCL	%OCL-	
6.5	90.0	10.0	
7.0	72.5	27.5	
7.2	66.0	34.0	←
7.5	50.0	50.0	
7.8	32.0	68.0	←
8.0	21.5	78.5	

The lower the pH the more effective the chlorine



## COMBINED AVAILABLE CHLORINE (CAC)



Photo by andrea  
piacquadio

- FREE CHLORINE THAT REACTS WITH PERSPIRATION, URINE, AND OTHER NITROGEN CONTAINING CONTAMINANTS
- CHLORAMINES CAUSE UNPLEASANT ODOR AND EYE IRRITATION
- CHLORAMINES PROVIDE NO DISINFECTION

**CHLORINE + AMMONIA = CHLORAMINES**



TOTAL AVAILABLE CHLORINE IS  
THE SUM OF FAC AND CAC

FAC IS FOUND USING DPD 1 & 2  
TAC IS FOUND USING DPD 1, 2 & 3

CAC IS THE DIFFERENCE  
BETWEEN TAC & FAC  
(CAC = TAC - FAC)



EXAMPLE 1:

$$\text{FAC} = \text{DPD 1 \& 2} = 3.0$$

$$\text{TAC} = \text{DPD 1, 2, \& 3} = 4.0$$

$$\text{CAC} = \text{TAC} - \text{FAC} = 1.0$$

IF CAC IS GREATER THAN 0.4 THAN YOU  
HAVE TOO MUCH!







HOW DOES THE POOL OPERATOR  
ELIMINATE THE COMBINED CHLORINE?

## BREAKPOINT CHLORINATION

- METHOD OF REDUCING INORGANIC COMBINED CHLORINE (AMMONIA NITROGEN BASED)
- WILL NOT REDUCE MORE PREVALENT ORGANIC BASED CHLORAMINES
- BREAKPOINT IS ACHIEVED BY RAISING THE FREE AVAILABLE CHLORINE 10 TIMES THE COMBINED CHLORINE LEVEL



EXAMPLE 1:

$$\text{FAC} = \text{DPD 1 \& 2} = 3.0$$

$$\text{TAC} = \text{DPD 1, 2, \& 3} = 4.0$$

$$\text{CAC} = \text{TAC} - \text{FAC} = 1.0$$

$$\text{BREAKPOINT} = \text{FAC} + (10 \times \text{CAC})$$

$$\text{BREAKPOINT} = 3.0 + (10 \times 1)$$

$$\text{BREAKPOINT} = 13$$

FOR THIS EXAMPLE, THE POOL OPERATOR  
MUST RAISE THE CHLORINE LEVEL TO 13 PPM.





# WHAT ABOUT SUPERCHLORINATION OR SHOCKING THE POOLB

## SUPERCHLORINATION

- METHOD OF ACHIEVING BREAKPOINT BY RAISING THE FREE AVAILABLE CHLORINE 10 TO 15 PPM SO THAT BREAKPOINT IS SURPASSED AND THE REMAINING RESIDUAL IS LEFT FOR OPENING THE NEXT DAY.
- SHOCKING IS THE METHOD OF SUPERCHLORINATION BY ADDING ALL NECESSARY CHLORINE TO THE POOL AT THE END OF THE DAILY OPERATING HOURS.
- SUFFICIENT AMOUNT OF CHLORINATING COMPOUND (10 PPM – 30 PPM) CAN REDUCE CLOUDY WATER, SLIME FORMATION, MUSTY ODORS, ALGAE, BACTERIA, IMPROVE ABILITY TO MAINTAIN SANITIZER RESIDUAL, AND PROVIDE A RESPONSE TO A FECAL CONTAMINATION



EXAMPLE 2:

$$\text{FAC} = \text{DPD 1 \& 2} = 2.5$$

$$\text{TAC} = \text{DPD 1, 2, \& 3} = 3.0$$

$$\text{CAC} = \text{TAC} - \text{FAC} = 0.5$$

$$\text{BREAKPOINT} = \text{FAC} + (10 \times \text{CAC})$$

$$\text{BREAKPOINT} = 2.5 + (10 \times 0.5)$$

$$\text{BREAKPOINT} = 7.5$$

SUPERCHLORINATION OF 10 PPM WILL  
RESULT IN A FAC READING OF 5.0 PPM.



HOW MUCH CHLORINE DO YOU ADD TO THE POOL?

CALCIUM HYPOCHLORITE RAISES FAC 1.0 FOR EVERY 2 OZ PER 10,000 GALLONS

ADD 20 OZ FOR EVERY 10,000 GALLONS TO SUPERCHLORINATE





SO DOES THE POOL OPERATOR USE  
BREAKPOINT TO LOWER CHLORINE?

**NO!!!**

- **SODIUM THIOSULFATE** (DE-CHLOR) IS A CHLORINE NEUTRALIZER THAT WILL ELIMINATE FREE AVAILABLE CHLORINE FROM THE WATER.
- BREAKPOINT CHLORINATION WILL ONLY LOWER CHLORINE LEVELS IF THERE IS COMBINED CHLORINE PRESENT.
- SODIUM THIOSULFATE IS USED AT A RATE OF 1 OZ. PER 10,000 GALLONS TO REDUCE 1 PPM, AND ½ OZ. PER 10,000 GALLONS WILL REDUCE BROMINE 1 PPM.





## WHAT IS HYPERCHLORINATION

THE INTENTIONAL RAISING OF THE CHLORINE LEVEL TO INACTIVATE PATHOGENS DUE TO A FECAL CONTAMINATION OR OTHER CONTAMINATION.

- **LOWERING THE CYA TO 15 PPM OR LOWER**
- RAISING THE FAC TO 20 PPM OR HIGHER
- CLOSING THE FACILITY FOR AT LEAST 8 ½ HOURS (LENGTH OF CLOSURE DEPENDS ON AMOUNT OF FAC IN THE WATER: 20 PPM IS 28 HOURS; 30 PPM IS 18 HOURS; AND 40 PPM IS 8 ½ HOURS)
- REOPEN WHEN FAC IS BELOW MAXIMUM ACCEPTABLE LEVEL





## MAINTAINING CHLORINE LEVELS

- USE CHEMICAL FEEDERS
- USE AUTOMATED CONTROLLERS
- FREQUENT TESTING (AT LEAST EVERY TWO HOURS)
- MAINTAIN APPROPRIATE PH LEVEL





# BROMINE



**Water + Bromine**  $\longrightarrow$  **Hypobromous Acid + Hypobromite Ions**

- NOT AS EFFECTIVE AS CHLORINE EXCEPT IN VERY HOT WATER
- CAN NOT BE USED OUTDOORS BECAUSE IT CAN NOT BE STABILIZED
- CONTINUES TO DISINFECT AFTER IT COMBINES WITH AMMONIA
- EXCELLENT FOR USE IN INDOOR HOT TUBS AND THERAPY POOLS



# ACCEPTABLE FREE AVAILABLE BROMINE LEVELS



Photo by pixabay

## THE MODEL AQUATIC HEALTH CODE RECOMMENDS:

- FOR INDOOR SWIMMING POOLS :  
3.0 - 8.0 PPM
- FOR SPAS :  
4.0 - 8.0 PPM
- IDEAL RANGE FOR POOLS:  
3.0 – 5.0 PPM
- IDEAL RANGE FOR SPAS:  
4.0 – 6.0 PPM

Public swimming pools should check the main pool every 2 hours and spas & baby pools every hour





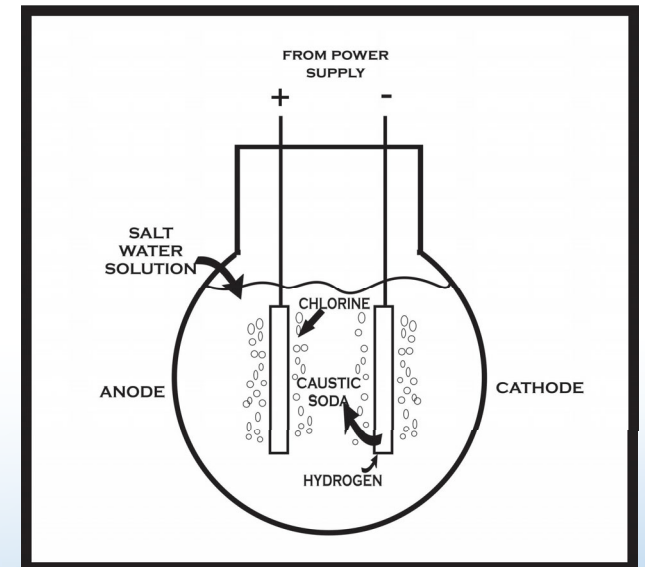
# ULTRAVIOLET LIGHT SYSTEMS

- ULTRAVIOLET LIGHT SYSTEMS ARE SECONDARY DISINFECTION SYSTEM
- POOL OPERATOR WILL STILL BE REQUIRED TO MAINTAIN AN ACCEPTABLE FREE AVAILABLE CHLORINE LEVEL
- ELIMINATES COMBINED CHLORINE THROUGHOUT THE DAY. BY ELIMINATING CHLORAMINES IT IMPROVES AIR QUALITY
- CHLORINE CONSUMPTION WILL INCREASE WITH THE USE OF UV SYSTEMS.



# CHLORINE GENERATION SALTWATER POOLS

- ELECTROLYSIS OF SALT (SODIUM CHLORIDE)  
(REMOVES CHLORINE FROM SALT TO CREATE ITS OWN DISINFECTANT)
- WHEN FAC USED UP, REVERTS BACK TO SALT TO BE REGENERATED AGAIN
- WHEN THE POOL OPERATOR REMOVES WATER, ADDITIONAL SALT WILL NEED TO BE ADDED
- TDS WILL BE HIGHER (USUALLY MAINTAINED AT 3,000-3,400 PPM). WHEN CALCULATING SATURATION INDEX THE POOL OPERATOR SHOULD USE -12.2 TO REPRESENT TDS IN THE CALCULATION.



# OTHER SANITIZERS

The following miscellaneous sanitizers are acceptable forms of disinfection in some jurisdictions. Check with the local Health Department regulations for prior approval BEFORE considering one of these sanitizers. Some of them may be required to be used combination with chlorine or bromine.

## OZONE

A BLUISH, IRRITATING GAS WITH A PUNGENT ODOR. IT KILLS BACTERIA AND OXIDIZES ORGANIC COMPOUNDS INCLUDING CHLORAMINES, SOAPS, OILS, AND WASTES. OZONE OXIDIZES RAPIDLY AND COMPLETELY. OZONE CONTINUALLY SHOCKS THE WATER

## IONIZATION

UTILIZES A CURRENT BETWEEN SILVER AND COPPER ELECTRODES AND RELEASES THESE

## BAQUACIL

THIS IS A HYDROGEN PEROXIDE BASED PRODUCT LINE. THIS IS ONE PRODUCT THAT MUST NOT BE USED WITH CHLORINE OR BROMINE. ONLY PRODUCTS THAT ARE PART OF THE BAQUACIL LINE CAN BE USED. A WARNING STICKER, STATING THAT BAQUACIL IS USED SHOULD BE PLACED ON THE FILTER.

## IODINE

THE REDUCED PH DEPENDENCE, INCREASED PATHOGENIC ORGANISM EFFECTIVENESS, AND RESIDUAL PERSISTENCE, GIVE IODINE SUPERIOR ADVANTAGES OVER CONVENTIONAL CHLORINATION OR BROMINATION OF SWIMMING POOLS

# LET'S REVIEW!

Fill in the chart with the range of each reading and the chemical name that raises and lowers each reading:

Chemical	Minimum	Maximum	Raise	Lower
Chlorine	1.0	10	Add chlorine	Add Sodium Thiosulfate
pH	7.2	7.8	Add Soda Ash	Add Muriatic Acid
Total Alkalinity	60	180	Add Baking Soda	Add Muriatic Acid
Calcium Hardness	150	400	Add Calcium Chloride	Drain And Fill
Cyanuric Acid	30	50	Add Cyanuric Acid	Drain And Fill
Total Dissolved Solids	No Minimum	1500	Not Applicable	Drain And Fill





**CHAPTER  
FIVE**

**TROUBLESHOOTING**





# CLOUDY WATER MOST COMMON WATER PROBLEM

THE ONLY WAY TO HAVE CLEAR WATER ALL THE TIME  
IS NEVER LET ANYONE SWIM IN IT!

## POSSIBLE CAUSES OF CLOUDY WATER:

- UNUSUALLY HIGH BATHER LOAD
- WATER NOT PROPERLY BALANCED
- LOW SANITIZER LEVEL
- PROBLEM WITH CIRCULATION SYSTEM
- PROBLEM WITH FILTRATION SYSTEM
- AIR IN YOUR SYSTEM

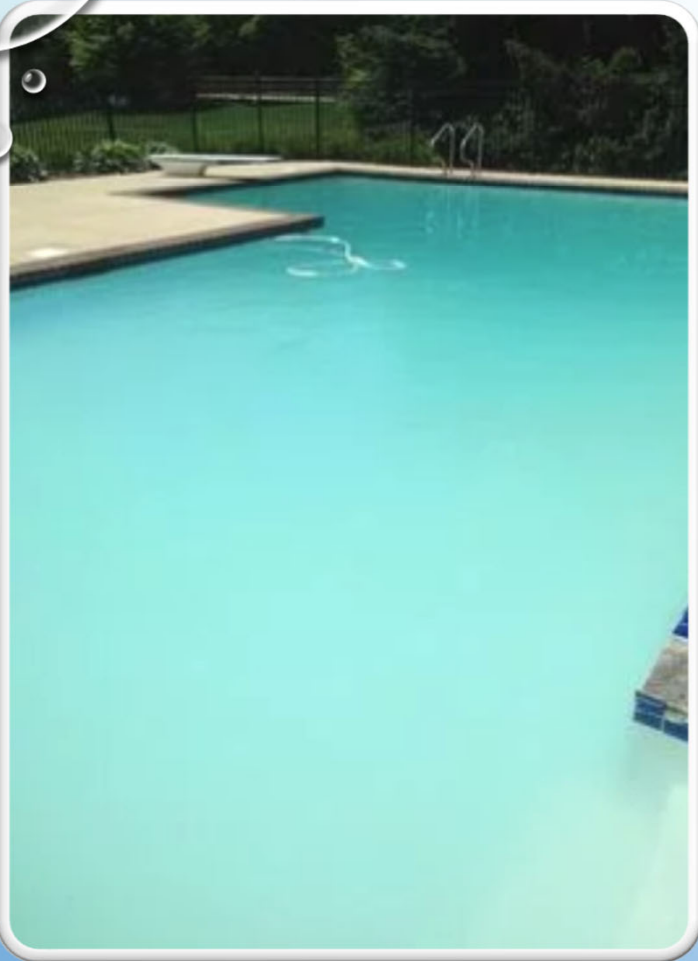




Photo by Sarah  
schorer

## CLEARING UP THE POOL

### UNUSUALLY HIGH BATHER LOAD

- THIS WILL USUALLY CORRECT ITSELF ON ITS OWN AFTER A FULL TURNOVER OF THE WATER.
- ALL POOLS HAVE AN ASSIGNED CAPACITY LEVEL. BE SURE TO ADHERE TO THE APPROPRIATE BATHER LOAD.



## CLEARING UP THE POOL

### WATER NOT PROPERLY BALANCED

- CHECK ALL OF THE CHEMICAL RANGES
- MAKE APPROPRIATE ADJUSTMENTS
- RECORD ALL ADJUSTMENTS MADE
- PROBLEM COULD TAKE A DAY OR SO TO RESOLVE



## CLEARING UP THE POOL

LOW SANITIZER LEVEL / HIGH COMBINED CHLORINE

- POOL OPERATOR SHOULD CHECK IF SUPERCHLORINATION IS NECESSARY.
- ADD CHLORINE IF LEVELS ARE LOW.



Photo by rob fox

## CLEARING UP THE POOL

### CIRCULATION OR FILTRATION PROBLEM

- IS PRESSURE TOO HIGH IN FILTER TANKS?
- IS FLOW RATE TOO LOW?
- DOES SYSTEM NEED TO BE BACKWASHED?

# CLEARING UP THE POOL

## AIR IN THE SYSTEM

- IS THE WATER LEVEL TOO LOW?
- IS A SKIMMER BLOCKED?
- CAN YOU SEE AN AIR BUBBLE ON THE TOP OF FILTER?
- STOP AIR FROM ENTERING SYSTEM
- RAISE WATER LEVEL IF NECESSARY
- USE AIR BLEEDER VALVE TO RELEASE AIR



# PROLONGED CLOUDINESS



CLEAR IN THE MORNING & CLOUDY IN AFTERNOON

- **USUALLY CAUSED BY SYSTEM BEING TOO SMALL FOR THE AMOUNT OF WATER OR BATHER LOAD**
- IF A REPAIR ON THE SYSTEM WAS RECENTLY MADE THIS IS PROBABLY THE ANSWER
- IF PUMP OR FILTER SIZE IS TOO SMALL, IT WILL BE ABLE TO KEEP UP WHEN THERE IS NO BATHERS BUT NOT DURING PEAK SWIMMING HOURS.

# SPECIALTY CHEMICALS

- WATER IS COMPLEX, AFFECTED BY MANY OUTSIDE FACTORS
- SPECIALTY CHEMICALS DESIGNED TO CORRECT ONE OF THOSE PROBLEMS
- MOST NEED TO BE USED IN ADDITION TO A SANITIZER







## MINERAL STAINING

Always check the fill water for calcium, iron, copper, manganese, etc.

To help determine if water has a high mineral content:  
Green = Copper    Brown = Iron    Black = Cobalt  
And Pink = Cobalt

## SEQUESTERING AGENT

Use a chelating or **sequestering agent** to increase the water's ability to hold metals in solution, instead of allowing them to precipitate out and stain the pool's surface. Sequestering agent prevent further staining, they do not remove stains!

Photo by  
culator.com



# ALGAE

The appearance of algae indicates there has been a period of time where the pool has not received enough chlorine or there is a part of the pool that is not getting enough chlorine.

## GREEN ALGAE

GREEN ALGAE ARE USUALLY FREE-FLOATING. (REFERENCE 1, PG. 44) WHEN IT ATTACHES TO THE WALLS, IT IS EASILY BRUSHED AWAY, LEAVING THE OPERATOR TO BELIEVE THE PROBLEM HAS BEEN SOLVED. SIMPLY BRUSHING IT IS NOT THE SOLUTION. WITHOUT PROPER TREATMENT, CERTAIN STRAINS OF GREEN ALGAE CAN CAUSE AN "ALGAE BLOOM". THIS MEANS THE POOL CAN LOOK LIKE SPLIT-PEA SOUP IN A MATTER OF HOURS. KEEPING THE DISINFECTANT LEVEL IN THE PROPER RANGE IS THE BEST MEANS OF PREVENTION. REGULAR BRUSHING OF THE WALLS IS ALSO VERY IMPORTANT IN ALGAE PREVENTION.

## MUSTARD ALGAE

MUSTARD (YELLOW) ALGAE ARE OFTEN FOUND ON THE SHADED SIDE OF THE POOL. (REFERENCE 1, PG. 45) IT IS A WALL CLINGING ALGAE THAT IS VERY DIFFICULT TO KILL. ACTUALLY A MEMBER OF THE GREEN ALGAE FAMILY, YELLOW ALGAE EXHIBITS CERTAIN CHARACTERISTICS THAT MAKE IT VERY DIFFICULT TO KILL.

## BLACK ALGAE

BLUE-GREEN (BLACK ALGAE) ARE ROUND SLICK SPOTS, USUALLY CONGREGATING IN AREAS OF POOR CIRCULATION. BLACK ALGAE HAVE A COVERING THAT MAKES IT EXTREMELY DIFFICULT TO KILL. THIS OUTER COVERING MUST BE REMOVED PRIOR TO TREATING THE ALGAE. IT IS REMOVED BY USING AN ALGAE BRUSH WITH STAINLESS STEEL BRISTLES, BRUSHING BACK AND FORTH MANY TIMES TO REMOVE THE ENTIRE COVER.

# PREVENTION OF ALGAE

- BRUSH THE POOL WALLS REGULARLY. (DAILY IF THE POOL HAS DEVELOPED ALGAE IN THE PAST)
- KEEP THE DISINFECTANT LEVELS IN THE PROPER RANGES AT ALL TIMES.
- KEEP SKIMMER BASKETS CLEAN AND SCRUB THE INSIDES OF THE SKIMMERS DAILY IN POOLS THAT HAVE DEVELOPED ALGAE IN THE PAST. SCRUB THE INSIDES OF THE SKIMMERS REGULARLY NO MATTER WHAT. UNDER THE WEIRS AND IN THE SKIMMER THROAT IS A HAVEN FOR ALGAE GROWTH.
- USE ALGISTATS IF THE POOL HAS A HISTORY OF ALGAE GROWTH, ESPECIALLY DURING A VERY HOT SUMMER WHEN THE WATER TEMPERATURE CAN BECOME VERY WARM.
- KEEP THE CYANURIC ACID (CYA) LEVEL BELOW 50 PPM.
- KEEP THE FILTER(S) CLEAN. BACKWASH AS NEEDED, BUT TOO OFTEN.
- IMMEDIATELY TREAT THE FIRST SIGN OF ALGAE.



# TREATING VISIBLE ALGAE

- UNLIGHT IS NEEDED FOR ALGAE TO GROW. TREAT IT DURING THE DAY WHEN THE SUN IS SHINING.
- BRUSH, BRUSH, AND BRUSH MORE!!! USE A STAINLESS-STEEL ALGAE BRUSH (UNLESS THE POOL SURFACE IS A VINYL LINER OR ANOTHER SURFACE WHERE THE MANUFACTURER DOES NOT RECOMMEND USING IT. A NYLON BRUSH WILL WORK JUST NOT AS WELL.
- METHODS OF “KILLING IT”:
  - SUPERCHLORINATE THE POOL TO 30 PPM. THIS IS GOING TO CAUSE DOWN TIME BUT IS ONE OF THE MORE EFFECTIVE METHODS. BE SURE ADDING THIS AMOUNT OF CHLORINE WILL NOT DAMAGE THE POOL SURFACE, AND REMEMBER TO ADD A SEQUESTERING AGENT BEFORE, IF HIGH MINERAL CONTENT IS A PROBLEM.
  - USE SOME TYPE OF A QUAT, POLYQUAT, OR COPPER BASED ALGAECIDAL TREATMENT. FOLLOW THE MANUFACTURER’S RECOMMENDATIONS EXPLICITLY. ADDING LESS THAN THE REQUIRED AMOUNT WILL NOT TAKE CARE OF PART OF THE PROBLEM. THE ENTIRE AMOUNT IS NEEDED FOR THE TREATMENT TO BE EFFECTIVE. THIS TYPE OF TREATMENT CAN BE EXTREMELY EXPENSIVE.
  - BLACK ALGAE CAN BE SPOT TREATED USING TABS OF TRI-CHLOR OR A SPECIALLY MADE ALGAE TREATMENT FOR SPOT TREATMENT. THIS USUALLY INVOLVES TURNING OFF THE SYSTEM AT NIGHT AND PLACING THE TABS DIRECTLY ON THE BLACK ALGAE SPOTS, THEN REMOVING THEM BEFORE OPENING. CONTINUE TREATING THE SPOTS FOR SEVERAL DAYS AFTER VISIBLE SIGNS OF ALGAE HAVE DISAPPEARED.
- IF IT IS POSSIBLE, USE A SEPARATE VACUUMING SYSTEM TO AVOID HAVING THE ALGAE RUN THROUGH THE FILTRATION SYSTEM. EVEN THOUGH, IT HAS BEEN TREATED, IT ONLY TAKES A LIVE SPORE TO ATTACH WITHIN THE SYSTEM AND BEGIN TO GROW AGAIN.
- AFTER CLEANING UP THE ALGAE, BE SURE TO THOROUGHLY CLEAN ALL EQUIPMENT THAT WAS USE IN THE PROCESS. DISINFECT ALL BRUSHES, VACUUM HEADS, POLES, ETC.



# LET'S REVIEW!

Fill in the chart with the range of each reading and the chemical name that raises and lowers each reading:

Chemical	Minimum	Maximum	Raise	Lower
Chlorine	1.0	10	Add chlorine	Add Sodium Thiosulfate
pH	7.2	7.8	Add Soda Ash	Add Muriatic Acid
Total Alkalinity	60	180	Add Baking Soda	Add Muriatic Acid
Calcium Hardness	150	400	Add Calcium Chloride	Drain And Fill
Cyanuric Acid	30	50	Add Cyanuric Acid	Drain And Fill
Total Dissolved Solids	No Minimum	1500	Not Applicable	Drain And Fill





**CHAPTER  
SIX**

**WATER  
TESTING**



# WATER TESTING



1. READ THE TEST KIT DIRECTIONS THOROUGHLY.
2. TAKE THE WATER SAMPLE WELL BELOW THE SURFACE OF THE WATER.
3. RINSE THE COMPARATOR/CELLS BEFORE BEGINNING THE TEST. PUT THE COMPARATOR INTO THE WATER UPSIDE DOWN AND THEN TURN RIGHT SIDE UP SEVERAL INCHES BELOW THE WATER'S SURFACE. THE COMPARATOR IS THE DEVICE THAT THE WATER READINGS ARE TESTED IN. THERE ARE MANY DIFFERENT TYPES OF COMPARATORS; SOME TEST KITS USE INDIVIDUAL TESTING CELLS. REFER TO INDIVIDUAL TEST KIT DIRECTIONS.
4. CHECK WITH THE LOCAL HEALTH DEPARTMENT AS TO THE FREQUENCY OF TESTING AND THE PROPER RANGES.
5. KEEP ALL TEST KITS AND REAGENTS OUT OF SUNLIGHT AND OUT OF THE REACH OF CHILDREN.
6. REPLACE REAGENTS ACCORDING TO MANUFACTURER RECOMMENDATIONS. USUALLY LIQUID REAGENTS SHOULD BE REPLACED ANNUALLY AND TABLETS AT LEAST EVERY 3 YEARS.
7. COMPARE ALL TESTS USING THE SAME LIGHT SOURCE.
8. ALTERNATE TESTING SITES TO ASSURE ALL "DEAD SPOTS" ARE ADEQUATELY TREATED. (DEAD SPOTS ARE AREAS OF POOR CIRCULATION).
9. NEUTRALIZE EXTREMELY HIGH CHLORINE/BROMINE LEVELS (ABOVE 10 PPM) BEFORE TESTING THE PH. USE A DROP OF SODIUM THIOSULFATE (FOUND IN THE TEST KIT), BEFORE USING THE PHENOL RED.
10. ALWAYS DOUBLE-CHECK A CHLORINE/BROMINE READING IF IT APPEARS TO BE "0." IT CAN "BLEACH OUT." **BLEACHING OUT** CAN HAPPEN IF A CHLORINE OR BROMINE READING IS EXTREMELY HIGH (ABOVE 10 PPM)







# WATER TEST KITS

- DISINFECTANT DETERMINATION
- DPD (N,N-DIMETHYL-P-PHENYLONE-DIAMINE), LIQUID OR TABLET FORM, IS THE MOST COMMON CHLORINE/BROMINE TESTING REAGENT. THIS INDICATOR TURNS THE SAMPLE PINK IN THE PRESENCE OF DISINFECTANT RESIDUAL. THE PINKER THE SAMPLE, THE HIGHER THE CHLORINE/BROMINE READING.
- TABLETS:
  - DPD #1: TESTS FREE AVAILABLE CHLORINE AND TOTAL AVAILABLE BROMINE
  - DPD #3: TESTS TOTAL AVAILABLE CHLORINE (THERE IS NO #2 TABLET)
- LIQUID:
  - DPD #1 AND #2: TESTS FREE AVAILABLE CHLORINE AND TOTAL AVAILABLE BROMINE
  - DPD #3: TESTS TOTAL AVAILABLE CHORINE
- POWDER
  - FILL COMPARATOR TUBE TO 25 ML
  - ADD TWO LEVEL SCOOPS TO SAMPLE
  - COUNT DROPS OF TITRATING REAGENT UNTIL SAMPLE TURNS FROM PINK TO CLEAR. MULTIPLY NUMBER OF DROPS BY .2





# WATER TESTING TEST KITS

## WHICH TEST KIT IS RIGHT FOR MY FACILITY?

- WHAT DO YOU NEED TO TEST FOR
- HOW OFTEN ARE YOU GOING TO USE IT
- HOW MANY PEOPLE ARE GOING TO USE IT
- HOW MUCH TIME DO YOU WANT TO SPEND TESTING

## ARE TEST STRIPS ACCEPTABLE?

- ONLY PROVIDE INDICATOR OF A RANGE
- NOT THE ACCURACY REQUIRED BY MOST HEALTH DEPARTMENTS



## WATER TESTING TESTING PROCEDURES

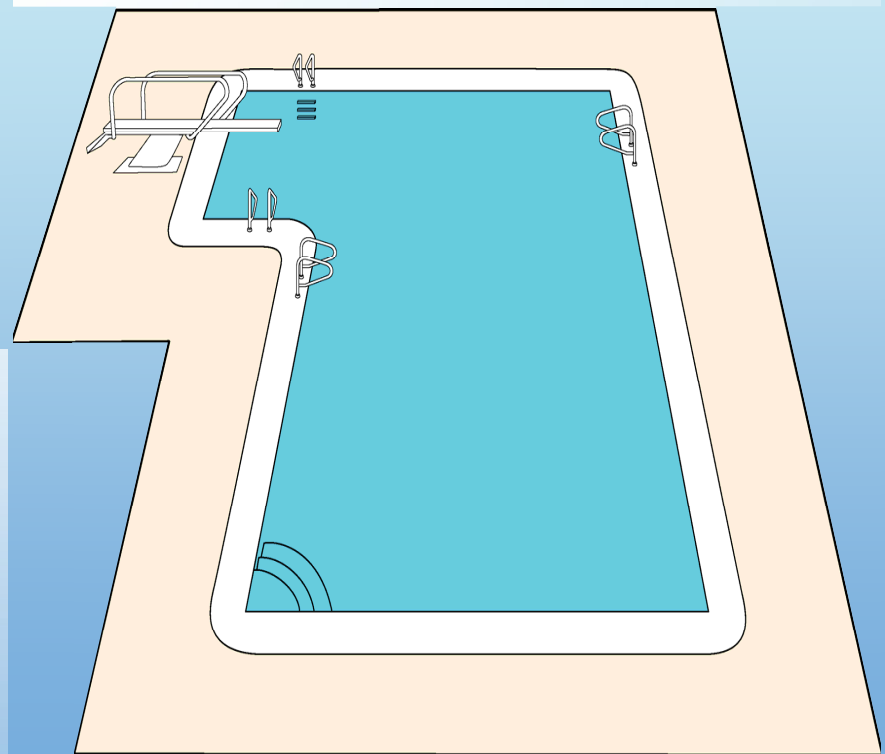
- RINSE THE TEST TUBE WITH TEST WATER (RINSE 3 TIMES WITH WATER TO BE TESTED)
- MEASURE CAREFULLY
- HOLD VERTICAL, CAP, AND DON'T USE FINGERS



# WATER TESTING TESTING PROCEDURES

## COLLECTING WATER SAMPLE

- TAKE SAMPLE FROM VARYING SPOTS AROUND THE POOL THROUGHOUT THE DAY
- BE SURE THAT SAMPLES REPRESENT ENTIRE BODY OF WATER
- TAKE APPROXIMATELY 12" - 18" BELOW WATER SURFACE
- DO NOT TAKE WATER SAMPLE IN FRONT OF INLET FITTING OR AT THE SAME LOCATION EACH TIME





# WATER TESTING TESTING PROCEDURES

- TEST SANITIZER RESIDUAL FIRST
  - IF TOO HIGH WILL AFFECT OTHER TESTS
- STORE TEST EQUIPMENT PROPERLY
- ANTICIPATE REPLACEMENT REAGENT REQUIREMENTS
- TESTS ARE ONLY AS ACCURATE AS PERSON DOING TEST
  - FEWER PEOPLE – MORE ACCURATE
  - PEOPLE VIEW COLORS DIFFERENTLY
    - DPD TITRANT SYSTEM DOES NOT REQUIRE COLOR MATCHING



# WATER TESTING SANITIZERS


- PROBLEMS WITH HIGH SANITIZER
  - BLEACHING OUT OF REAGENT
  - WHEN IN EXCESS OF 10 – 15 PPM PRODUCES FALSE LOW READING OR NO READING AT ALL
  - WORKAROUND
    - DILUTION
    - ADD MORE DPD
    - USE DPD-FAS





# TAYLOR TEST KIT DIRECTIONS

Reproduced from K-2006 instruction:

<p><b>Guide book (#2004B) amplifies these instructions and should be read to use this product properly.</b></p>		<p><b>POOL &amp; SPA WATER TESTS</b></p>		<p>3. Store test kit in cool, dark place. 4. Replace reagents once each year. 5. Do not dispose of solutions in pool or spa.</p>	<p>6. Rinse tubes before and after each test. 7. Obtain samples 18" (45 cm) below water surface. 8. Hold bottle vertically when dispensing.</p>	<p>Instr. #5138</p>
<p><b>Chlorine (Free, Combined) Test</b></p> <p>1. Rinse and fill large comparator tube to desired mark with water to be tested. NOTE: For 1 drop = 0.2 ppm, use 25 mL sample. For 1 drop = 0.5 ppm, use 10 mL sample. 2. Add 2 dippers R-0870. Swirl until dissolved. If free chlorine is present, sample will turn pink. NOTE: If pink color disappears, add R-0870 until color turns pink. 3. Add R-0871 dropwise, swirling and counting after each drop, until color changes from pink to colorless. 4. Multiply drops in Step 3 by drop equivalence (Step 1). Record as parts per million (ppm) free chlorine (FC). 5. Add 5 drops R-0003. Swirl to mix. If combined chlorine is present, sample will turn pink. 6. Add R-0871 dropwise, swirling and counting after each drop, until color changes from pink to colorless. 7. Multiply drops in Step 6 by drop equivalence (Step 1). Record as ppm combined chlorine (CC).</p>		<p><b>Total Alkalinity Test</b></p> <p>1. Rinse and fill large comparator tube to 25 mL mark with water to be tested.* 2. Add 2 drops R-0007. Swirl to mix. 3. Add 5 drops R-0008. Swirl to mix. Sample should turn green. 4. Add R-0009 dropwise. After each drop, count and swirl to mix until color changes from green to red. 5. Multiply drops in Step 4 by 10. Record as parts per million (ppm) total alkalinity as calcium carbonate. *When high TA is anticipated, this procedure may be used: Use 10 mL sample, 1 drop R-0007, 3 drops R-0008, and multiply drops in Step 4 by 25.</p>		<p><b>Calcium Hardness Test</b></p> <p>1. Rinse and fill large comparator tube to 25 mL mark with water to be tested.* 2. Add 20 drops R-0010. Swirl to mix. 3. Add 5 drops R-0011L. Swirl to mix. If calcium hardness is present, sample will turn red. 4. Add R-0012 dropwise. After each drop, count and swirl to mix until color changes from red to blue. 5. Multiply drops in Step 4 by 10. Record as parts per million (ppm) calcium hardness as calcium carbonate. *When high CH is anticipated, this procedure may be used: Use 10 mL sample, 10 drops R-0010, 3 drops R-0011L, and multiply drops in Step 4 by 25.</p>		
<p><b>pH Test</b></p> <p>1. Rinse and fill large comparator tube to 44 mL mark with water to be tested. 2. Add 5 drops R-0004. Cap and invert to mix. 3. Match color with color standard. Record as pH units and save sample if pH needs adjustment. If sample color is between two values, pH is average of the two. To LOWER pH: See acid demand test. To RAISE pH: See base demand test.</p> <p><b>Acid Demand Test</b></p> <p>1. Use treated sample from pH test. 2. Add R-0005 dropwise. After each drop, count, mix, and compare with color standards until desired pH is matched. See treatment tables to continue.</p> <p><b>Base Demand Test</b></p> <p>1. Use treated sample from pH test. 2. Add R-0006 dropwise. After each drop, count, mix, and compare with color standards until desired pH is matched. See treatment table to continue.</p>		<p><b>Cyanuric Acid Test</b></p> <p>1. Rinse and fill CYA dispensing bottle (#9191) to 7 mL mark with water to be tested. 2. Add R-0013 to 14 mL mark. Cap and mix for 30 seconds. 3. Slowly transfer cloudy solution to small comparator tube until black dot on bottom just disappears when viewed from top. 4. Read tube at liquid level on back of comparator block. Record reading as parts per million (ppm) cyanuric acid.</p>				

Powder/Titrating Reagent Testing Method Instructions



# TESTING DISINFECTANT LEVEL

## POWDER DIRECTIONS

### **Chlorine (Free, Combined) Test**

1. Rinse and fill large comparator tube to desired mark with water to be tested.

NOTE: For 1 drop = 0.2 ppm, use 25 mL sample. For 1 drop = 0.5 ppm, use 10 mL sample.

2. Add 2 dippers R-0870. Swirl until dissolved. If free chlorine is present, sample will turn pink.

NOTE: If pink color disappears, add R-0870 until color turns pink.

3. Add R-0871 dropwise, swirling and counting after each drop, until color changes from pink to colorless.

4. Multiply drops in Step 3 by drop equivalence (Step 1). Record as parts per million (ppm) free chlorine (FC).

5. Add 5 drops R-0003. Swirl to mix. If combined chlorine is present, sample will turn pink.

6. Add R-0871 dropwise, swirling and counting after each drop, until color changes from pink to colorless.

7. Multiply drops in Step 6 by drop equivalence (Step 1). Record as ppm combined chlorine (CC).

## LIQUID DIRECTIONS

### **Free, Combined & Total Chlorine (DPD)**

1. Fill small tube to 9 mL mark with sample water.
2. Add 5 drops R-0001 and 5 drops R-0002. Cap and invert to mix.
3. Match color.\* Record as ppm free chlorine ( $Cl_2$ ).
4. Add 5 drops R-0003. Cap and invert to mix.
5. Match color immediately. Record as ppm total chlorine ( $Cl_2$ ).
6. Subtract free chlorine (FC) from total chlorine (TC). Record as ppm combined chlorine (CC) as ( $Cl_2$ ).  
Formula:  $TC - FC = CC$ .

### **Total Bromine**

1. Fill small tube to 9 mL mark with sample water.
  2. Add 5 drops R-0001 and 5 drops R-0002. Cap and invert to mix.
  3. Match color.\* Record as ppm total bromine ( $Br_2$ ).
- \* If color is off-scale:** Repeat test using 4.5 mL sample diluted to 9 mL mark with tap water. Multiply reading by 2 to obtain approximate sanitizer level.
- If color is still off-scale:** Repeat test using 1.8 mL sample diluted to 9 mL mark with tap water. Multiply reading by 5 to obtain approximate sanitizer level.

### **Free & Combined Chlorine (FAS-DPD)**

1. Fill large tube to desired mark with sample water.  
NOTE: For 1 drop = 0.2 ppm, use 25 mL sample. For 1 drop = 0.5 ppm, use 10 mL sample.
2. Add 2 dippers R-0870. Swirl until dissolved. If free chlorine is present, sample will turn pink.  
NOTE: If pink color disappears or no pink color develops, add R-0870 until color turns pink.
3. Add R-0871 dropwise, swirling and counting after each drop, until color changes from pink to colorless.
4. Multiply drops in Step 3 by drop equivalence (Step 1). Record as ppm free chlorine ( $Cl_2$ ).
5. Add 5 drops R-0003. Swirl to mix. If combined chlorine is present, sample will turn pink.
6. Add R-0871 dropwise, swirling and counting after each drop, until color changes from pink to colorless.
7. Multiply drops in Step 6 by drop equivalence (Step 1). Record as ppm combined chlorine ( $Cl_2$ ).

OR





# TESTING PH

## ***pH Test***

1. Rinse and fill large comparator tube to 44 mL mark with water to be tested.
2. Add 5 drops R-0004. Cap and invert to mix.
3. Match color with color standard. Record as pH units and save sample if pH needs adjustment. If sample color is between two values, pH is average of the two. To LOWER pH: See acid demand test. To RAISE pH: See base demand test.

## ***Acid Demand Test***

1. Use treated sample from pH test.
2. Add R-0005 dropwise. After each drop, count, mix, and compare with color standards until desired pH is matched. See treatment tables to continue.

## ***Base Demand Test***

1. Use treated sample from pH test.
2. Add R-0006 dropwise. After each drop, count, mix, and compare with color standards until desired pH is matched. See treatment table to continue.



## TESTING TOTAL ALKALINITY

### ***Total Alkalinity Test***

1. Rinse and fill large comparator tube to 25 mL mark with water to be tested.\*
2. Add 2 drops R-0007. Swirl to mix.
3. Add 5 drops R-0008. Swirl to mix. Sample should turn green.
4. Add R-0009 dropwise. After each drop, count and swirl to mix until color changes from green to red.
5. Multiply drops in Step 4 by 10. Record as parts per million (ppm) total alkalinity as calcium carbonate.

***\* When high TA is anticipated,*** this procedure may be used: Use 10 mL sample, 1 drop R-0007, 3 drops R-0008, and multiply drops in Step 4 by 25.



# TESTING CALCIUM HARDNESS

## ***Calcium Hardness Test***

1. Rinse and fill large comparator tube to 25 mL mark with water to be tested.\*
2. Add 20 drops R-0010. Swirl to mix.
3. Add 5 drops R-0011L. Swirl to mix. If calcium hardness is present, sample will turn red.
4. Add R-0012 dropwise. After each drop, count and swirl to mix until color changes from red to blue.
5. Multiply drops in Step 4 by 10. Record as parts per million (ppm) calcium hardness as calcium carbonate.

\* ***When high CH is anticipated***, this procedure may be used: Use 10 mL sample, 10 drops R-0010, 3 drops R-0011L, and multiply drops in Step 4 by 25.



## TESTING CYANURIC ACID

### *Cyanuric Acid Test*

1. Rinse and fill CYA dispensing bottle (#9191) to 7 mL mark with water to be tested.
2. Add R-0013 to 14 mL mark. Cap and mix for 30 seconds.
3. Slowly transfer cloudy solution to small comparator tube until black dot on bottom just disappears when viewed from top.
4. Read tube at liquid level on back of comparator block. Record reading as parts per million (ppm) cyanuric acid.



## OTHER WATER TESTS

- TEMPERATURE
- METALS
- TDS
- HYDROGEN PEROXIDE
- SALT
- NITRATE AND PHOSPHATE
- TEST STRIPS

REMEMBER AT LEAST ONCE PER WEEK YOU WILL ALSO NEED TO CALCULATE THE SATURATION INDEX TO DETERMINE IF YOUR WATER IS IN BALANCE. USE THE WATERGRAM FOR THIS READING!







# LET'S REVIEW!

Fill in the chart with the range of each reading and the chemical name that raises and lowers each reading:

Chemical	Minimum	Maximum	Raise	Lower
Chlorine	1.0	10	Add chlorine	Add Sodium Thiosulfate
pH	7.2	7.8	Add Soda Ash	Add Muriatic Acid
Total Alkalinity	60	180	Add Baking Soda	Add Muriatic Acid
Calcium Hardness	150	400	Add Calcium Chloride	Drain And Fill
Cyanuric Acid	30	50	Add Cyanuric Acid	Drain And Fill
Total Dissolved Solids	No Minimum	1500	Not Applicable	Drain And Fill





**CHAPTER  
SEVEN**

**CHEMICAL  
AUTOMATION  
& CHEMICAL  
FEEDERS**





# CHEMICAL AUTOMATED TESTING

## WHAT YOU NEED TO KNOW ABOUT ORP:

### OXIDATION REDUCTION POTENTIAL READING

- ORP IS A MEASURE OF THE OXIDIZING CAPABILITY PRESENT IN THE WATER.
- SANITIZERS PRODUCE MEASURABLE ELECTRICAL OUTPUT THAT CAN BE MEASURED IN MILLIVOLTS.
- THE ORP DOES NOT TELL HOW MUCH CHEMICAL IS IN THE WATER, RATHER IT EXPRESS THE EFFECTIVENESS OF THE HYPOCHLOROUS ACID IN THE WATER.
- THE ORP DRINKING WATER STANDARD IS 650 MV MEETS.
- MOST POOLS AND HOT TUBS WILL BE KEPT AROUND 750 MV .
- THE ORP READING ON AN AUTOMATED TESTING SYSTEM MAY VARY FROM POOL TO POOL WHEN THE NUMBER IS COMPARED TO THE MANUAL TEST RESULT. THE IMPORTANT THING IS TO FIND THE CORRECT ORP READING THAT PRODUCES THE DESIRED MANUAL READING AND SET THE SYSTEM TO MAINTAIN THAT READING.



# CHEMICAL AUTOMATED TESTING

## WHAT YOU NEED TO KNOW ABOUT PH:

### PH READING

- PH IS MEASURED ON A 14 POINT SCALE AND WILL GIVE A NUMBER SIMILAR TO MANUAL READINGS.
- THE POOL OPERATOR WILL NEED TO FREQUENTLY CALIBRATE THE AUTOMATED TESTING MACHINE TO MATCH THE MANUAL READING.
- ALWAYS ASSUME THAT THE MANUAL READING IS CORRECT AND THAT THE AUTOMATED TESTING SYSTEM NEEDS TO BE ADJUSTED.
- THE AUTOMATED TESTING SYSTEM CAN BE SET TO MAINTAIN THE PH AND ACTIVATE THE CARBON DIOXIDE OR MURIATIC ACID FEEDER TO LOWER THE PH AS IT INCREASES.



# CHEMICAL AUTOMATED TESTING

## PROBLEMS WITH THESE SYSTEMS:

### PROBLEMS THAT EXIST

- MANY POOL OPERATOR BELIEVE THAT THE AUTOMATED TESTING SYSTEM IS HANDLING ALL OF THE READINGS...IT IS NOT! THESE SYSTEM MAINTAIN DISINFECTANT AND PH ONLY!
- AUTOMATED SYSTEMS THAT UTILIZE CARBON DIOXIDE TO MAINTAIN THE PH, OFTEN EXPERIENCE VERY HIGH TOTAL ALKALINITY BECAUSE OF THE HYDROXIDES PRODUCED. TOTAL ALKALINITY MUST BE MAINTAINED MANUALLY.
- OFTEN THE CHEMICAL FEEDERS THAT ARE ATTACHED TO THE AUTOMATED TESTING SYSTEM ARE TOO LARGE AND CHANGE THE POOL TOO QUICKLY. CHECK THE FEEDER SIZES.
- MONTHLY MAINTENANCE NEEDS TO BE MAINTAINED ON THE PROBES
- MANUAL TESTS CAN BE TAKEN FROM THE RESERVOIR AREA BUT THIS SHOULD NOT BE THE PRIMARY TESTING SPOT!



# LIQUID CHEMICAL FEEDERS

- ALL CHEMICALS FOR DISINFECTION MUST BE FED THROUGH MECHANICAL FEEDERS
- USES
  - DISPENSES DISINFECTANTS
  - DISPENSES ACIDS
  - SLURRY FEEDING DE
- WORKS BY ROLLERS APPLYING PRESSURE TO THE FEED TUBE WHICH DRAWS CHEMICAL UP FROM THE VAT AND PUSHING IT THROUGH THE TUBING INTO THE EFFLUENT SIDE OF THE CIRCULATION SYSTEM



# GAS FEEDERS

- TWO MAIN TYPES
  - GAS CHLORINE (RARELY USED)
  - CARBON DIOXIDE (USED IN COMBINATION WITH AN AUTOMATED TESTING SYSTEM)
- BASIC EQUIPMENT
  - INJECTOR TO PULL GAS OUT
  - CONTROL DEVICE
  - YOKE, OR CYLINDER UNIT
- CARBON DIOXIDE GAS IS NOT DANGEROUS
- GAS METERED IN STANDARD CUBIC FEET PER HOUR (SCFH)
- VAPOR CRYSTALS ON FEED EQUIPMENT DURING OPERATION – NORMAL
- ANNUALLY – INSPECT CHECK-VALVE AND TUBING





# SOLID CHEMICAL FEEDERS

## EROSION FEEDERS



- EROSION – GRADUAL WEARING AWAY OF A SOLID BY FRICTION
- CHEMICAL ERODED BY FLOW AND THE SOLUTION COMES OUT DISCHARGE AND THE AMOUNT IS CONTROLLED BY THE FLOW
- CALLED BY NAME OF CHEMICAL DISPENSED (I.E. CHLORINATOR, BROMINATOR)
- USE ONLY ONE DISINFECTANT IN FEEDER AND MAKE SURE THAT THE BRAND IS COMPATIBLE WITH THE FEEDER
- FLOWMETER MAY BE REQUIRED, CHECK CODES
- WHENEVER BACKWASHING CLOSE FEEDER REGULATING VALVES TO PREVENT UNNECESSARY EROSION AND BUILD UP OF PRESSURE IN THE FEEDER
- IN PRESSURE FEEDERS
  - DO NOT OPEN LID WITH PUMP RUNNING
  - BEFORE OPENING CHECK TO RELEASE PRESSURE

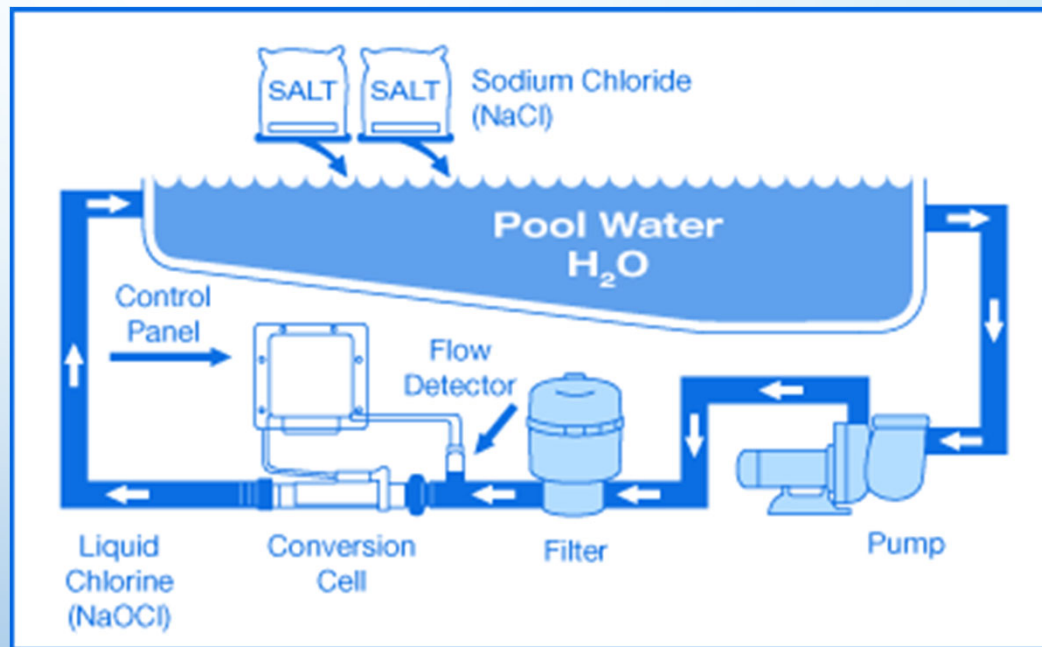


- 1. Vent valve**  
bleeds air to ease lid removal.
- 2. Safety lock-ring**  
spins off, freeing lid and providing easy access. Safety snap-lock provides secure seal.
- 3. Screen and check ball**  
are easy to reach for cleaning and periodic maintenance.
- 4. Flow-through circulation**  
allows maximum output of sanitized water.
- 5. External dial control**  
for easy flow rate adjustment.
- 6. Drain valve**  
makes it easy to drain feeder for safe recharging and winterizing.



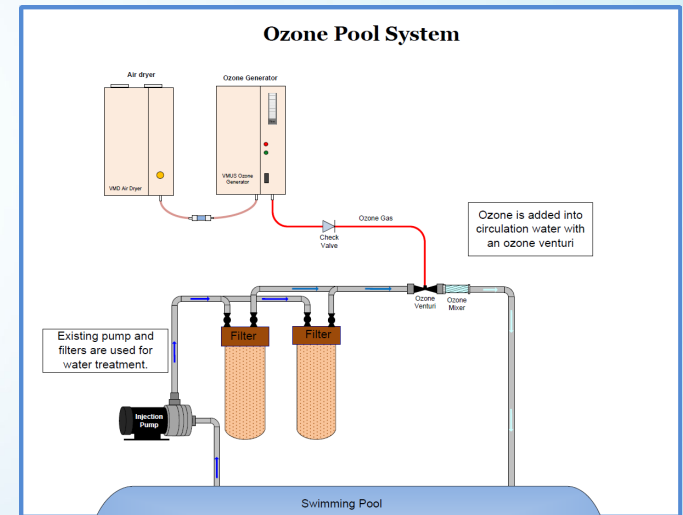
# CHLORINE/BROMINE GENERATORS

- THROUGH ELECTROLYSIS GENERATE ACTIVE SANITIZER FROM SALTS
  - SODIUM CHLORIDE
  - SODIUM BROMIDE
- TWO METHODS
  - BRINE TANK
  - IN-LINE



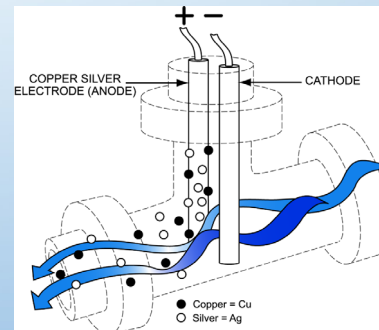
# OZONATORS

- CREATE OZONE THROUGH UV LAMPS AND ELECTRICAL VOLTAGE (CORONA DISCHARGE)
- OZONE MUST BE USED WITH EPA REGISTERED SANITIZER



# IONIZERS AND MINERAL SYSTEMS

- IONIZERS – ELECTROLYTIC DEVICES THAT DELIVER SILVER AND/OR COPPER IONS INTO WATER
- MUST BE USED IN COMBINATION WITH EPA REGISTERED SANITIZER



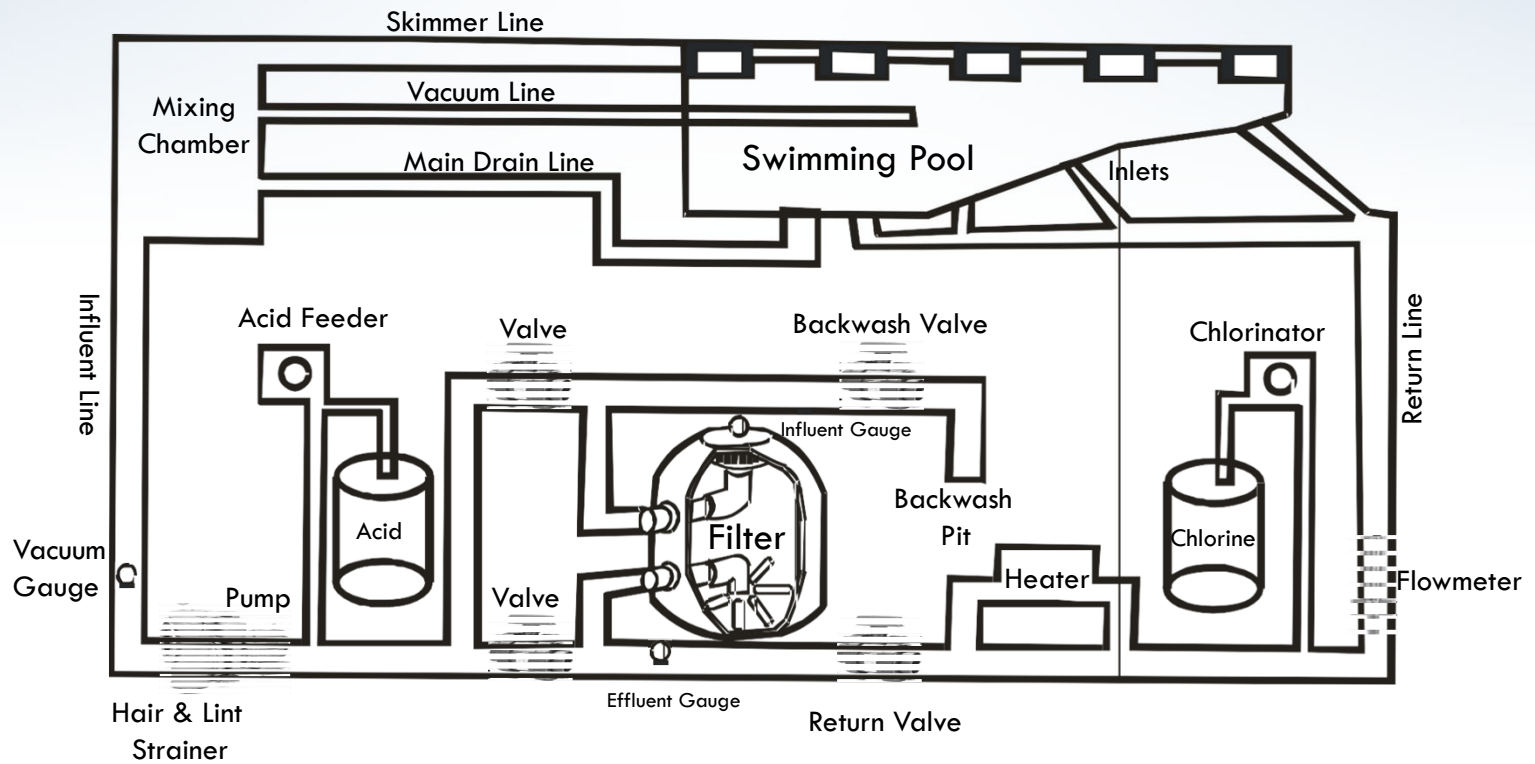


**CHAPTER  
EIGHT**

**WATER  
CIRCULATION**

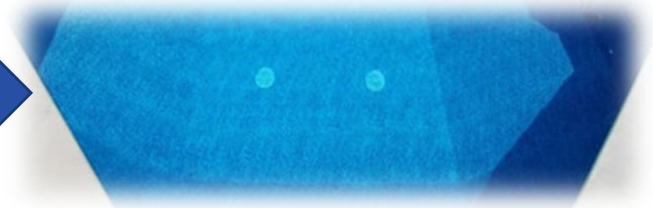


# WATER CIRCULATION



# INFLUENT LINES

MAIN DRAIN



SKIMMER



VACUUM



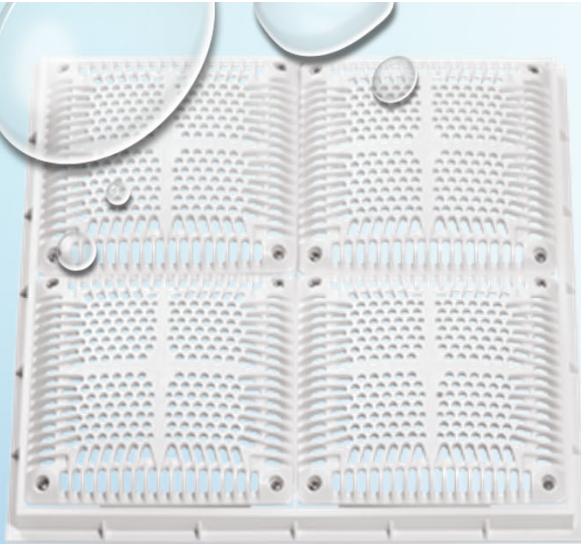
# MAIN DRAIN

- DEEPEST PART OF THE POOL
- ON THE BOTTOM OF THE POOL
- AT LEAST TWO DRAIN COVERS
- ANTI-VORTEX COVERS
- MUST BE SECURED AND NOT BROKEN TO BE OPEN



# DUAL MAIN DRAINS

- 2009 VIRGINIA GRAEME BAKER ACT
  - REQUIRE TO HAVE DUAL MAIN DRAINS THAT CANNOT BE ISOLATED AND ARE AT LEAST THREE FEET APART
- BY HAVING TWO DRAINS, AN INDIVIDUAL CANNOT CREATE A VACUUM SEAL BY THEMSELVES
- **ANTI-VORTEX DRAIN COVER IS LARGER AND ROUNDED INSTEAD OF FLUSH WITH THE POOL BOTTOM. THE WATER IS PULLED IN AND UNDER THE COVER**



Photos provided by Hayward

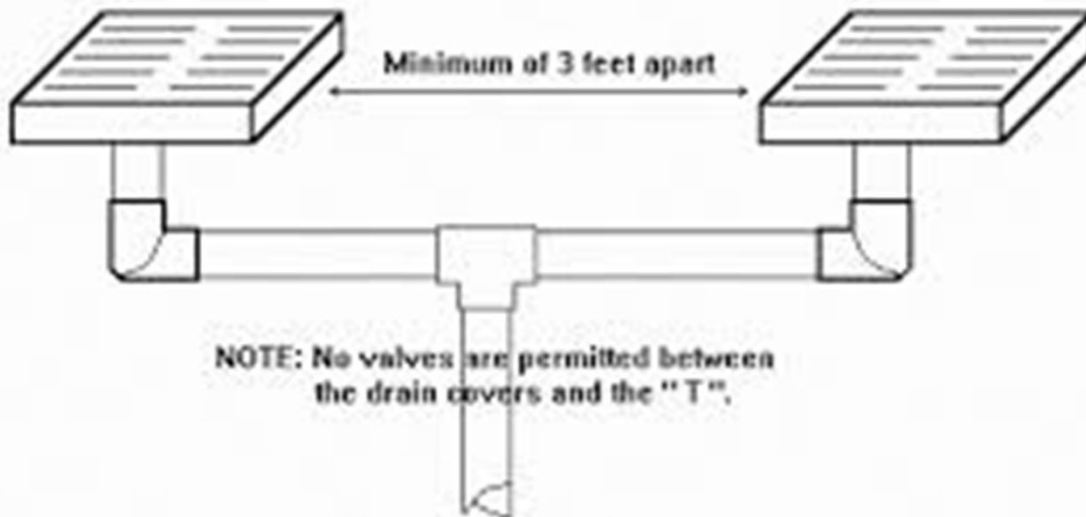


Figure 1. Dual Drain System



# SUCTION ENTRAPMENT GUIDELINES

- TO PREVENT HAIR ENTRAPMENT, LIMB ENTRAPMENT, MECHANICAL ENTRAPMENT, BODY ENTRAPMENT, & EVISCERATION
- SUCTION ENTRAPMENT PREVENTION MUST BE PART OF YOUR DAILY SAFETY INSPECTION

Daily Checklist
<input type="checkbox"/> Main drain, vacuum, inlet covers and/or fittings in place, secured and unbroken (hourly)
<input type="checkbox"/> Skimmers checked (baskets, weirs, lids & flow adjusters) for blockage (hourly)
<input type="checkbox"/> Warning/alert signs in place around the pool with emergency instructions and phone numbers
<input type="checkbox"/> On/Off switch to pump clearly labeled and location of pump clearly identified

(From U.S. Consumer Product Safety Commission, Guidelines for Entrapment Hazards: Making Pools and Spas Safer)

Items to be Checked in Filter Room and Pool Before Filling and After Periodic Maintenance and Cleaning Procedures
<input type="checkbox"/> Proper Suction drain covers installed and inspected for breakage (Main and wading pools)
<input type="checkbox"/> Suction drain covers firmly and properly affixed using manufacturer's recommended parts
<input type="checkbox"/> Suction drain covers firmly and properly affixed using manufacturer's recommended parts
<input type="checkbox"/> Proper return covers installed (Main and wading pools)
<input type="checkbox"/> Skimmers checked (baskets, weirs, lids & flow regulators) for blockage
<input type="checkbox"/> All skimmer throats checked for blockage (Main and wading pools)
<input type="checkbox"/> All valves and filter lines labeled and functional
<input type="checkbox"/> Vacuum covers or fittings in place (if applicable)
<input type="checkbox"/> Location of the On/Off switch to circulation pump clearly identified
<input type="checkbox"/> On/Off switch to circulation pump clearly labeled



# **DAILY ENTRAPMENT PREVENTION**

- Check Drain Covers Daily**
- Check Water Is Above Bottom of Skimmers**
- Empty Skimmer Baskets Daily**
- On/Off Switch is Clearly Marked**



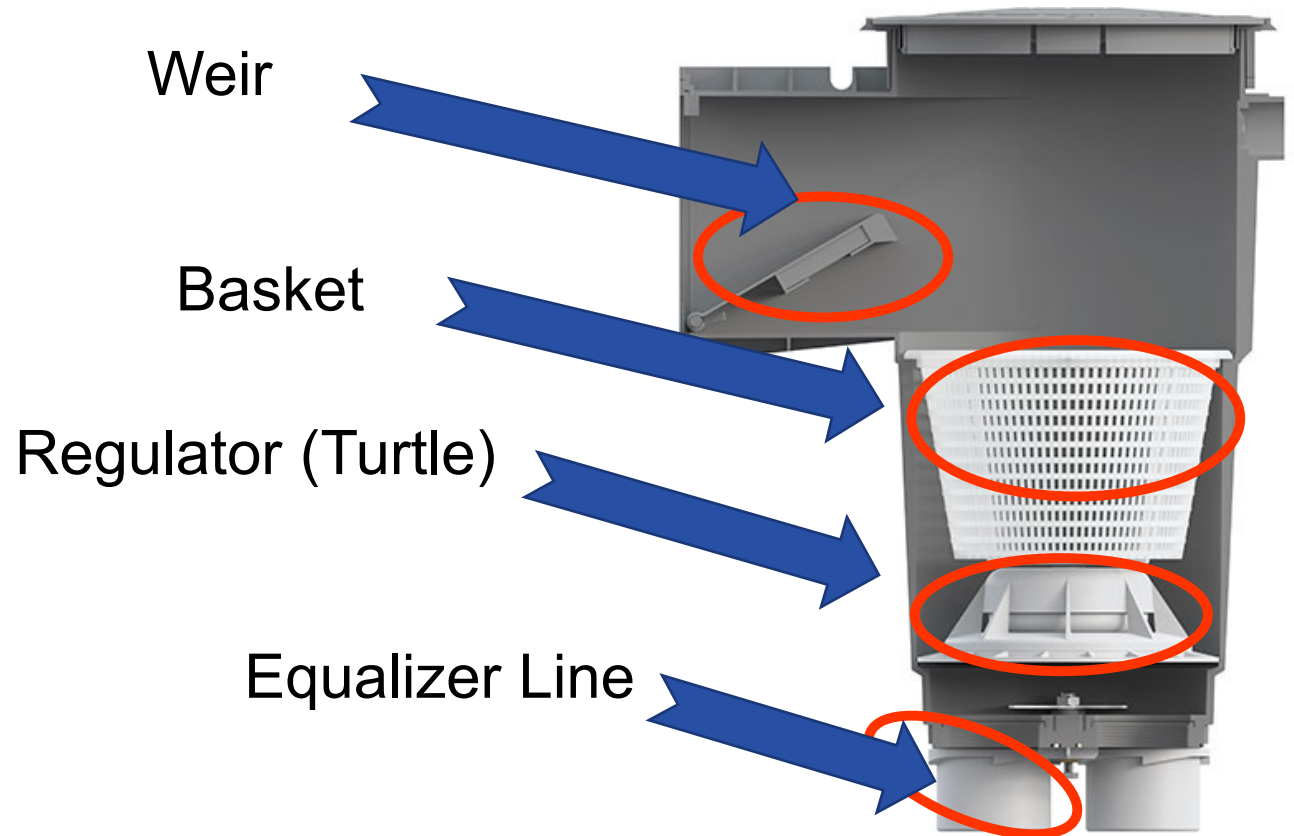


# HYDROSTATIC RELIEF VALVE

RELEASES UNDERGROUND  
WATER PRESSURE WHEN THE  
POOL IS EMPTY



# SKIMMERS



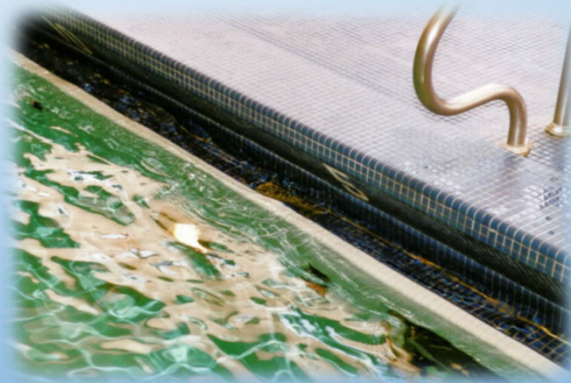
The skimmer line is responsible for removing debris off the surface of the water. Water level should be maintained at halfway to three quarters of the way up the hole in order to ensure that water is pulled into the system. At least 50% of the water entering the circulation system has to come through the skimmer line.

*Photos provided by Hayward*



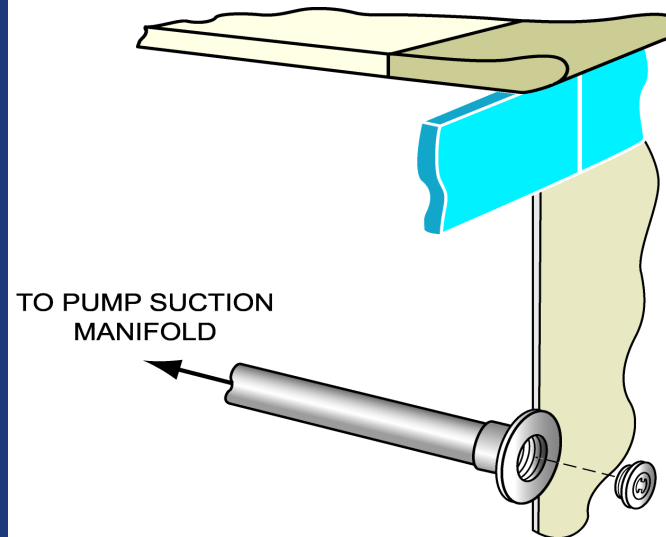
# GUTTERS

- ONE BIG SKIMMER AROUND THE PERIMETER OF THE POOL
- USE BALANCING (SURGE) TANK
- CONTINUOUS OVERFLOW INTO TROUGH IN ORDER TO REDUCE WAVES
- USES GRAVITY TO HAVE WATER FLOW INTO POOL PIPING SYSTEM



# VACUUM LINE

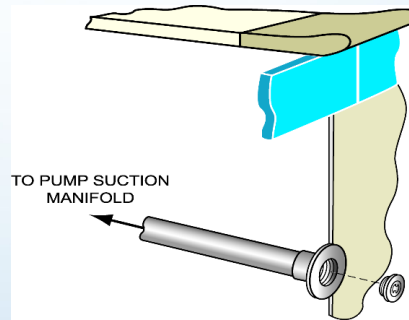
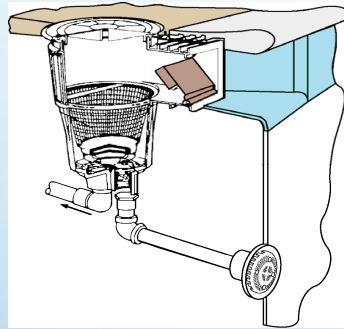
## VACUUM OUTLETS



- ONE METHOD OF CONNECTING SUCTION VACUUMING EQUIPMENT TO CIRCULATION SYSTEM
- SHUT VALVE, CAP LINE WHEN NOT IN USE



# THE CIRCULATION SYSTEM METHODS OF VACUUMING



# VACUUM PROCEDURES

- Attach the vacuum head to the pole, and the hose to the vacuum head.
- Place the vacuum head into the water and allow it to sink to the bottom.
- Prime the vacuum hose by filling it with water BEFORE hooking it up to the port.
- This is most easily achieved by feeding the hose straight down the wall into the water, so it fills with water.
- Plug the other end of the hose into the vacuum hole or bottom of the skimmer, whichever is used to vacuum

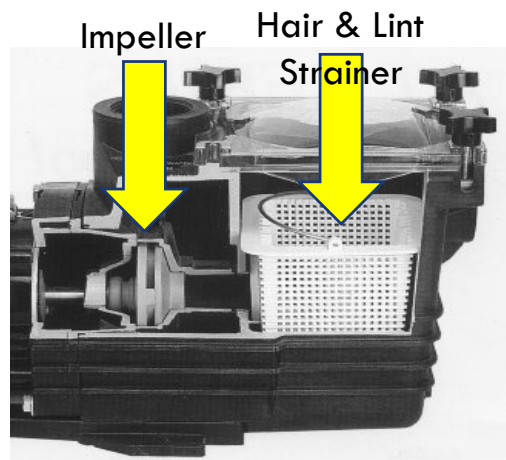


DO NOT PLUG THE VACUUM HOSE INTO THE PORT, UNLESS IT HAS WATER IN IT!



# HAIR & LINT STRAINER

- IS ALWAYS LOCATED BEFORE OR ABOVE THE PUMP. IT IS THE LAST DEFENSE FOR THE PUMP.
- PROTECTS THE IMPELLER WHICH IS WHAT IS MOVING THE WATER THROUGH THE SYSTEM
- PUMP CAVITATION OCCURS WHEN WATER IS NOT MAKING IT TO THE MOTOR
- CAVITATION IS USUALLY THE RESULT OF A FULL BASKET



Photos provided by Hayward



# CLEANING THE BASKET & PRIMING THE PUMP

**Cleaning the basket** should be done each time the system is backwashed

1. Turn of the pump
2. Close influent and effluent valves
3. Open Hair & Lint Basket area
4. Remove dirty basket
5. Replace with clean basket (facilities should have two basket)
6. Clean dirty basket after it has dried

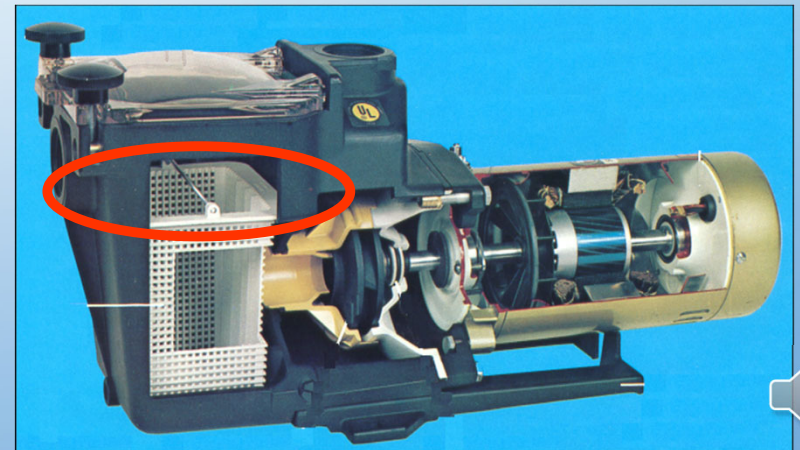
**Priming the Pump** is essential. The system must always have water running through it.

1. Fill basket area completely with water (Use buckets of water or a hose if needed).
2. Secure the lid on the hair and lint pot area. Any air on the suction side of the pump will prohibit a pump from “catching prime”.
3. Open effluent line valve
4. Turn on pump influent valves to begin flow as vacuum pressure begins to increase



## VACUUM GAUGE

- PRIOR TO PUMP
- INDICATES HOW HARD THE PUMP IS WORKING TO GET WATER FROM POOL/SPA



# PROBLEMS TO AVOID WHEN PRIMING THE PUMP

**Above ground pumps** are the most difficult with which to work. The header valves must be closed prior to turning off the pump in order to clean the strainer. If these valves are not closed, all the water will run back to the pool and the pump will need to be re-primed prior to starting the system..

**Below level systems** (flooded suction) are the easiest to work with. All header valves must be closed before cleaning the hair strainer to prevent the water from rushing from the pool and out of the hair strainer. These pumps are called “flooded suction” pumps and prime themselves with gravity

# PUMP/MOTOR

THE MOTOR MAKES THE PUMP WORK. THE IMPELLER IS ATTACHED TO THE SHAFT OF THE MOTOR, AND THE MOTOR SPINS THE IMPELLER, CIRCULATING THE WATER. THE NUMBER OF GALLONS PER MINUTE A PUMP WILL PRODUCE IS DEPENDENT ON THE POWER OF THE MOTOR.

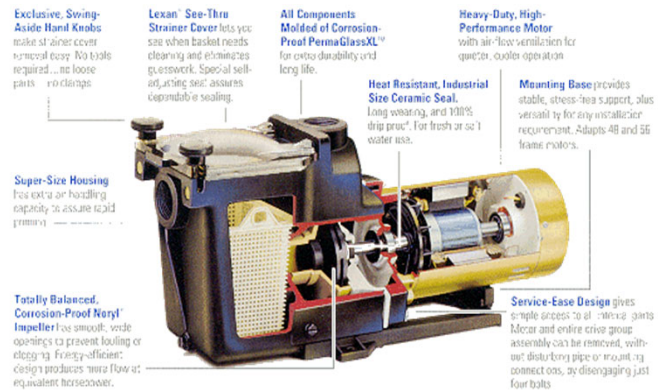


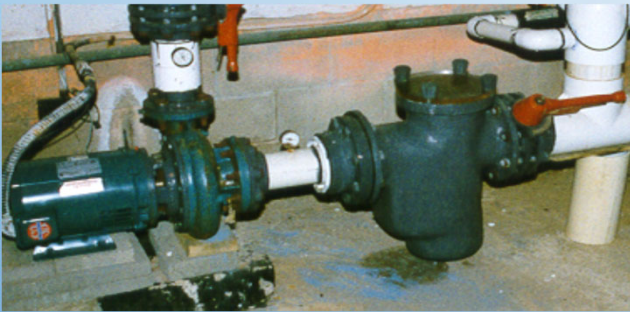
Photo courtesy of Hayward



# CIRCULATION PUMP



- SELF PRIMING (AT OR ABOVE GRADE)
  - DESIGNED TO BE INSTALLED AT POOL SURFACE LEVEL OR HIGHER
  - SELF PRIMING
    - AFTER INITIAL CHARGE
  - DRAWS AIR OUT OF SUCTION LINE OF SYSTEM



- FLOODED SUCTION (BELOW GRADE)
  - DESIGNED TO BE INSTALLED BELOW WATER SURFACE LEVEL
  - POOR AT VENTING AIR FROM SYSTEM
  - USES GRAVITY TO BEING SUCTION WATER TO PUMP



# PROBLEMS WITH THE CIRCULATION PUMP

IF SELF PRIMING PUMP WON'T PRIME, CAUSE MAY BE

- INSUFFICIENT LIQUID IN PUMP BODY
- AIR LEAK
- LEAK AT STRAINER COVER
- BLOCKED SUCTION LINE
- PLUGGED IMPELLER
- BAD MECHANICAL SEAL (PACKING GLAND)
- NOT ENOUGH WATER IN POOL
- SUCTION VALVE CLOSED
- DIRTY FILTER



# FILTER

Filters remove small particulate matter from the pool. Filters do not remove microorganisms.

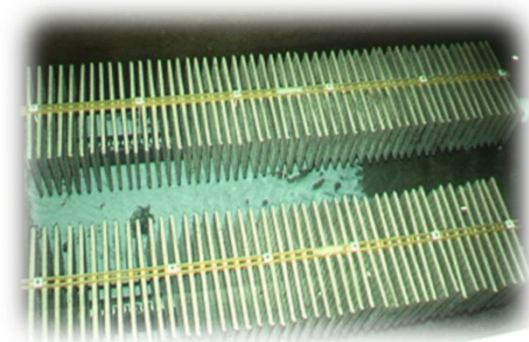
The water passes through the filter media (sand, DE, or cartridge) leaving behind any particulate matter that is trapped by the filter media.



Sand Filter



Cartridge Filter



Diatomaceous Earth Filter

Refer to Chapter 9 for more information on filters.

Photos courtesy of Hayward and \_\_\_\_\_



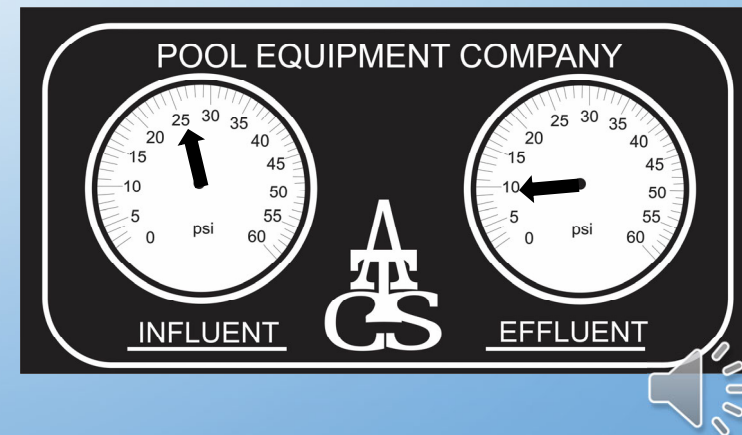
# INFLUENT PRESSURE GAUGE AND EFFLUENT PRESSURE GAUGE



- INFLUENT GAUGE LOCATED ON TOP OF THE FILTER, ON PIPING FROM PUMP TO FILTER, ON GAUGE PANEL
- INDICATES PRESSURE EXERTED BY PUMP TO PUSH WATER THROUGH FILTER



- EFFLUENT GAUGE LOCATED ON PIPING AFTER FILTER OR ON GAUGE PANEL
- INDICATES PRESSURE EXERTED BY PUMP TO PUSH WATER BACK TO THE POOL AFTER IT EXITS THE FILTER





# TURNOVER RATE

**Turnover rate** is the amount of time it takes all of the water in the pool to go through the filtration system and return to the pool basin.

## MODEL AQUATIC HEALTH CODE RECOMMENDS:

- 6 HOURS FOR MAIN POOLS
- 1 HOURS FOR BABY POOLS
- 30 MINUTES FOR HOT TUBS

## MARYLAND HEALTH CODE RECOMMENDS:

- 6 HOURS FOR MAIN POOLS
- 2 HOURS FOR BABY POOLS
- 30 MINUTES FOR HOT TUBS



## Turnover Rate

$$\text{Flow Rate} = \frac{\text{Volume}}{(\text{turnover rate} \times 60)}$$

Examples will be done in Chapter 9



# HEATER

**Heater:** If the pool has a Heater, it should be installed after the filter and before the chemical injection points. Most systems require the heater to be turned off when the circulation pump is turned off for any reason, even if it will be off for only a few minutes. Some heaters are to be turned off several minutes before turning off the rest of the system (see manufacturer's instructions). Only turn the heater back on after the main (re-circulating) pump has been turned back on.

**In order to size the heater, the operator  $BTUs = Gallons \times 8.33 \times \text{°F temperature rise}$**

## TYPES OF HEATERS AVAILABLE

- GAS/OIL FIRED
- ELECTRIC
- HEAT PUMP
- SOLAR
- HEAT EXCHANGER

*Photos courtesy of Hayward*



# CHLORINATOR

**Chemical Feeders** introduce the disinfectant chemicals and pH adjusters to the filtered water. Most systems have the chemical injection points AFTER the filters and heaters.



*Photos courtesy of Hayward and Blue & White Industries*

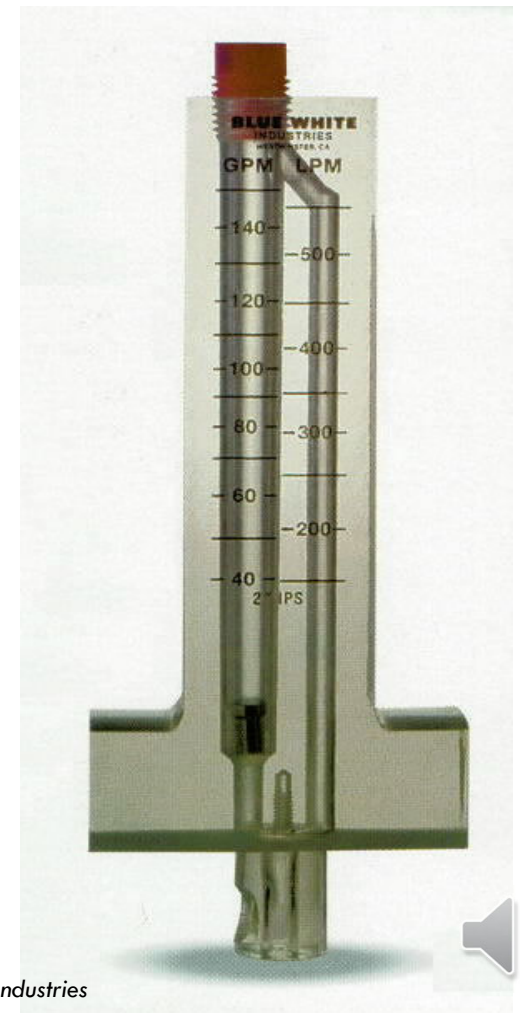


# FLOW METER

- INDICATES RATE OF FLOW IN GALLONS PER MINUTE
- MINIMUM FLOW RATE CALCULATED TO MEET TURNOVER RATE
- INSTALLED ON RETURN LINE AFTER HEATER, BEFORE CHEMICAL FEEDER
- REQUIRES TO BE INSTALLED STRAIGHT PIPE

IF FLOW RATE DROPS TO, OR BELOW THE MINIMUM FLOW RATE:

- CLEAN THE SKIMMER BASKETS
- CLEAN THE HAIR & LINT BASKET
- BACKWASH THE FILTER
- CHANGE THE FILTER MEDIA
- CHECK FOR OBSTRUCTION IN PIPING OR EQUIPMENT



Photos courtesy of Blue & White Industries

# THE CIRCULATION SYSTEM

## Turnover Rate

$$\text{Flow Rate} = \frac{\text{Capacity}}{(\text{Turnover Rate} \times 60)}$$

The flowrate indicates whether the turnover rate is being maintained

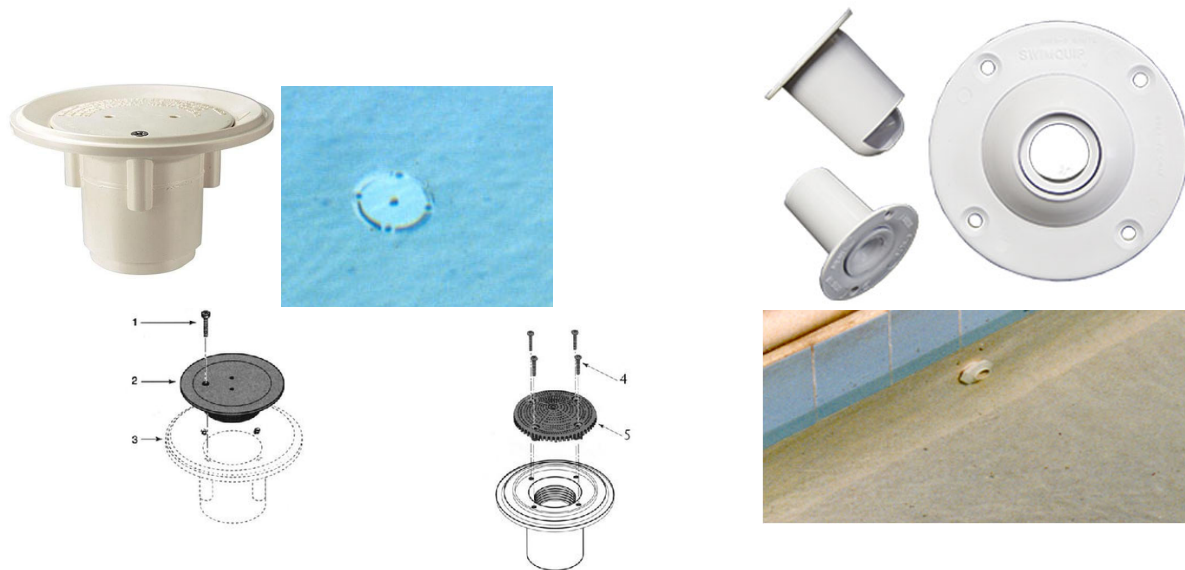


# RETURN LINE AND RETURN INLETS

**Return Line and Inlets (Returns):** The return line carries cleaned, heated, and chemically treated water back to the pool. It enters the pool through inlets or returns.

These inlets or returns are located on the sides or the bottom of the pool. On a gutter pool, the returns can be located around the bottom edge of the gutter.

Inlets are adjustable to ensure that each one has the same amount of pressure thus producing an equal distribution of disinfected water.



Photos courtesy of Pentair



# VALVES

There will be many valves and pipes in the pump room. Most commercial valves are open if the handle is running with the pipe and closed if the handle is running across the pipe.

Most jurisdictions require that all valves and pipes are colored and/or labeled for easy identification. A valve legend should be posted in the pump room, easily visible and clearly reflect the locations of all the pertinent valves and pipes and the color coding for each.



*Photos courtesy of Pentair*





# VALVES

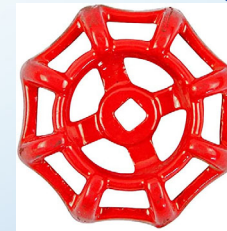


Valve handle is 90°, or perpendicular to pipe –  
Valve is closed

Valve handle in line with pipe –  
Valve is open



Photos provided by Hayward



Righty Tighty



Lefty Loosey



## Appendix C Sample Data Sheet

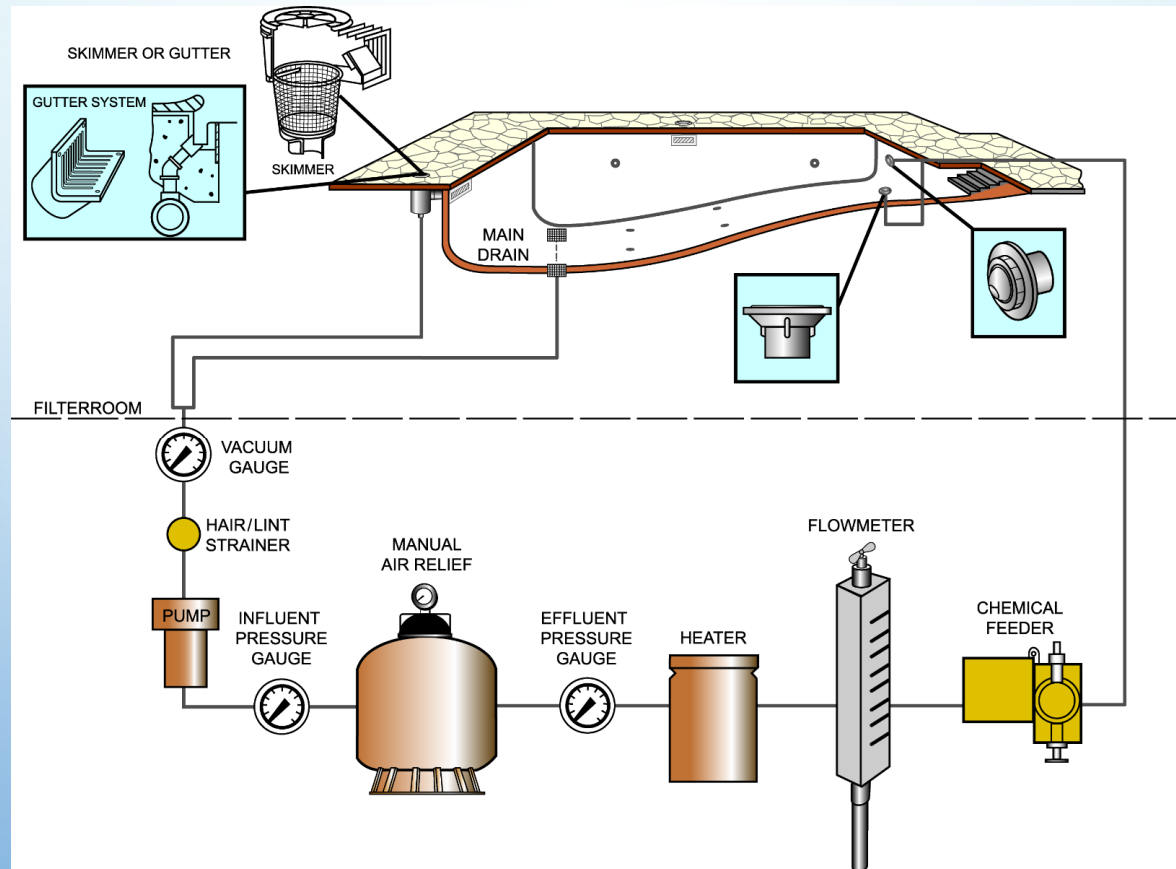
SWIMMING POOL DATA SHEET		
	Main Pool	Wading Pool
<b>Gallon Capacity</b>	83,330 Gallons	1,687 Gallons
<b>Surface Area</b>	2,580 Square Feet	256 Square Feet
<b>Flow Rate</b>	173 GPM	14 GPM
<b>Inlets</b>	12 Floor Returns	3 Floor Returns
<b>Sanitizer</b>	Sodium Hypochlorite	Sodium Hypochlorite

OPERATING INSTRUCTIONS		
	Main Pool	Wading Pool
<b>To Filter:</b>	Open valves labeled skimmer, main drain, filter, and return.  Close valves labeled vacuum, backwash, and waste.	Set multi-port valve to FILTER.
<b>To Backwash:</b>	Open valves labeled skimmer, main drain, backwash, and waste.  Close valves labeled vacuum, filter, and return.	Set multi-port valve to BACKWASH, and after backwashing, set multi-port valve to rinse for about 20 seconds.
<b>To Vacuum:</b>	Open valves labeled skimmer, vacuum, and return.  Close valves labeled main drain, backwash, and waste.	Set multi-port valve to FILTER. Vacuum through the skimmer. The other skimmers may need to be closed in order to get enough suction.
<b>Important Notes:</b>	<ol style="list-style-type: none"> <li>1. Always turn off pump before changing any valve settings</li> <li>2. Never run chlorinator while system is off or backwashing</li> </ol>	

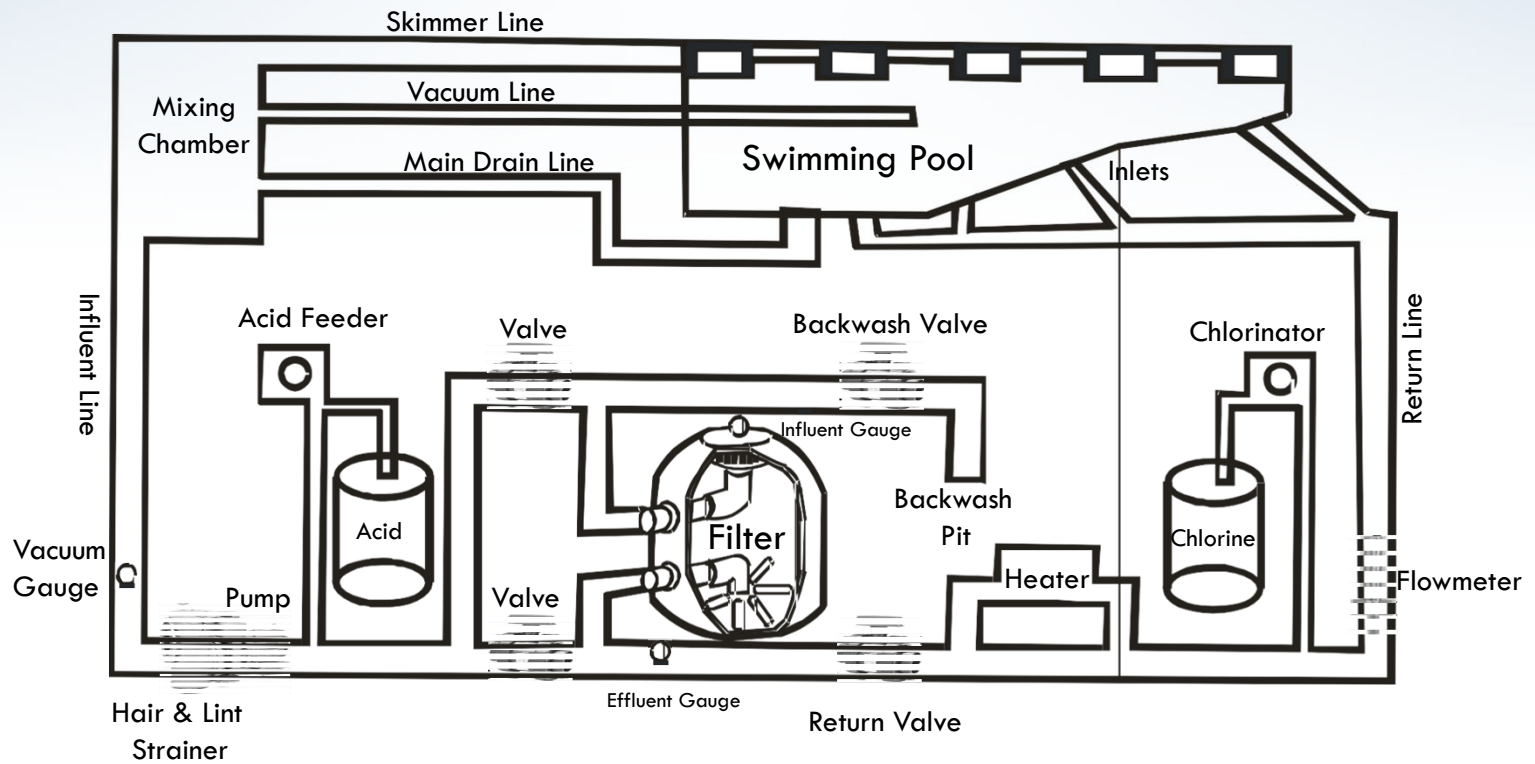
Remember: If you have questions about how to operate your system, ask your supervisor. Do not turn the system on with valves in a position that you are not certain about. The water must be able to get out of the pool via the main drain, skimmer and/or vacuum line and it must have somewhere to go via the return or backwash line. **IF YOU DON'T KNOW.....ASK!!!!**



# THE CIRCULATION SYSTEM



# WATER CIRCULATION



# AIR CIRCULATION SYSTEM

Air Handling System is equipment that brings in outdoor air into a building and removes air from a building for the purpose of introducing air with fewer contaminants and removing air with contaminants created while bathers are using aquatic venues. The system contains components that move and condition the air for temperature, humidity, and pressure control, and transport and distribute the air to prevent condensation, corrosion, and stratification, provide acceptable indoor air quality, and deliver outside air to the breathing zone.

The air system should maintain the relative humidity between 40% and 60%. It should also maintain the temperature ideally within 2 °F (no more than 5 °F) of the water temperature. The inability to maintain the appropriate humidity and temperature level promotes corrosion, poor air quality, mold, mildew and bather discomfort.

Air return vent should be low to the floor to extract chloramines hovering above the surface of the water in order to improve air quality. Air system should be changing 4 to 8 times per hour depending on whether mechanical cooling are not



# LET'S REVIEW!

Fill in the chart with the range of each reading and the chemical name that raises and lowers each reading:

Chemical	Minimum	Maximum	Raise	Lower
Chlorine	1.0	10	Add chlorine	Add Sodium Thiosulfate
pH	7.2	7.8	Add Soda Ash	Add Muriatic Acid
Total Alkalinity	60	180	Add Baking Soda	Add Muriatic Acid
Calcium Hardness	150	400	Add Calcium Chloride	Drain And Fill
Cyanuric Acid	30	50	Add Cyanuric Acid	Drain And Fill
Total Dissolved Solids	No Minimum	1500	Not Applicable	Drain And Fill





**CHAPTER  
NINE**

# FILTRATION





# FILTRATION SYSTEMS

FILTER MEDIA – MATERIAL WITHIN THE FILTER THAT COLLECTS THE DIRT

- SAND FILTERS
- CARTRIDGE
- DIATOMACEOUS EARTH (DE)

- PARTICLES MEASURED IN MICRONS
  - 25 MICRONS = 1/1000 OF AN INCH
  - HUMAN VISION – DOWN TO 40 MICRONS
- EFFECTIVENESS
  - DEPENDS ON TYPE OF FILTER, SURFACE AREA, VELOCITY OF WATER, CONDITION OF MEDIA, & AMOUNT OF PARTICULATE MATERIAL IN WATER





# FILTERS



- REMOVES FINE SUSPENDED PARTICLES FROM WATER
- MOST COMMON MEDIA
  - SAND
  - DIATOMACEOUS EARTH
  - FABRIC CARTRIDGES

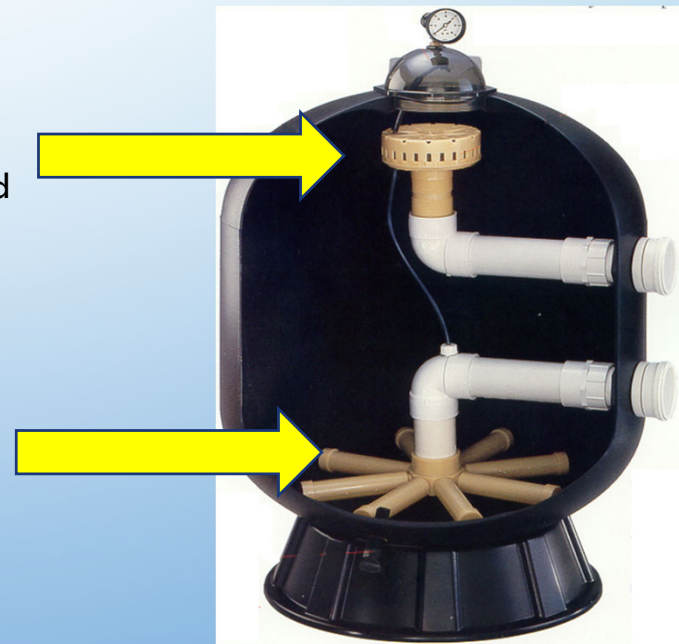


# SAND FILTERS

**Sand Filters:** Sand filters typically can filter particulate matter down to about 15 microns. One micron is one millionth of a meter (0.00004 inch). The human can see particles 40 microns or larger. Water comes into the filter through the filter influent line (the top pipe) and is dispersed through the diffuser.

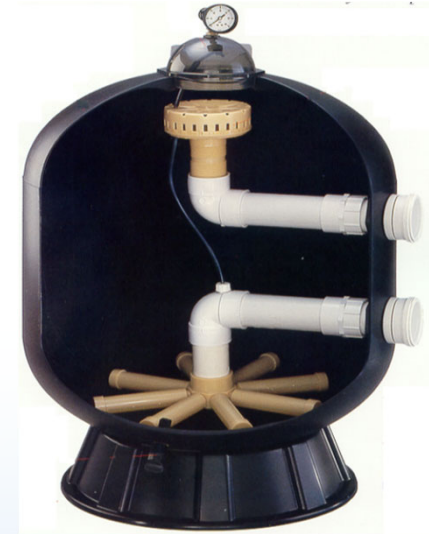
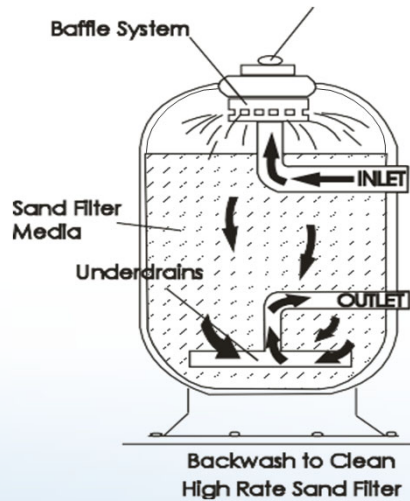
The **diffuser** evenly distributes the water through the sand bed. The sand in a filter is #20 grade filter sand. This is the only grade of sand that should be placed in the filter. Most manufacturers recommend that a layer of pea gravel be placed in the bottom of the tank to support the laterals.

**Laterals** are the rungs located in the bottom of the filter tank. They have very fine slots in them that allow the water to pass through but not sand. A broken lateral can allow sand into the return lines and thus into the pool basin



# HIGH RATE SAND FILTERS

## FILTERING



- REMOVES SMALL PARTICULATE MATTER FROM THE WATER
- WATER MOVES IN THROUGH THE INFLUENT LINES AND THROUGH THE DIFFUSER (BAFFLE SYSTEM)
- DIRT COLLECTS ON THE TOP OF THE SAND OR IN THE FIRST FEW INCHES OF THE SAND MEDIA
- CLEAN WATER EXITS THE TANK THROUGH THE LATERALS
- HIGH RATE SAND FILTER USE ONLY ONE SIZE OF SAND
- A BROKEN LATERAL WILL ALLOW SAND TO ENTER THE POOL THROUGH THE RETURN LINES



# MANUAL AIR RELIEF VALVES

- BLEEDS OUT EXCESS AIR TRAPPED INSIDE OF THE TANK
- RELIEF VALVE IS LOCATED ON THE TOP OF THE TANK
- ALWAYS TURN OFF THE PUMP AND OPEN THE AIR-RELIEF BEFORE LOOSENING CLAMPS OR LIDS



# FILTER PERFORMANCE

## Design Flow Rate

$$\begin{array}{rcl} \text{Filter Rate} & \times & \text{Filter Area} & = & \text{Flow Rate} \\ (\text{gpm/sq.ft.}) & & (\text{sq.ft.}) & & (\text{gpm}) \end{array}$$

Example:

A high rate sand filter has a filter rate of 25 gpm and a filter area of 5.0 sq.ft.

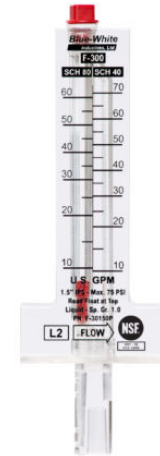
$$\begin{array}{r} 25 \\ \times 5.0 \\ \hline 125 \text{ gpm} \end{array}$$

What is the design flow rate of the filter?

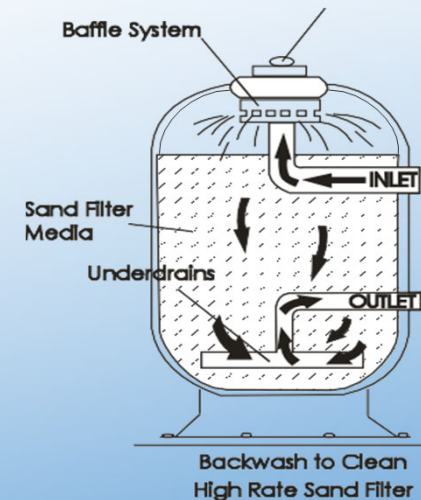




# BACKWASHING



- THE REVERSE OF FLOW THROUGH THE FILTER IN ORDER TO CLEAN IT
- TWO THINGS INDICATE IT IS TIME TO BACKWASH
- FLOWRATE IS TOO LOW (BELOW MINIMUM)
- PRESSURE DIFFERENTIAL IS MORE THAN 15



# CALCULATING THE PRESSURE DIFFERENTIAL

## INFLUENT PRESSURE

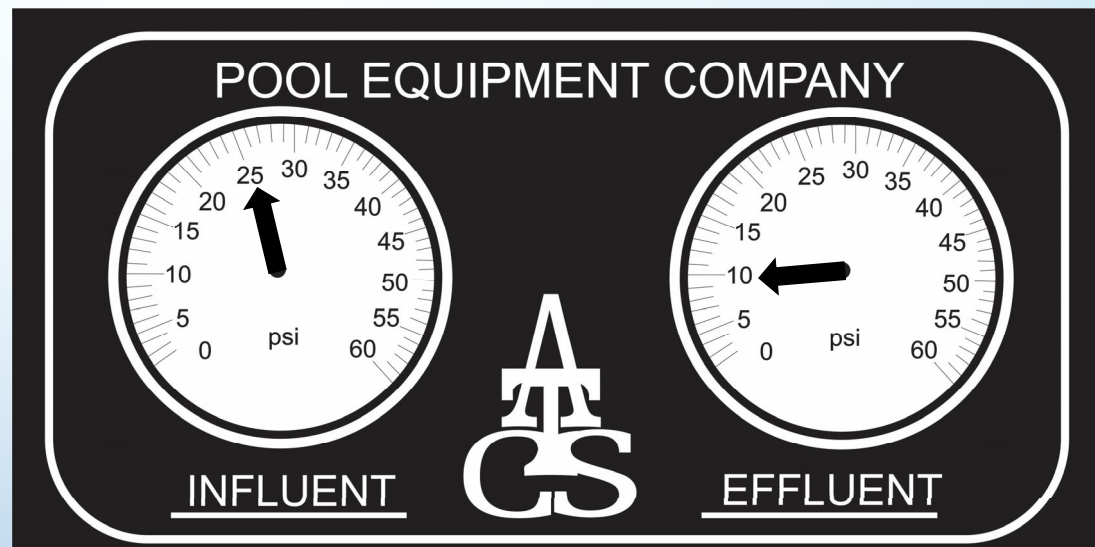
- GAUGE LOCATED ON TOP OF THE FILTER, ON PIPING FROM PUMP TO FILTER, ON GAUGE PANEL INDICATES PRESSURE EXERTED BY PUMP TO PUSH WATER THROUGH FILTER

## EFFLUENT PRESSURE

- GAUGE LOCATED ON PIPING AFTER FILTER OR ON GAUGE PANEL INDICATES PRESSURE EXERTED BY PUMP TO PUSH WATER BACK TO THE POOL AFTER IT EXITS THE FILTER

## PRESSURE DIFFERENTIAL

- THE DIFFERENCE BETWEEN INFLUENT AND EFFLUENT PRESSURE



## EXAMPLE:

INFLUENT = 25 PSI

EFFLUENT = 10 PSI

DIFFERENTIAL = 15 PSI



# WHEN TO CLEAN THE FILTER BASED ON PRESSURE DIFFERENTIAL

- WHEN THE PRESSURE DIFFERENTIAL INCREASES FROM ITS STARTING DIFFERENTIAL TO THE FOLLOWING, ITS TIME TO BACKWASH/CLEAN THE FILTER:

- RAPID RATE SAND FILTERS            8 PSI
- HIGH RATE SAND FILTERS            15 PSI
- CARTRIDGE FILTERS                    10 PSI
- PRESSURE DE                            10 PSI
- VACUUM DE                                15" OF HG.

(UNLESS MANUFACTURERS INSTRUCTIONS DIFFER)





***Important!***

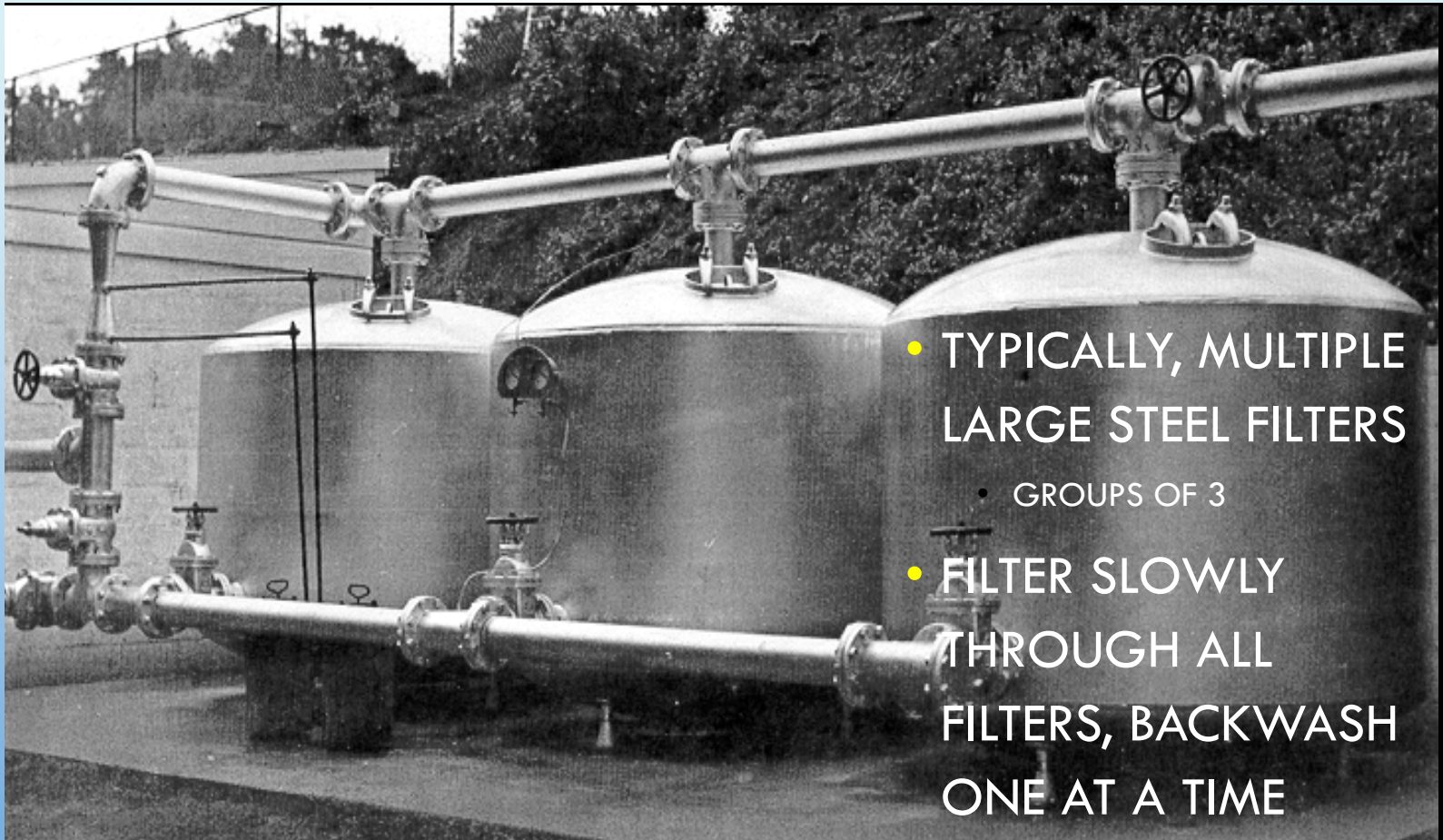
## **BACKWASHING**

Do not open and close valves while the pump is running.

Always shut the pump “off” first



# RAPID RATE SAND FILTERS

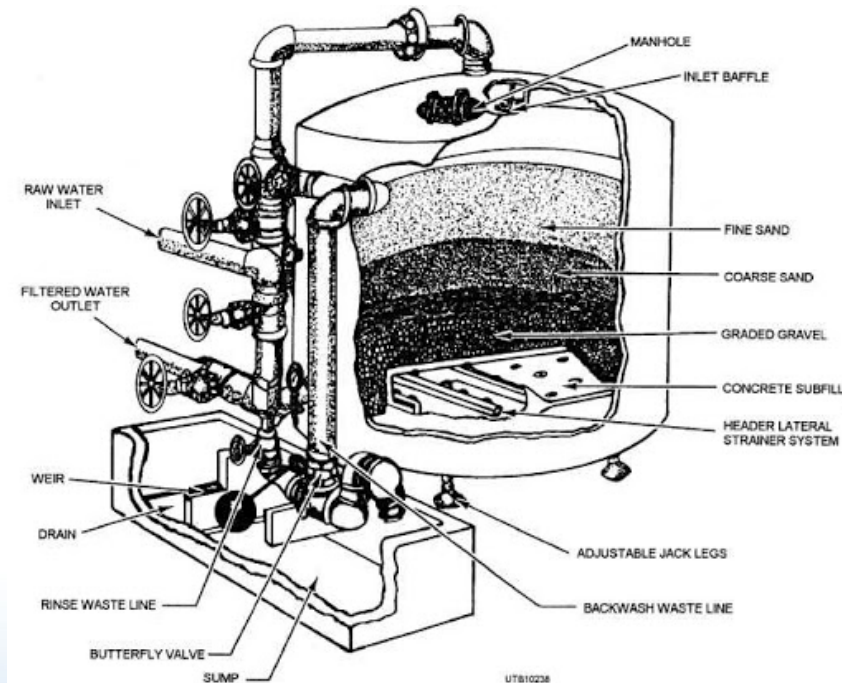


- TYPICALLY, MULTIPLE LARGE STEEL FILTERS
  - GROUPS OF 3
- FILTER SLOWLY THROUGH ALL FILTERS, BACKWASH ONE AT A TIME



# RAPID RATE SAND FILTERS

- THIN BED OF SAND SUPPORTED BY LAYERS OF GRADED GRAVEL
  - MAJORITY OF FILTRATION DONE IN TOP 3-4 INCHES
- REQUIRE USE OF FILTER ALUM (ALUMINUM SULFATE)
- USUALLY HAS MULTIPLE TANKS THAT ARE BACKWASHED ONE AT A TIME



## FILTER SAND

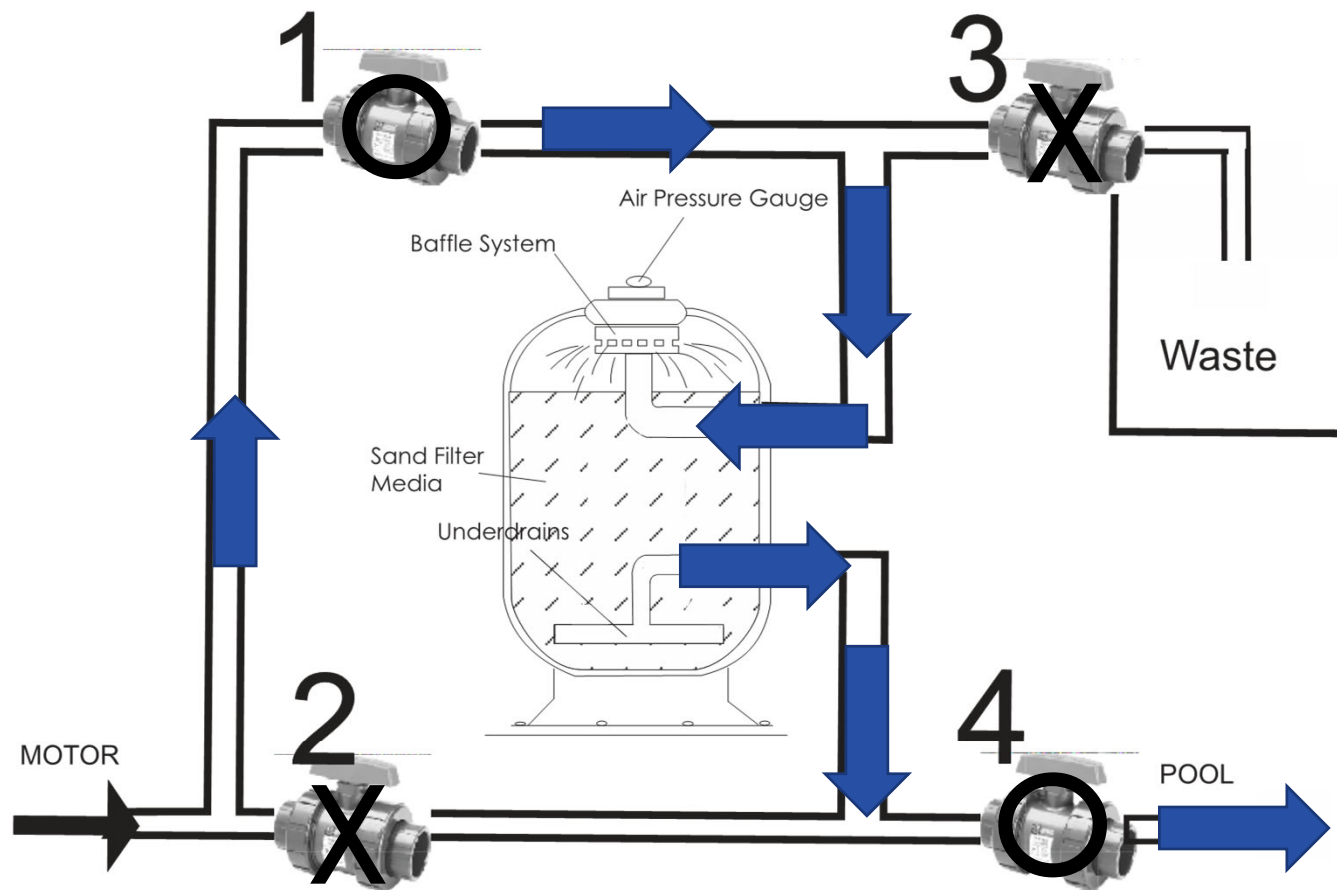
SAND IS GENERALLY SCHEDULED TO BE CHANGE EVERY 2-3 YEARS IN AN INDOOR POOL AND EVERY 7-8 YEARS IN AN OUTDOOR SEASONAL POOL. THIS IS CALLED RECHARGING THE SYSTEM. SAND HAS VERY SHARP EDGES OBSERVED UNDER A MICROSCOPE. OVER TIME, THE SAND PARTICLES LOSE THEIR SHARP EDGES AND THEY SMOOTH MAKING IT MORE DIFFICULT TO CATCH SMALL PIECES OF DIRT.

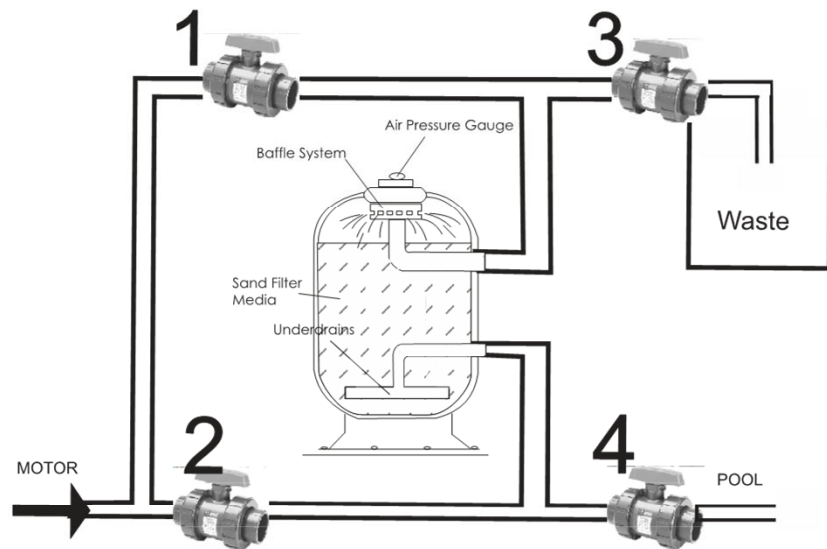
## FLOCCULENT

SOMETIMES A COAGULANT OR FLOCCULENT IS USED TO AID A SAND FILTER. A FLOCCULENT CREATES AN ADDITIONAL LAYER ON THE FILTER BED THAT CAUSES SMALL PIECES OF DIRT TO STICK TOGETHER. NOW THAT THE DIRT IS BIGGER, THE FILTER IS ABLE TO REMOVE THE DIRT. **ALUMINUM SULFATE (ALUM) IS THE MOST COMMONLY USED FLOCCULATING AGENT.** A FLOCCULENT IS ONLY GOOD FOR ONE FILTER RUN AS IT WILL BE REMOVED BY BACKWASHING.



# FILTER

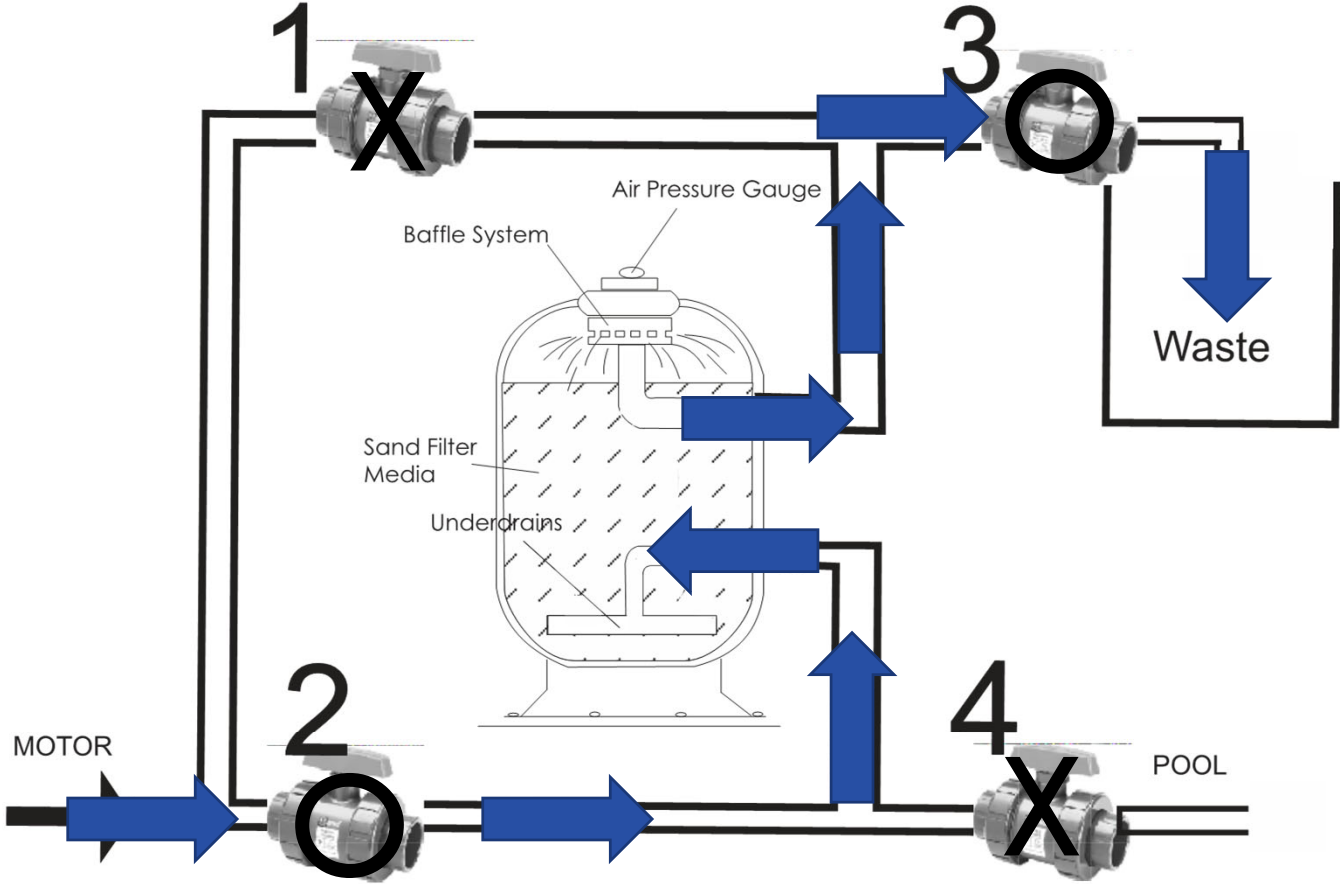


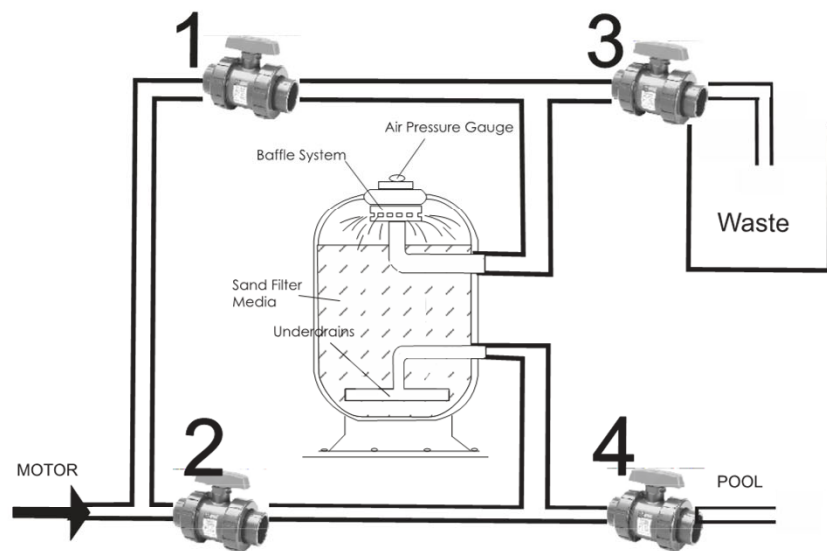


	OPEN	CLOSED
FILTER	1, 4	2, 3
BACKWASH	_____	_____
DRAIN	_____	_____
BYPASS	_____	_____



# BACKWASH



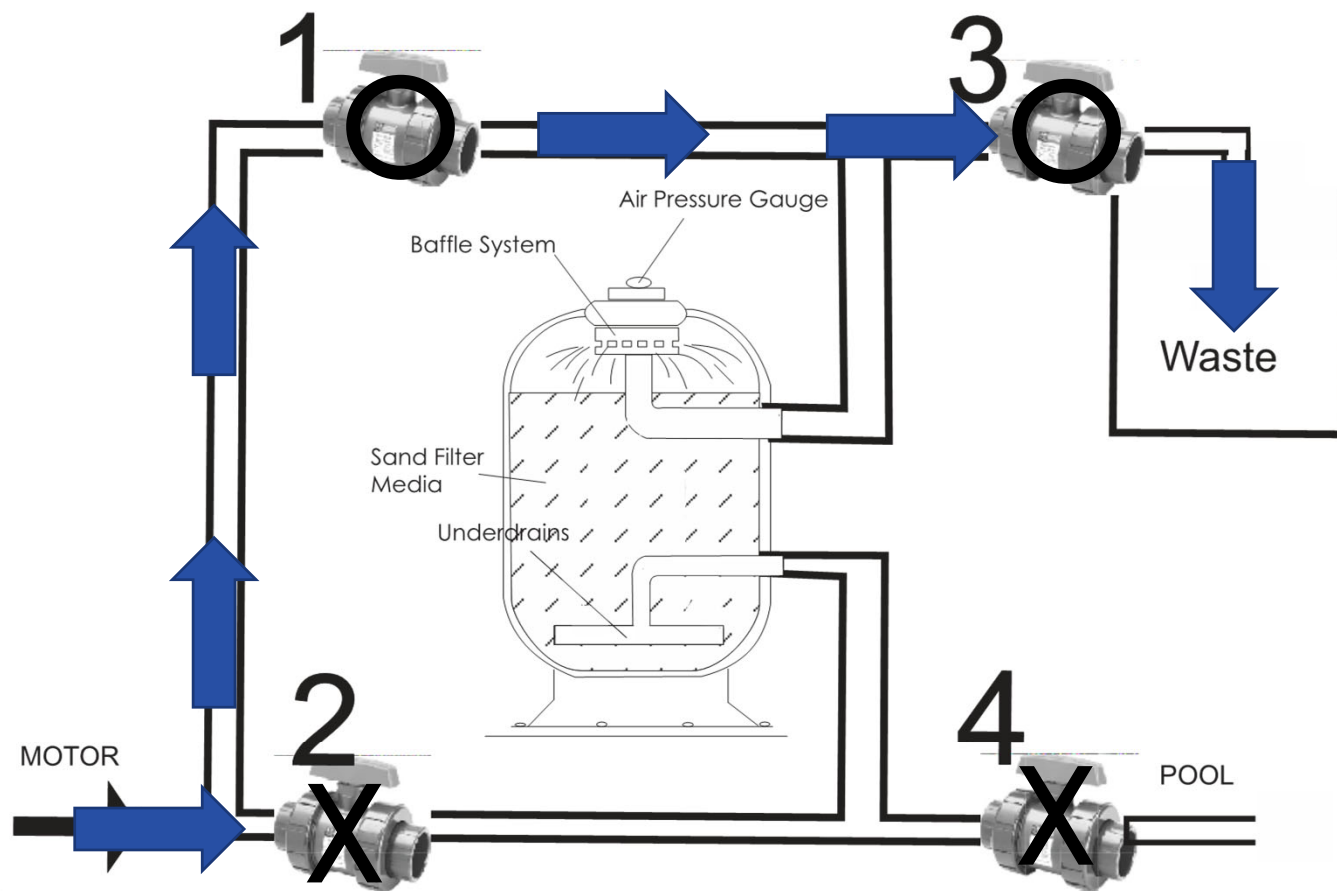


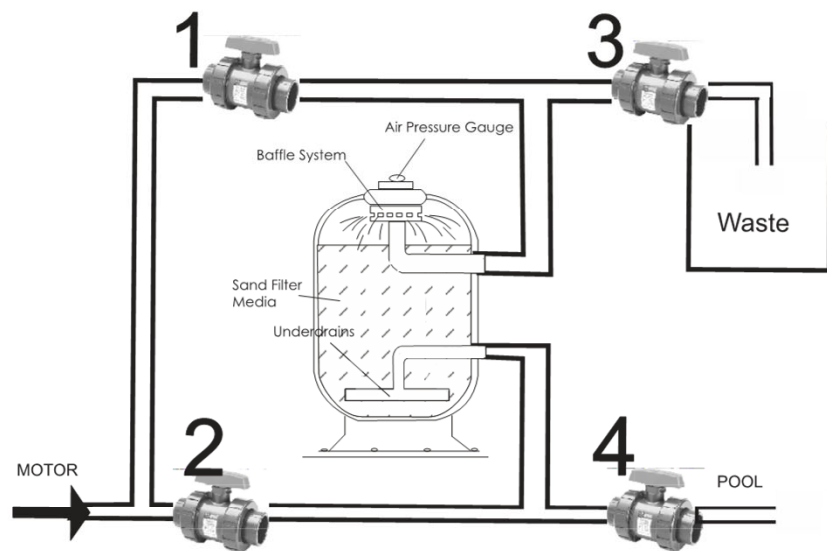
	OPEN	CLOSED
FILTER	<u>1, 4</u>	<u>2, 3</u>
BACKWASH	<u>2, 3</u>	<u>1, 4</u>
DRAIN	_____	_____
BYPASS	_____	_____





# DRAIN

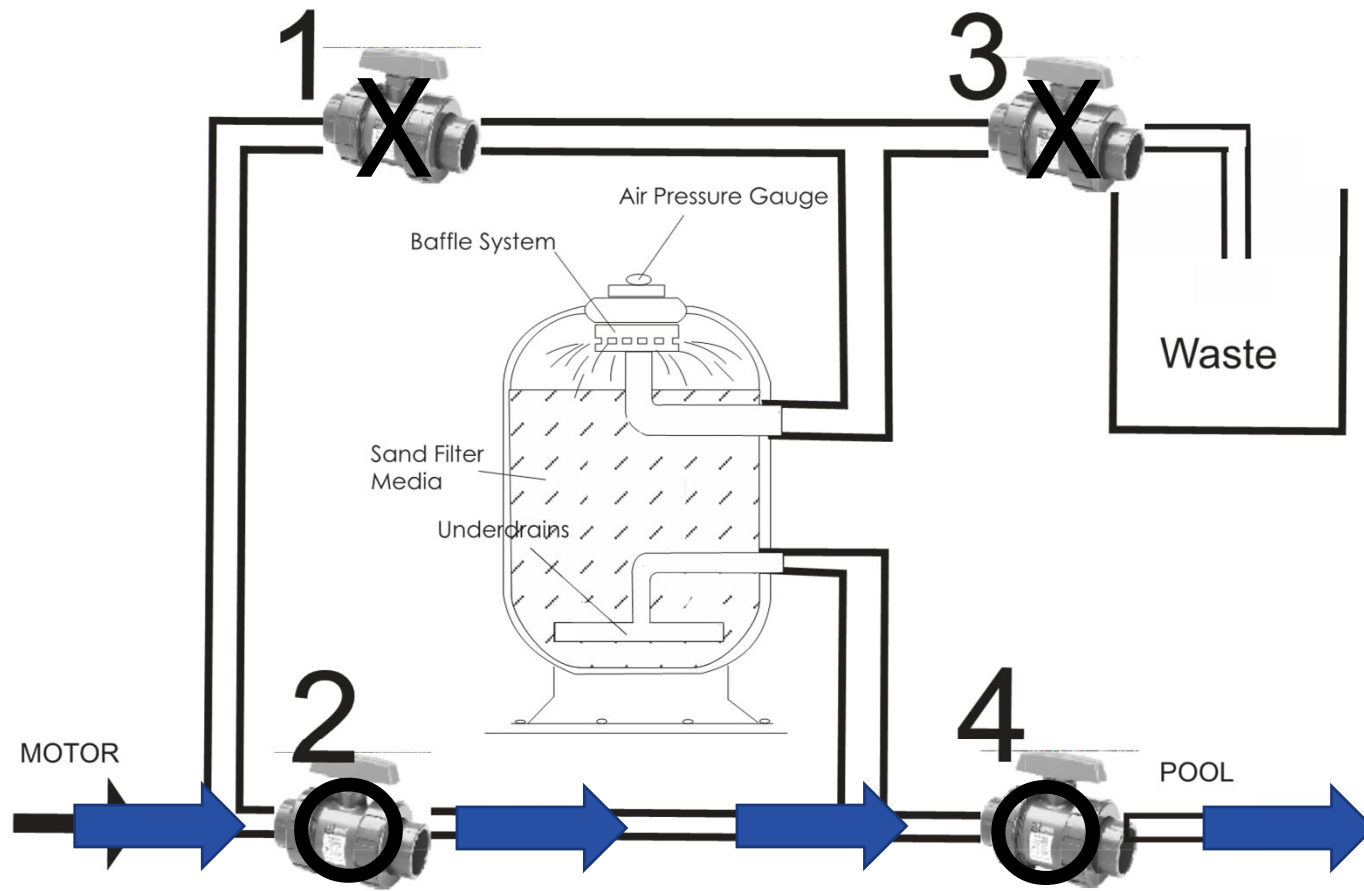


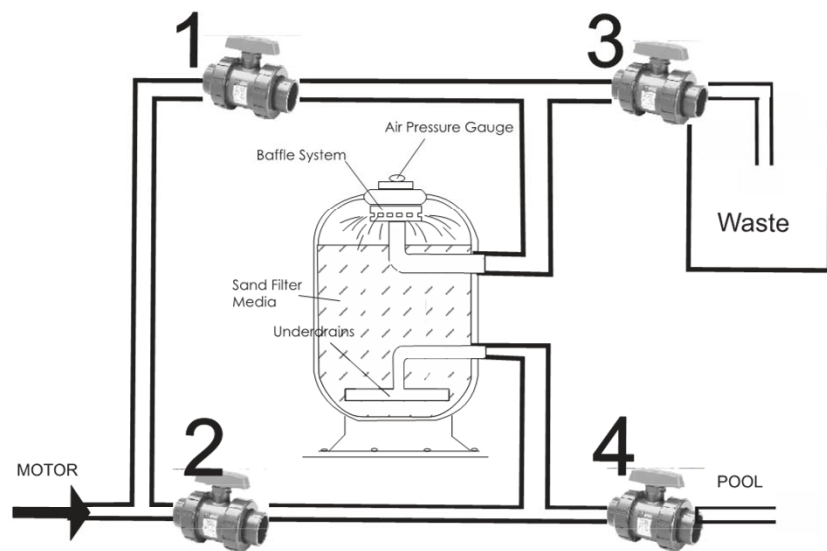


	OPEN	CLOSED
FILTER	<u>1, 4</u>	<u>2, 3</u>
BACKWASH	<u>2, 3</u>	<u>1, 4</u>
DRAIN	<u>1, 3</u>	<u>2, 4</u>
BYPASS	<u>          </u>	<u>          </u>



# BYPASS





	OPEN	CLOSED
FILTER	<u>1, 4</u>	<u>2, 3</u>
BACKWASH	<u>2, 3</u>	<u>1, 4</u>
DRAIN	<u>1, 3</u>	<u>2, 4</u>
BYPASS	<u>2, 4</u>	<u>1, 3</u>





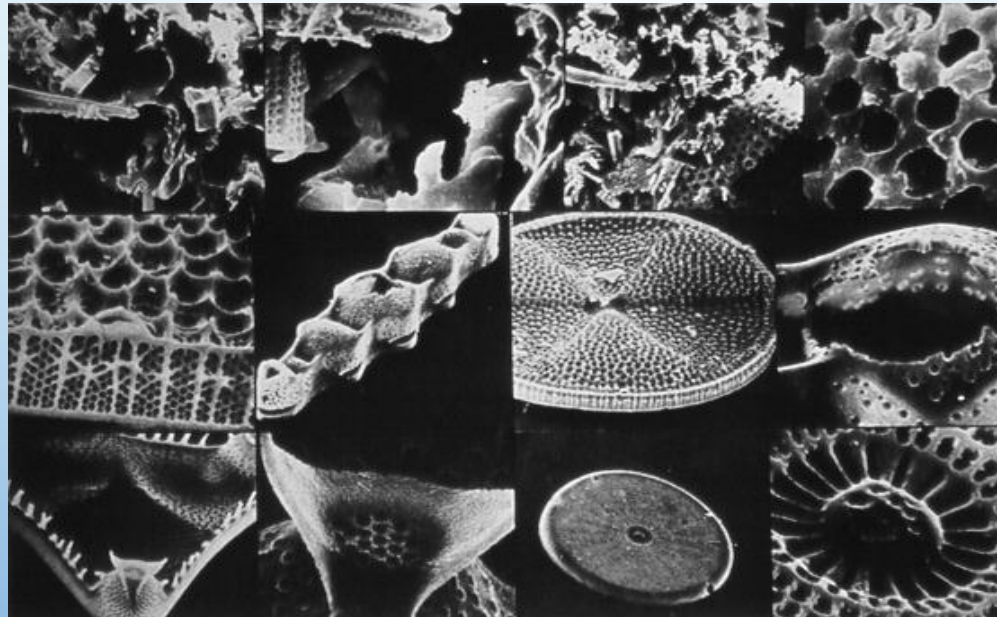
## MULTI-PORT VALVES

**MULTI-PORT VALVES** ARE USED ON SMALLER SYSTEMS AND CAN SET THE FILTER SYSTEM TO FILTER, BACKWASH AND DRAIN IN THE TURN OF A HANDLE. IT IS IMPORTANT TO ALWAYS TURN OFF THE PUMP BEFORE TURNING THE HANDLE. IT IS USUALLY A GOOD IDEA TO PUT THE SYSTEM IN RINSE FOR A FEW MOMENTS AFTER BACKWASHING BEFORE PUTTING THE SYSTEM BACK INTO FILTER.

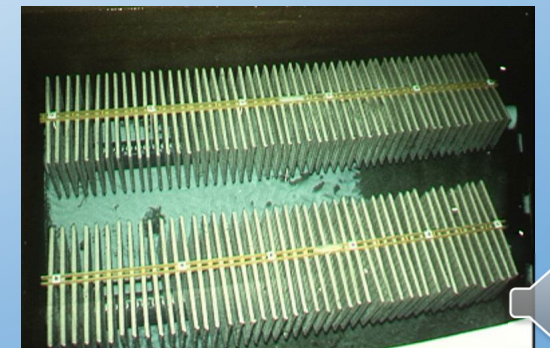
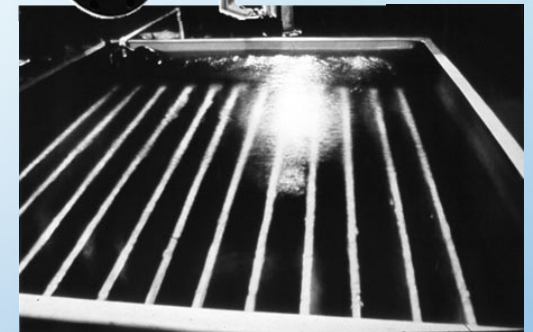
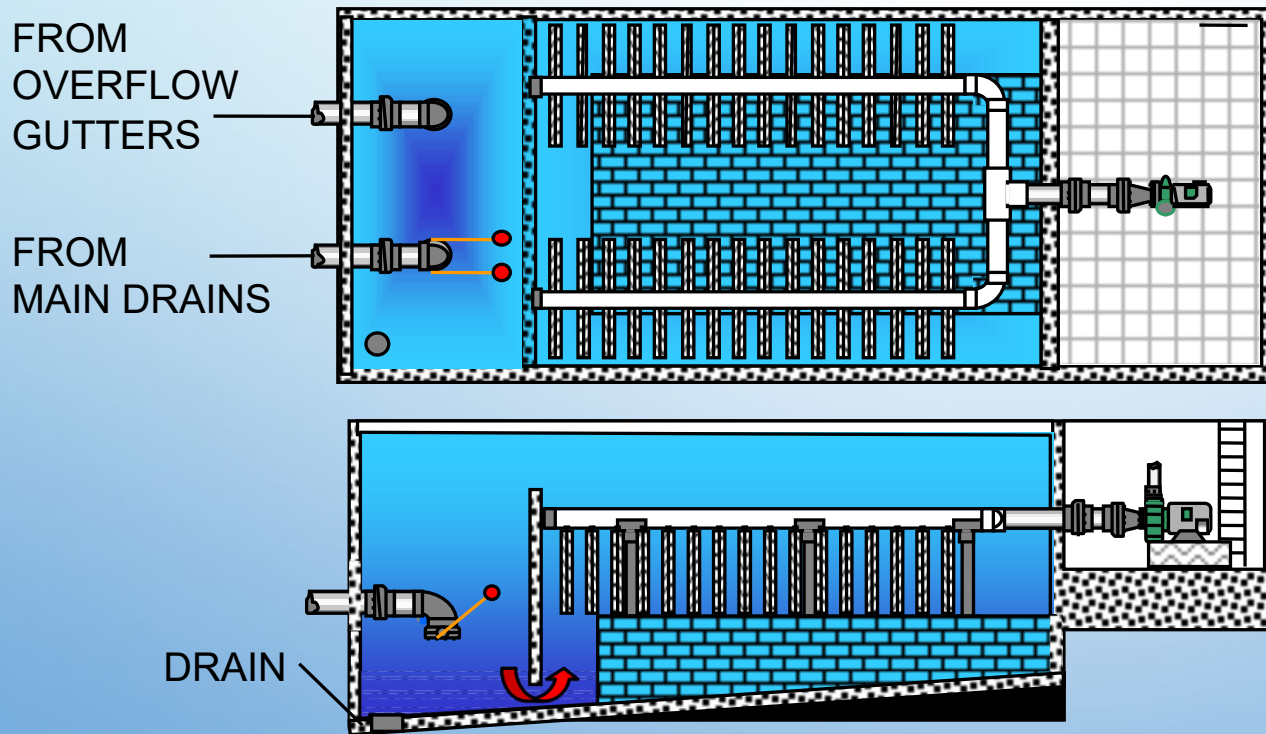


# DIATOMACEOUS EARTH FILTERS

- DIATOMACEOUS EARTH ARE FOSSIL REMAINS OF MARINE LIFE - DIATOMS

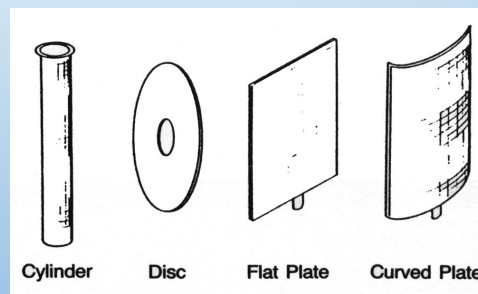


# VACUUM DE FILTERS



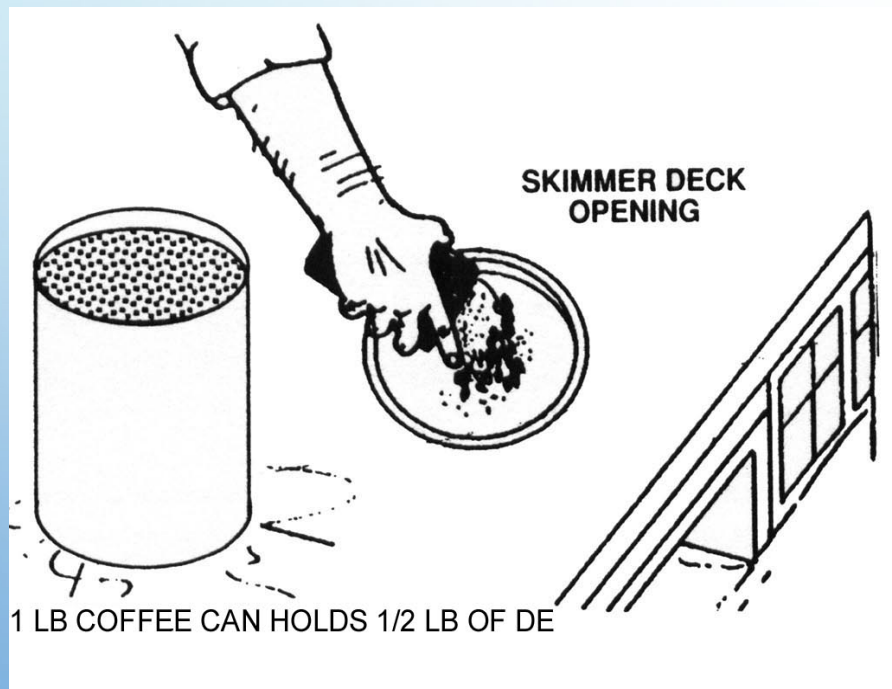
# DE FILTERS PRESSURE

- DE SUPPORTED IN THE TANK BY FABRIC COVERED LEAVES, RODS OR DISKS (SEPTUMS)





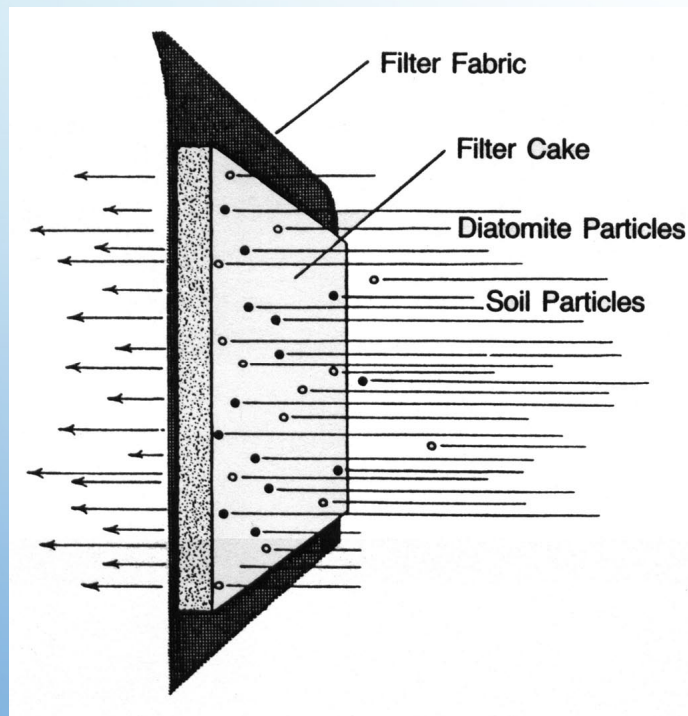
# PRE-COATING THE CARTRIDGES



- **PRE-COAT POT:** A ONE-SHOT DEAL BETWEEN FILTER RUNS. THE ENTIRE AMOUNT THAT THE FILTER HOLDS IS ADDED ALL AT ONCE. IT THOROUGHLY COATS ALL THE ELEMENTS AND, UNLESS THE SYSTEM GOES OFF FOR ANY REASON, IT WILL REMAIN THERE UNTIL THE FILTER RUN IS OVER AND NEEDS TO BE CLEANED. (SEE “CLEANING D.E FILTERS” BELOW) THE ELEMENTS SHOULD BE PRECOATED WITH **2 OUNCES OF D.E. PER SQUARE FOOT OF FILTER SURFACE AREA.** DO NOT FORGET TO COUNT THE FRONT AND THE BACK OF THE ELEMENT WHEN CALCULATING THE TOTAL SURFACE AREA OF THE ELEMENT.



# HOW THE DE CARTRIDGES WORK



- WATER MUST PASS THROUGH DE COATED SEPTUMS TO GET CLEAN
- THE DE AND THE DIRT EVENTUALLY FALLS TO THE BOTTOM OF THE PIT
- THIS WILL HAVE TO BE MANUALLY CLEANED



# SLURRY FEED

**SLURRY FEEDER:** THE D.E IS ALSO ADDED SLOWLY DURING THE FILTER RUN. BY THE END OF THE FILTER RUN, THE ELEMENTS WILL HAVE THE D.E. “CAKED ON”. THIS MAKES THEM A LITTLE HARDER TO CLEAN BUT IT ALSO EXTENDS THE FILTER RUN AND, SOME SAY, THIS METHOD ALLOWS THE FILTER TO WORK EVEN BETTER. THE DOWNFALL CAN BE THAT THE “CAKED ON” DIATOMACEOUS EARTH IS VERY DIFFICULT TO REMOVE FOR CLEANING.

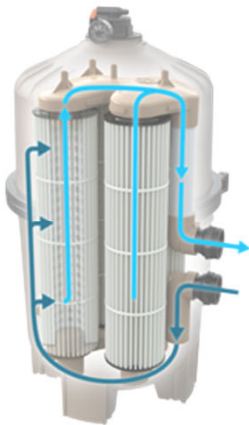
## CLEANING THE DE FILTER

FOR VACUUM DE FILTER

- CLEAN ALL SKIMMER BASKETS
- TURN OFF ALL CHEMICAL FEEDERS
- TURN OFF THE SYSTEM (THIS WILL ALLOW THE D.E TO FALL TO THE BOTTOM OF THE FILTER TANK)
- WASH OFF THE ELEMENTS AND REMOVE THE D.E FROM THE TANK (IT IS THROWN AWAY)\*\*
- CLEAN ELEMENTS AND REPLACE OR PUT A SECOND SET OF ELEMENTS IN AND SOAK THE DIRTY ONE IN A CLEANING SOLUTION TO DEGREASE IT
- PRE-COAT OR SLURRY FEED THE D.E BACK INTO THE FILTER AND RESUME OPERATION
- TURN ON ALL CHEMICAL FEEDERS

PRESSURE DE FILTERS ARE BACKWASHED





*Photos provided by Hayward*

## CARTRIDGE FILTERS

- SINGLE LARGE OR CLUSTER OF CARTRIDGES
- FILTERS OUTSIDE TO INSIDE OF CARTRIDGE
- DIRT CAPTURED IN FABRIC



## CARTRIDGE FILTERS

- CLEAN WHEN PRESSURE DIFFERENTIAL INCREASES 10 PSI
  - TURN SYSTEM OFF
  - REMOVE DIRTY CARTRIDGES
  - INSTALL CLEAN SPARES
  - MANUALLY CLEAN DIRTY ONES

# VALVES



Valve handle is 90°, or perpendicular to pipe –  
Valve is closed

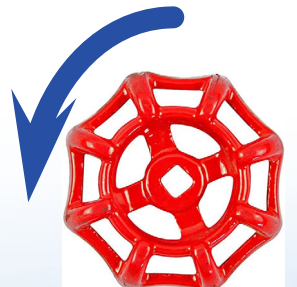
Valve handle in line with pipe –  
Valve is open



*Photos provided by Hayward*



Righty Tighty



Lefty Loosey



# MULTI-PORT VALVES (ROTARY)



**Multi-port valves** are used on smaller systems and can set the filter system to filter, backwash and drain in the turn of a handle. It is important to always turn off the pump before turning the handle. It is usually a good idea to put the system in rinse for a few moments after backwashing before putting the system back into filter.



Photos provided by Hayward



## FORMULAS FOR POOL CAPACITY

L = length      W = width      V = volume      D = depth      r = radius (half of the diameter of a circle)  
 $\pi$  = (pi) 3.14 (a factor used in calculations with circles)

### SURFACE AREA

Rectangular pool =  $L \times W$       Circular pool =  $r^2 \times \pi$       Right triangle =  $(L \times W) \div 2$

### AVERAGE DEPTH

For constant slope:  $[D \text{ (minimum)} + D \text{ (maximum)}] \div 2 = \text{AVERAGE DEPTH}$

Note: For multi-depth pools calculate the volume in sections of constant slope and add them together.

### CUBIC FEET OF VOLUME (surface area times average depth)

Rectangular pool  $V = L \times W \times D$       Circular pool  $V = r^2 \times \pi \times D$

### POOL GALLONAGE IN CUBIC FEET (cubic foot of water = 7.5 gallons)

Rectangular pool gallons =  $L \times W \times D \times 7.5$       Circular pool gallons =  $r^2 \times \pi \times D \times 7.5$

### TURNOVER RATE

Pool Volume / Flow rate / 60 = Hours

### FLOW RATE (MINIMUM)

Pool Volume / Turnover Rate (hours) / 60 = Gallons per Minute

### FILTER SURFACE AREA

Flow rate / Filtering Ability = Sq Feet of Filter Surface Area Needed

### FILTER FLOW CAPACITY

Filter Area x Filtering Ability x Required Turnover x 60 = Filter Flow Capacity

### MAXIMUM USER CAPACITY

Surface Area / Health code bather capacity = Max User Capacity



# Quick Reference Conversion Table

**Ounces to Pounds**

**Ounces  $\div$  16 = Pounds**

**Fluid Ounces to Gallons**

**Fluid Ounces  $\div$  128 = Gallons**

**Liters to Gallons**

**Liters  $\div$  3.785 = Gallons**

**Fluid Ounces to Cups**

**Fluid Ounces  $\div$  8 = Cups**

**Yards to Feet**

**Yards  $\times$  3 = Feet**

**Cubic Feet to Gallons**

**Cubic Feet  $\times$  7.48 = Gallons**

**Milliliters to Liters**

**Milliliters  $\div$  1000 = Liters**

**Meters to Feet**

**Meters  $\times$  3.28 = Feet**

**Fluid Ounce to Milliliters**

**Fluid Ounce  $\times$  29.57 = Milliliters**

**Celsius ( $^{\circ}$ C) to Fahrenheit ( $^{\circ}$ F)**

**$^{\circ}$ F =  $(9/5 \times ^{\circ}$ C) + 32**

**Fahrenheit ( $^{\circ}$ F) to Celsius ( $^{\circ}$ C)**

**$^{\circ}$ C =  $5/9 \times (^{\circ}$ F - 32)**

**Quarts to Gallons**

**Quarts  $\div$  4 = Gallons**

# CALCULATIONS

## EXAMPLE 1: Rectangle Pool Volume in Gallons & Flowrate

Length of pool: 75 Feet  
Width of pool: 50 Feet  
Shallow End Depth: 3 Feet  
Deep End Depth: 5 Feet  
Turnover Rate: 6 Hours

$$\begin{aligned}\text{Volume of pool in gallons} &= \text{Length} \times \text{Width} \times \text{Average Depth} \times 7.5 \\ \text{Volume of pool in gallons} &= 75 \times 50 \times 4 \times 7.5 = 112,500 \text{ gallons}\end{aligned}$$

$$\text{Minimum Flow Rate} = \frac{\text{Capacity}}{(\text{Turnover Rate} \times 60)} = \frac{112,500}{(6 \times 60)} = 312.5 \text{ gpm}$$

# CALCULATIONS

## EXAMPLE 2: Circular Pool Volume in Gallons & Flowrate

Diameter of pool:	8 Feet
Shallow End Depth:	3 Feet
Deep End Depth:	4 Feet
Turnover Rate:	30 Minutes

$$\begin{aligned}\text{Volume of pool in gallons} &= \text{Radius} \times \text{Radius} \times 3.14 \times \text{Average Depth} \times 7.5 \\ \text{Volume of pool in gallons} &= 4 \times 4 \times 3.14 \times 3.5 \times 7.5 = 1,319 \text{ gallons}\end{aligned}$$

$$\text{Minimum Flow Rate} = \frac{\text{Capacity}}{(\text{Turnover Rate})} = \frac{1,319}{(30)} = 44 \text{ gpm}$$

# CALCULATIONS

## EXAMPLE 3: Current Turnover Rate

Volume of pool: 150,000  
Current Flow Rate: 350 gpm

$$\text{Current Turnover Rate} = \frac{\text{Capacity}}{(\text{Flow Rate} \times 60)}$$

Turnover Rate in hours:  $150,000 \div 350 \div 60 = 7.14$  hours



CHAPTER  
TEN

# SPA MAINTENANCE





# HOT WATER CHEMISTRY AND SPAS

- MENTAL AND PHYSICAL RELAXATION DERIVED
- CONDUCIVE TO RELIEVING ACHES AND PAINS
- EVERY PUBLIC SPA SHOULD BE MAINTAINED BY A PSPO

# HOT WATER CHEMISTRY AND SPAS

*However*

A CASUAL ATTITUDE BY PSPO CAN  
QUICKLY PRODUCE AN  
UNHEALTHY AND UNSAFE  
ENVIRONMENT



# CHANGE THE WATER FREQUENTLY

- SMALLER VOLUME OF WATER THAN POOL,  
MORE BATHERS PER GALLON
- GREATER AMOUNT OF BATHER WASTE IN  
A SPA



The formula for changing the water

Volume of water  $\div$  3 = Number of Uses before change

For example, if there are 600 gallons of water in the hot tub then every 200 uses ( $600 \div 3 = 200$ ) it should be changed.

If 20 people each day use the hot tub then the water should be changed every 10 days ( $200 \div 20 = 10$ )





# FOAMING

- CAUSED BY COMBINATION OF

- ORGANIC WASTES AND AERATION; OR
- OVERUSE OF SIMPLE QUAT ALGICIDES;
- HIGH BATHER LOADS;

- HIGH PH;
- HIGH TDS;
- HIGH TEMPERATURE



- DEFOAMERS – SILICON BASED EMULSIONS
  - BREAK-UP AND DISSIPATE FOAM
  - THEY ONLY HIDE THE PROBLEM, THEY DON'T SOLVE THE PROBLEM

## WATER REPLACEMENT INTERVAL (WRI)

- IF TDS EXCEEDS THE SOURCE-WATER TDS BY 1,500 PPM OR MORE; OR
- WRI LESS THAN OR EQUAL TO THE NUMBER DAYS SINCE LAST TIME WATER WAS DRAINED;
  1. DRAIN THE SPA COMPLETELY
  2. CLEAN IT THOROUGHLY
  3. REFILL IT WITH SOURCE WATER



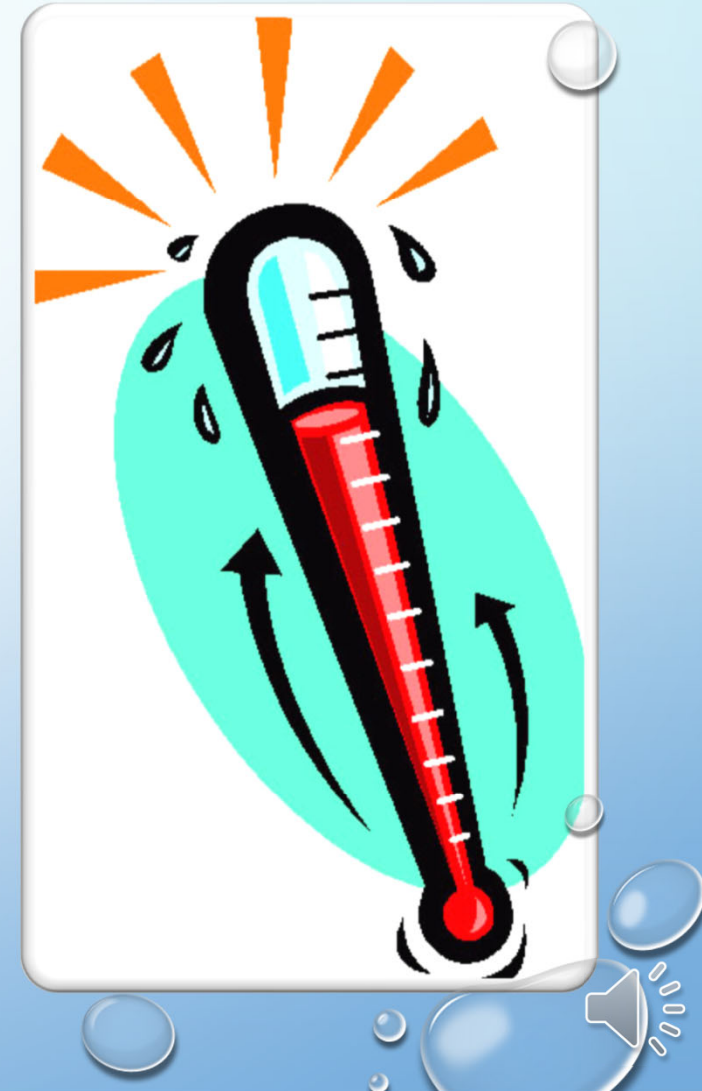


# TEMPERATURE

- EFFECTS
  - BACTERIAL GROWTH
  - CHEMICAL REACTION TIME
  - HIGH TDS
  - HIGHER LEVELS OF ORGANIC WASTE
- IDEAL 98° - 104°
- MAXIMUM SUGGESTED TEMPERATURE 104°

## DISEASE PREVENTION

- TOTAL DISSOLVED SOLIDS WILL DEVELOP QUICKLY IN A SMALL VOLUME OF WATER
- AT 102° THE AVERAGE PERSON PERSPIRES 3 PINTS OF FLUID PER HOUR, COMPARED TO 1 PINT PER HOUR FOR A SWIMMER IN A POOL
- PSEUDOMONAS LIVES IN WATER 89 DEGREES OR WARMER
- SANITIZER WILL NOT LAST AS LONG IN HIGH TEMPERATURES
- VIRUSES AND BACTERIA CAN GROW MORE RAPIDLY AT HIGH TEMPERATURES.



# HEATING PROBLEMS

- ALWAYS REMEMBER TO TURN OFF THE HEATER WHEN TURNING OFF THE CIRCULATION SYSTEM FOR ANY REASON. THE HEATING ELEMENTS CAN BE DAMAGED IF THE HEATER IS LEFT ON WHEN THE WATER IS NOT CIRCULATING.
- USING A COVER AT NIGHT OR WHEN THE SPA IS CLOSED WILL HELP RETAIN HEAT AND HELP CUT DOWN ON EVAPORATION.
- BE FAMILIAR WITH THE LOCAL HEALTH DEPARTMENT REGULATION FOR TEMPERATURE MAXIMUM. IT IS USUALLY 104°.



## HYPERTHERMIA

- NORMAL EXPOSURE TIME IN A SPA SHOULD BE 15 MINUTES
- LONGER EXPOSURE MAY RAISE BODY CORE TEMPERATURE TO AN ELEVATED LEVEL
- RESULTS IN ELEVATED BLOOD PRESSURE AND DROWSINESS

## RULES THAT SHOULD BE FOLLOWED:

- a. Use of the spa should be no longer than 15 minutes at a time. Blowers should be on a timer with a maximum of 15 minutes and should be out of reach of patrons using the spa.
- b. NO ALCOHOL.
- c. Persons with high blood pressure, elderly, children, and pregnant women should consult a physician before using the spa.
- d. Children under the age of five should not use the spa. (Most jurisdictions require higher disinfectant levels, and the higher temperatures can pose a health risk for young children. Many children tend to “swim” around and not just sit and relax. This movement can raise the body temperature even higher).

# SUCTION ENTRAPMENT

IN ADDITION TO THE CONSIDERATIONS COVERED  
IN CHAPTER 8:

- HAVE A STAFF PERSON WALK THROUGH SPA AREA AT LEAST EVERY ½ HOUR
- CHECK AND COMPLETE POOL AND SPA ENTRAPMENT HAZARD CHECKLIST DAILY PRIOR TO OPENING
- SPA SHOULD HAVE AN EMERGENCY SHUT OFF SWITCH CLOSE AT HAND



# TURNOVER RATE

- TURNOVER RATE - EVERY 30 MINUTES MINIMUM (SOME HEALTH DEPARTMENTS ARE 15 MINUTES I.E. VIRGINIA)
- THIS MEANS 48 TIMES PER DAY ALL OF THE WATER HAS GONE THROUGH THE FILTER





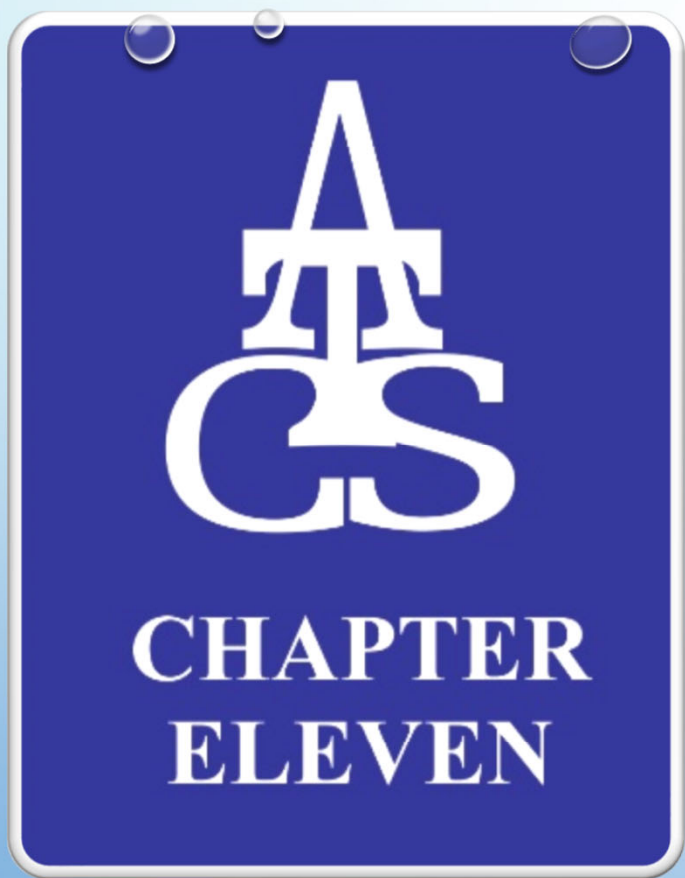


- SUGGESTED DISINFECTION RANGES:
  - CHLORINE 4.0 TO 10.0 PPM (IDEAL 4.0 - 5.0 PPM)
  - BROMINE 4.0 TO 8.0 PPM
- CHECK EVERY 1 HOUR

THE FOLLOWING QUALITIES MAKE BROMINE A BETTER DISINFECTANT IN SPA WATER:

1. BROMINE IS MORE STABLE THAN CHLORINE AT HIGHER TEMPERATURES
2. COMBINED BROMINE (BROMAMINES) PROVIDE DISINFECTION WHERE COMBINED CHLORINE (CHLORAMINES) DOES NOT.
3. BROMINE IS EFFECTIVE OVER A WIDER PH RANGE
4. COMBINED BROMINE ARGUABLY HAS LESS OF A DISAGREEABLE ODOR THAN COMBINED CHLORINE.
5. COMBINED BROMINE ELIMINATES SOME BATHER DISCOMFORT WHICH CAN BE CAUSED BY COMBINED CHLORINE.





# CHEMICAL SAFETY

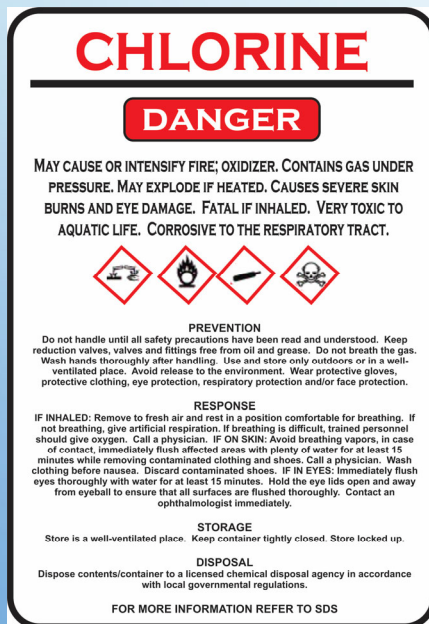


# CHEMICAL SAFETY

MANY CHEMICAL SAFETY HAZARDS EXIST IN THE AQUATIC ENVIRONMENT. TAKE PRECAUTIONS AT ALL TIMES!

MANY OF THE CHEMICALS USED IN A POOL PUMP ROOM ARE EITHER CORROSIVES OR OXIDIZERS.

ALL CONTAINERS IN THE PUMP ROOM MUST HAVE THE ORIGINAL CONTAINER LABEL ON THEM. NEVER PUT A CHEMICAL INTO AN UNLABELED CONTAINER!



Container labels should contain the following information:

- Identity of the chemical
- Reactivity of the chemical
- First Aid procedures
- Personal safety warnings, (i.e. do not breathe dust, use in ventilated area, use goggles, gloves, splash apron, and respirator)
- Name and address of the manufacturer
- All containers must have labels prominently affixed and shall provide chemical name and hazard warnings





# CHEMICAL STORAGE

1. ALWAYS KEEP THE DOOR LOCKED BUT HAVE THE KEY EASILY ACCESSIBLE IN CASE OF AN EMERGENCY.
2. KEEP BAGS OF CHEMICALS OFF THE FLOOR. KEEPING THE BAGS OF BULK CHEMICALS IN SMALL TRASH CANS WITH A LOCKABLE LID IS A GREAT WAY TO KEEP THE PUMP ROOM CLEANED, ORGANIZED AND WILL KEEP THE CHEMICALS DRY. ALSO, LAMINATE OR PUT IN A PLASTIC SHEET COVER, THE MSDS FOR THAT PRODUCT AND TAPE IT TO THE CONTAINER. YOUR CHEMICAL ROOM WILL BE LESS HAZARDOUS IF THE CHEMICALS ARE SECURELY STORED, AND THE MSDS WILL BE READILY ACCESSIBLE IF NEEDED.
3. MAKE SURE CONTAINERS ARE COVERED WITH THE ORIGINAL TOP.
  - A. ALL CHEMICALS SHOULD BE KEPT IN THEIR ORIGINAL CONTAINER (ESPECIALLY OXIDIZERS) OR A MARKED CONTAINER THAT INCLUDES THE MSDS SHEET ATTACHED TO IT.
  - B. SODIUM AND ACID VATS ARE TO BE KEPT COVERED AND AS FAR AWAY FROM EACH OTHER AS POSSIBLE. TRY NOT TO HAVE THESE VATS SITTING DIRECTLY UNDER THE CHEMICAL FEEDERS, AS IT WILL CAUSE THEM TO CORRODE EASILY.
  - C. ALL CONTAINERS ARE PROPERLY LABELED AND CONTAIN THE ORIGINAL CHEMICAL.
  - D. DUST COVERS ON ALL CORROSIVE MATERIALS; THEY WILL EASILY CORRODE METALS.
4. KEEP DIFFERENT CHEMICALS AWAY FROM EACH OTHER (EX: CHLORINES AND ACIDS).
  - A. STORE ALIKE CHEMICALS TOGETHER.
  - B. HAVE A SEPARATE STORAGE AREA FOR CLEANERS, ETC.
5. ALWAYS KEEP PUMP/CHEMICAL ROOM CLEAN, ORGANIZED, AND LOCKED.
6. ALWAYS CLEAN UP SPILLS IMMEDIATELY.



# ADDING CHEMICALS TO THE POOL

- AFTER PUTTING ON THE PROPER SAFETY EQUIPMENT, PUT WATER INTO A PLASTIC BUCKET (NOT METAL). ALWAYS ADD CHEMICAL TO WATER (STIR UNTIL DISSOLVED) AND UNLESS OTHERWISE DIRECTED, POUR AROUND THE POOL'S PERIMETER. NOTE: DO NOT USE METAL TO STIR THE CHEMICALS. USE A WOODEN OR PLASTIC SPOON OR HANDLE.
- NEVER ADD ANY CHEMICAL WHILE THE POOL IS IN USE.
- NEVER POUR CHEMICALS INTO THE SKIMMER, UNLESS SPECIFICALLY DIRECTED BY THE MANUFACTURER. THIS SENDS A STRONG CONCENTRATION DIRECTLY TO THE PUMP AND CAN DAMAGE THE MOTOR.
- READ MSDS ON ALL CHEMICALS BEFORE ADDING AND ALWAYS READ THE MANUFACTURERS RECOMMENDATIONS FOR CORRECT METHOD OF ADDING.



# PERSONAL PROTECTIVE EQUIPMENT

- MUST HAVE ONSITE THE PROPER SAFETY EQUIPMENT TO HANDLE THE CHEMICALS:
  - GOGGLES OR FACE SHIELD
  - CHEMICAL RESISTANT APRON
  - CHEMICAL RESISTANT GLOVES
  - DUST MASK OR RESPIRATOR
- EYEWASH STATION MUST BE PROVIDED



# CHEMICAL SAFETY TIPS

## SAFETY FIRST

### STORAGE & HANDLING OF CHEMICALS

All chemical shall be kept out of the reach of children

Chemical shall be stored in their original containers with their lids securely fastened.

Chemicals shall be kept in cool, well-ventilated area and out of direct sunlight.

Chemicals shall not be stored near a heat source, open flame or electrical equipment.

Liquid Chemicals shall not be stored above or adjacent to dry chemicals

Bags of chemicals shall be kept off the floor by at least 6 inches..

Acids shall be stored separately from bases.

Chemical storage areas should be kept cleaned with all exit areas unimpeded.

Manufacturer instruction labels shall be followed.

Safety Data Sheets shall be available for every chemical in the workplace.

Separate measuring devices shall be used for each chemicals. These measuring devices shall be clean, dry and shall be kept with the chemical. Glass devices shall not be used.

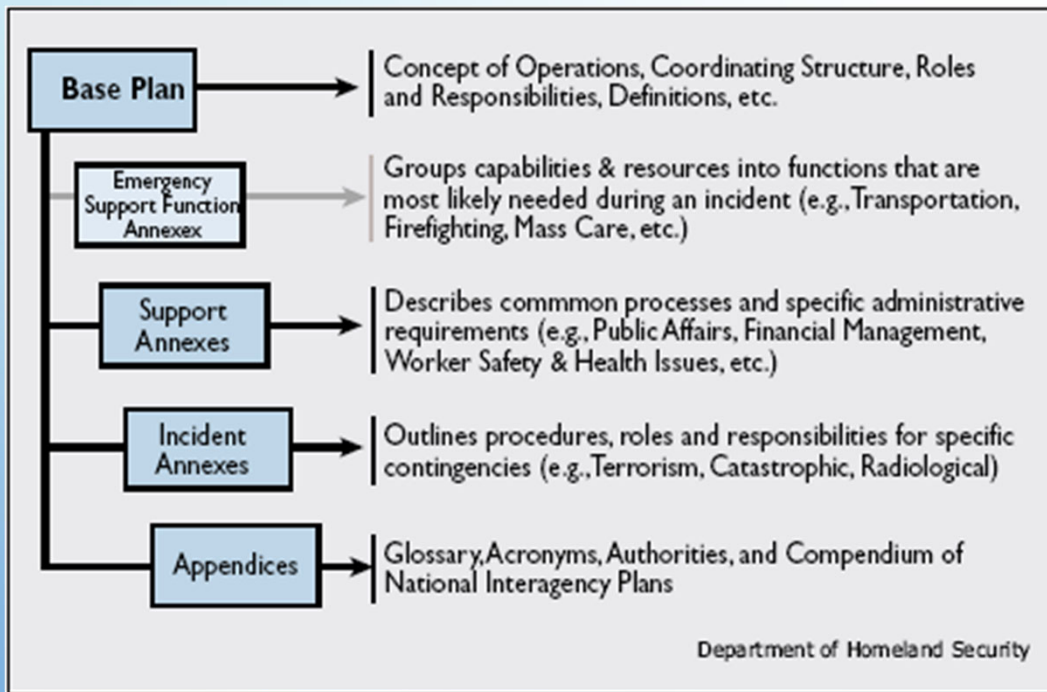
Chemical shall be added to water, water shall not be added to chemicals.

1. Read the label
2. Keep chemicals out of patrons' reach
3. Store chemicals off the floor in a clean well-ventilated area
4. Always add chemical to water; not water to chemicals
5. Wear proper safety equipment
6. Close all containers when finished
7. Keep reactive chemicals apart
8. Different measuring devices
9. Keep pool chemicals away from other chemical, paints and fertilizers
10. Do not reuse empty containers for other chemicals
11. Dispose of empties per MSDS
12. Never mix chemicals together unless directions tell you to
13. Pour with care
14. Broadcast dry chemicals with no bathers in area and upwind
15. Clean up spills quickly
16. Train staff
17. Be careful!





# EMERGENCY ACTION PLANS



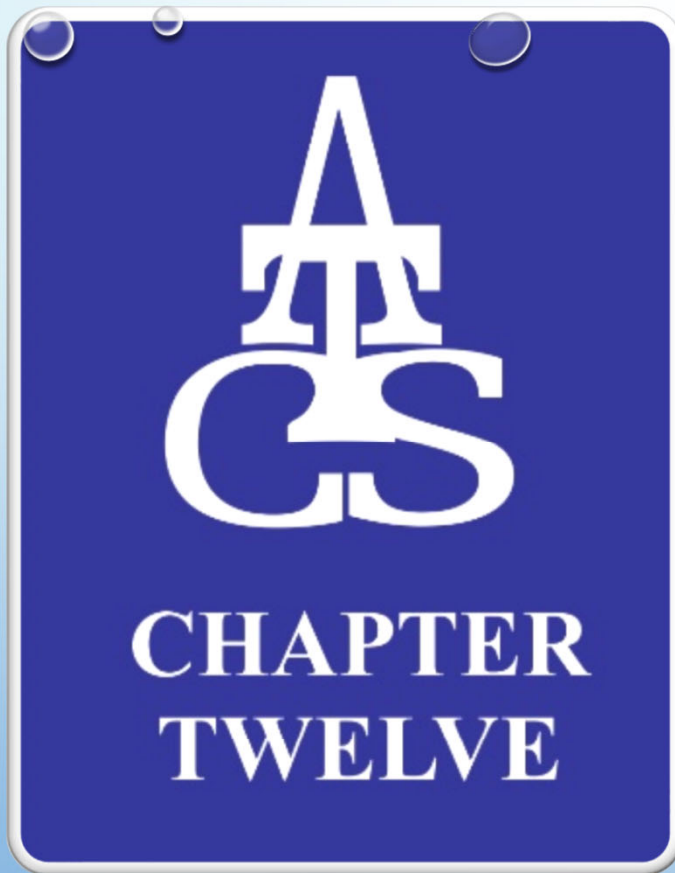
- GUIDELINES AND PROCEDURES FOR HANDLING EMERGENCIES
  - ACCIDENTS
  - EQUIPMENT FAILURES
  - UTILITY FAILURES
  - STORM WARNINGS
  - OTHER EMERGENCIES
- DEVELOP; PRACTICE; REVIEW; ADJUST



# IN THE EVENT OF A CHEMICAL SPILL

1. Keep all patrons and staff safe!
2. Close off the area.
3. Evacuate, if fumes are present.
4. Call 911, if fumes are present OR if someone is injured.
5. Consult your *Materials Safety Data Sheets*.
6. If the spill is minor and dry, sweep it up carefully and replace in original container or dispose of it.
7. If the spill is minor and liquid, dilute the area **WELL** with water.





**DAILY AND  
SEASON  
OPERATING  
PROCEDURES**



# PREVENTIVE & ROUTINE MAINTENANCE

VISUAL AND MANUAL INSPECTIONS OF THE POOL, EQUIPMENT, AND ENVIRONMENT NEED TO BE COMPLETED EVERY DAY. A CHECKLIST TO BE INITIALED BY THE STAFF WHEN THE INSPECTIONS ARE COMPLETED IS THE BEST WAY TO KEEP TRACK OF NEEDED MAINTENANCE AND SUPPLIES. NOTE ANY UNUSUAL FINDINGS SUCH AS ACTS OF VANDALISM OR BROKEN EQUIPMENT, ETC.



## TYPE OF AQUATIC VENUES

- FLAT WATER AQUATIC VENUES ARE WHERE THE WATER LINE IS STATIC EXCEPT FOR MOVEMENT MADE BY USERS USUALLY AS A HORIZONTAL USE AS IN SWIMMING.
- AGITATED WATER AQUATIC VENUES WHERE AQUATIC FEATURES DISCHARGE, SPRAY, OR MOVE THE WATER'S SURFACE ABOVE AND/OR BELOW THE STATIC WATER LINE THROUGH MECHANICAL MEANS .IN THESE VENUES PEOPLE ARE STANDING OR PLAYING VERTICALLY.
- HOT WATER AQUATIC VENUES ARE WHERE THE WATER TEMPERATURE IS MAINTAINED OVER 90 DEGREES F.



# CLASSIFICATIONS OF SWIMMING POOLS

- CLASS A: PRIVATE SWIMMING POOLS THAT SERVICE A RESIDENTIAL OR GROUP HOME.
- CLASS B: PUBLIC POOL THAT IS INTENDED FOR RECREATION USE.
- CLASS C: SEMI-PUBLIC POOL THAT IS INTENDED FOR USE BY A CLOSED GROUP SUCH AS A CONDOMINIUM ASSOCIATION OR HOTEL.
- CLASS D: SPECIAL USE SWIMMING POOL THAT USUALLY HAS AQUATIC FEATURE SUCH AS A WATERPARK OR WAVE POOL.
- CLASS E: THERAPY POOL THAT IS NOT FOR RECREATION AND IS USUALLY HOT WATER.
- CLASS F: WADING POOL IS VERY SHALLOW WATER AND IS REFERRED TO AS A "KIDDIE" OR "BABY" POOL.

# BARRIERS

All aquatic facilities must enclose their chemical and mechanical spaces to prevent access from the public. These enclosures could include walls or fencing that is at least six feet high. These enclosures cannot impede patron accessibility to the swimming pool, locker rooms or exit areas. All doors and gates that provide barriers in the facility must have self-latching gates. All emergency exit areas need to be well marked and unencumbered.



# SIGNAGE

Signage is extremely important at facilities. Signage that is required at all pools include:

- Emergency phone instructions
- Unauthorized area signs
- Chemical danger signs
- Variable water depth signs
- First aid location
- Management contact information
- Hours of operations
- Pool rules

Principal Investigator: Rob Fox 301-760-7114  
Safety Coordinator: Elizabeth Fox 301-760-7114  
Location: 448 W. South Street, Frederick, MD 21701

In case of emergency:  
Rob Fox 240-529-5774

**Hazard Diamond:** 3 (Red), 3 (Blue), 2 (Yellow), W (White), G (Green)

**PPE:**  
Attire: Lab coat, Closed-toe shoes  
Eye Protection: Safety glasses when working with hazards to the eyes.  
Gloves: Appropriate to the material being handled.

**High Hazard Chemicals:**  
Type: Acutely Toxic Chemical  
Requirements: This room contains chemicals with high acute toxicity. Only appropriately trained individuals may work with these materials.

**Other Hazards:**  
**DANGER**  
Corrosive (causing severe skin burns and eye damage)  
Hazardous to the environment (aquatic toxicity)  
Hazardous to the environment (aquatic toxicity)

**NO SMOKING, EATING OR DRINKING**

**Biohazard:**  
Biosafety Level 2  
Agents used: human material, venereal  
Special Conditions:

# DEPTH MARKERS



All swimming pools must have water depths clearly marked. Permanent depth markers must indicate the minimum and maximum depth on both sides of the pool and at the break in the floor slope where the pool goes from shallow to deep. Depth marking should be on the tile line, coping stones, and/or printed on the deck and shall be no more than 25 feet apart. For shallow water (less than five feet), the depth should be marked as the pool gets each foot deeper. In addition, all areas that are less than 5 feet deep must be marked with “No Diving” markers.





# EMERGENCY RESPONSE PLANS

**EMERGENCY SCRIPT**

**FOR EMERGENCY USE ONLY**


DIAL 911

*"An aquatic emergency exists at the Frederick Community Swimming pool. The address of the facility is 448 W South Street, Frederick, MD. There is an Emergency Gate at the rear of the swimming facility. The ambulance can access the gate at the end of the parking lot. Someone will meet you there and direct you to the pool.*

*There has apparently been a \_\_\_\_\_ injury."*

*Remain on the phone with the dispatcher until EMS arrives.*

**Do Not Hang Up**



**EMERGENCY ACTION PLAN**

**Assess Situation**

**Life Threatening/Non-Life Threatening**

Clear Swimming Pool or Emergency Area

Primary Rescuer Responds to Emergency Call 911(if necessary)

Alert Additional Staff

Provide Appropriate Emergency Response

Administer First Aid Where Appropriate

Assign Individual To Meet EMS At Emergency Door

Complete Incident/Accident Report

Meet With Staff To Debrief Situation

Contact Insurance Company

Each facility should have their Emergency Action Plan and Emergency Call Script posted.





# PREVENTIVE & ROUTINE MAINTENANCE PLAN

- PSPO RESPONSIBLE FOR
  - IMPLEMENTATION OF PROCEDURES
    - MAINTAIN AND PREVENT LONG-RANGE DETERIORATION OF FACILITY
    - INCLUDES DAILY, WEEKLY, MONTHLY, SEASONAL AND ANNUAL ACTIVITIES
  - PROCEDURES MUST BE IN WRITING
  - MASTER LIST OR CALENDAR
- WHEN ITEM COMPLETED – DATE AND INITIAL
- INVENTORY OF EQUIPMENT
  - LOCATION
  - DATE PURCHASED
  - REPAIR HISTORY
  - PARTS LISTS
  - OWNER'S MANUALS
  - ADD MFG. MAINTENANCE ITEMS TO MASTER LIST



# DAILY OPENING PROCEDURES

- The professional swimming pool operator shall ensure that a daily facility preventive maintenance inspection is done before opening and that includes:
- Drain covers, vacuum fitting covers, SKIMMER equalizer covers, and any other suction outlet covers are in place, secure, and unbroken.
- Skimmer baskets, weirs, lids, flow adjusters, and suction outlets are free of any blockage.
- Inlet and return covers and any other fittings are in place, secure, and unbroken.
- Safety warning signs and other signage are in place and in good repair.
- Entrapment prevention systems are operational.
- Recirculation, disinfection systems, controller(s), and probes are operating as required.
- Underwater lights and other lighting are intact with no exposed wires or water in lights.
- Slime and biofilm have been removed from accessible surfaces of the pool.
- Doors to nonpublic areas (chemical storage spaces, offices, etc.) Are locked.
- Fecal/vomit/blood incident contamination response protocols, materials, and equipment are available.
- Electrical devices are in good working condition.
- Water clarity is such that the bottom and objects on the bottom of the pool are clearly visible.



# DAILY CLOSING PROCEDURES

- DAILY CLOSING PROCEDURES MIGHT INCLUDE
  - EQUIPMENT TAKEN OUT – STORED
  - LOST AND FOUND DRIED, FOLDED, PUT AWAY
  - BATHHOUSES CLEANED AND SANITIZED
  - TRASH CANS EMPTIED
  - FILTER EQUIPMENT CHECKED
  - DECKS CLEANED
  - LIGHTS TURNED ON/OFF
  - SECURITY TURNED ON
  - DOORS AND ENTRY POINTS LOCKED



# OFF-SEASON MAINTENANCE

- PREOPENING PROCEDURES
  - WINTER DAMAGE WILL NEED TO BE REPAIRED
  - POOL WILL NEED TO BE DRAINED & CLEAN
  - EACH POOL WILL NEED TO BE CHECKED BY THE HEALTH DEPARTMENT PRIOR TO OPENING
- WINTERIZATION
  - VARIES DEPENDING ON CLIMATE
  - PURPOSE:
    - PREVENT DAMAGE TO SHELL
    - PREVENT OR REDUCE BUILD-UP OF RUST AND DETERIORATION OF EQUIPMENT
    - MINIMIZE VANDALISM
    - IDENTIFY REPAIR NEEDS PRIOR TO REOPENING



# UNDERWATER LIGHTING

## DIFFERENT TYPES OF LIGHTING

- WET NICHE - UNDERWATER
- DRY NICHE – LENS PART OF POOL
- LED

**WET-NICHE** lights are lights that are located within the pool (compared to dry-niche lights, which actually are outside of the pool). **NEVER turn a wet-niche light on when it is not in water.** The water of the pool is used to cool the light. If the glass is not wet, the light can explode





**CHAPTER  
THIRTEEN**

# **LEGAL AND LIABILITY**





# LIABILITY – RISK MANAGEMENT

- PSPO
  - PREVENT OR AT LEAST REDUCE LIABILITY FOR STAFF AND EMPLOYER
  - PREVENT OR AT LEAST REDUCE LIABILITY FOR FACILITY EXPOSURE TO LAWSUITS
- COULD THE STAFF, ACTING REASONABLY, FORESEE AND PREVENT SITUATION FROM OCCURRING?

# COMMON REASONS FOR LIABILITY SUITS

1. FAILURE TO PROVIDE SAFE FACILITIES
2. FAILURE TO PROVIDE SAFE ENVIRONMENT
3. FAILURE TO PROVIDE SAFETY EQUIPMENT
4. THE AREA LACKS RING BUOYS, LIFEBOATS, RESTRAINING LINES.
5. FAILURE TO SUPERVISE
6. FAILURE TO POST
7. FAILURE TO COMPLY TO LAWS
8. FAILURE TO PROVIDE SECURITY
9. FAILURE OF AN EMPLOYEE TO PROPERLY PERFORM HIS DUTIES
10. IMPROPER ACTION
11. LACK OF ACTION





# VIGILANCE

## THE KEY TO VICTIM RECOGNITION

### DEFINITION OF VIGILANCE :

The ability to see and recognize a signal event and then take the appropriate action.

A signal event is either occurring or has the high potential of occurring that could lead to an aquatic accident.

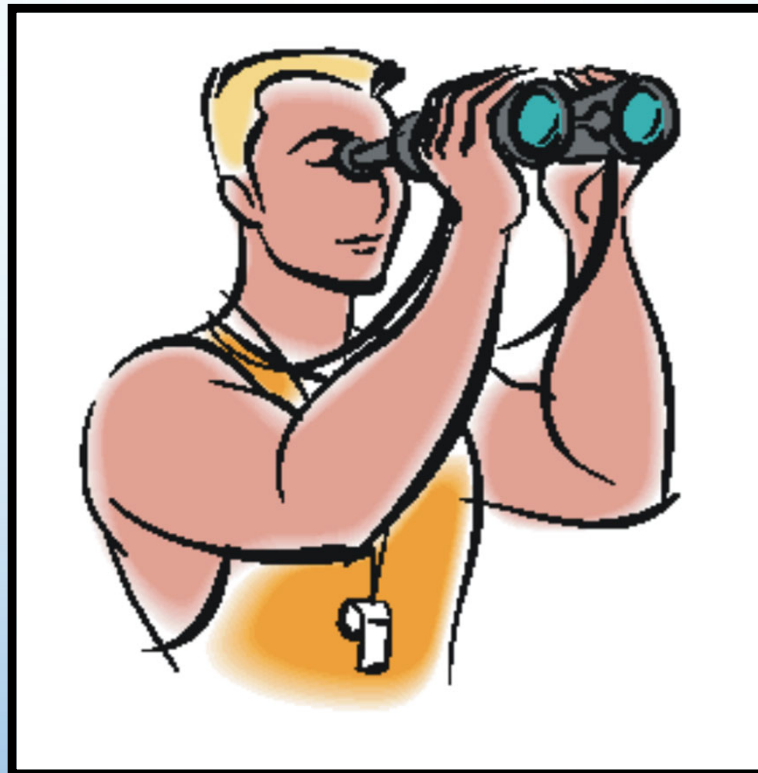
IF YOU DON'T  
EXPECT IT,  
YOU WON'T  
SEE IT!



*The Guards  
Must Believe  
It Can Happen  
To Them!*



# SCANNING



*Defend The Water...Not The People!*



**1** Head in constant motion

**5**

Scan  
Within 10 to  
15  
Seconds

**5 FACTORS  
TO SCANNING  
EFFECTIVELY**

**2** Keep A  
Consistent  
Pattern

**4**

Pick Up  
All Of The  
Corners

**3**

Pronounced  
Downward  
Head Swing



# WHAT'S THE DEAL WITH A “PRONOUNCED DOWNWARD HEAD SWING?”

The majority of drownings at guarded facilities happen  
within 6 feet of the guard chair.

If they don't look down, they'll never see them.



## WHAT TO AVOID WHEN SCANNING:

- LOOKING BUT NOT SEEING,
- SCANNING TOO FAST,
- FAVORING ONE SIDE,
- STUTTER SCANNING, &
- SCANNING  
“UNCONSCIOUSLY.”



# NO MOVEMENT FOR 10 - GET THEM!

WOULDN'T YOU RATHER SAVE 100  
PEOPLE THAT DIDN'T NEED IT, THAN  
MISS JUST ONE THAT DID?







THE NATIONAL AQUATIC SAFETY COMPANY

**No Movement for 10 -  
Get Them!**

5 ROLES OF A LIFEGUARD

- ① RECOGNITION
- ② PREVENTION
- ③ RESCUE
- ④ INITIATION
- ⑤ SUPPORT

**My Actions Should**  
**-Be Safe**  
**-Look Safe**  
**-Be Defendable**  
**-Be Liked by Patrons**



**THE WORST THING YOU CAN DO IS DO NOTHING!**



# LIABILITY – RISK MANAGEMENT


**Negligence** is a term that can sum up almost everything mentioned in this section. This one word is the single most common cause of lawsuits in the aquatic industry. Covering a broad area of topics, negligence is doing something wrong, not doing what should've been done, going beyond the level of training, etc. Not doing your job the exact way it should've been done. Negligence covers many things.

- Elements of negligence:
  - Legal duty of care
  - Breach of duty
  - Breach was cause of injury
  - Substantial damage
- For liability to exist all four elements must exist
- Close monitoring – difference between winning and losing a lawsuit



# WHEN AN ACCIDENT OCCURS

- IMMEDIATE AND TEMPORARY CARE
- FILL OUT ACCIDENT REPORT ASAP
- COPY REPORT TO ADMINISTRATOR WITHIN 1 HOUR
- CONTACT INSURANCE CARRIER, IF NECESSARY
- IF MAJOR ACCIDENT, DEBRIEF ALL INVOLVED



**Aquatic Training & Consulting Services**  
 448 W South Street  
 Frederick, MD 21701  
 301-760-7114  
[info@aquatictrainingservice.com](mailto:info@aquatictrainingservice.com)  
 Address \_\_\_\_\_

Incident/ Accident	
___ Accident ___ Incident	Staff on duty _____
Type of violation (if any) _____	_____
___ Pool and health club regulation	_____
___ Health department regulation	_____
___ General safety	Authorized patron YES NO
___ Other _____	Was this person's pool privilege suspended YES NO
Was 911 called YES NO	Length of suspension _____

Person(s) Involved/ Injured		
Name _____	Age _____	Name of parent if person is a minor _____
Address _____	Telephone _____	

Person(s) Involved/ Injured		
Name _____	Age _____	Name of parent if person is a minor _____
Address _____	Telephone _____	

Incident/ Accident Description	
Date and time _____	_____
Location (Health Club, Indoor Pool, Jacuzzi, Etc.) _____	_____
Description of Incident/ Accident _____	_____
_____	_____
Was this person taken to the hospital _____	Which one _____

Person(s) notified/ witnesses		
Witness _____	Address _____	Telephone _____
Witness _____	Address _____	Telephone _____
Witness _____	Address _____	Telephone _____

Office Use		
Insurance company notified _____	Date _____	By _____
Follow up/ deposition _____		





## SWIM AT YOUR OWN RISK

Lessens liability, does not negate it.  
Facility can still be considered negligence.  
Staff must enforce posted rules.



# LIGHTNING SAFETY



- MANY STATES REQUIRE SUSPENSION OF ACTIVITY WHEN THUNDERSTORMS ARE IN AREA
- PSPO NEEDS TO DETERMINE SAFE AND NON-SAFE AREAS WITHIN FACILITY
- FOLLOW 30/30 RULE TO MINIMIZE RISK, THERE IS NO DIFFERENCE BETWEEN THUNDER AND LIGHTNING
- IF POOL HAS BEEN CERTIFIED AS GROUNDED, THEN THE FACILITY MAY REMAIN OPEN



# AMERICANS WITH DISABILITIES ACT OF 1990

- NO INDIVIDUAL CAN BE DISCRIMINATED AGAINST BASED ON DISABILITY
- REASONABLE ACCOMMODATIONS MUST BE MADE

